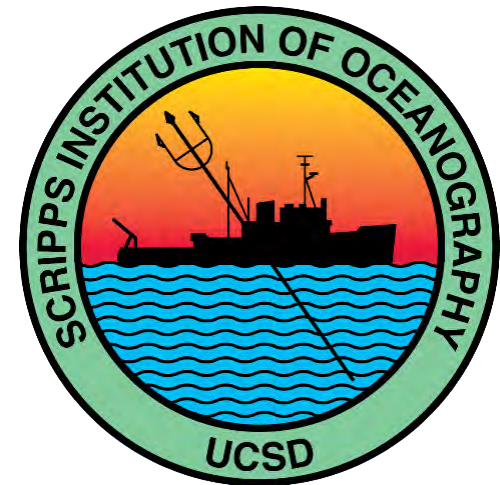


Optics of Marine Particles

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Sixth IOCCG Summer Lecture Series
4 - 16 November 2024, Hyderabad, India

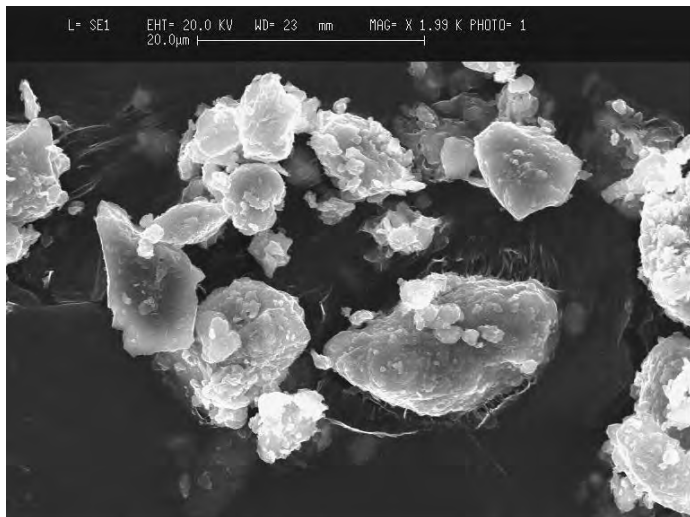
Seawater is a complex optical medium with a great variety of particle types and soluble species

- Molecular water
- Inorganic salts
- Dissolved organic matter
- Plankton microorganisms
- Organic detrital particles
- Mineral particles
- Colloidal particles
- Air bubbles

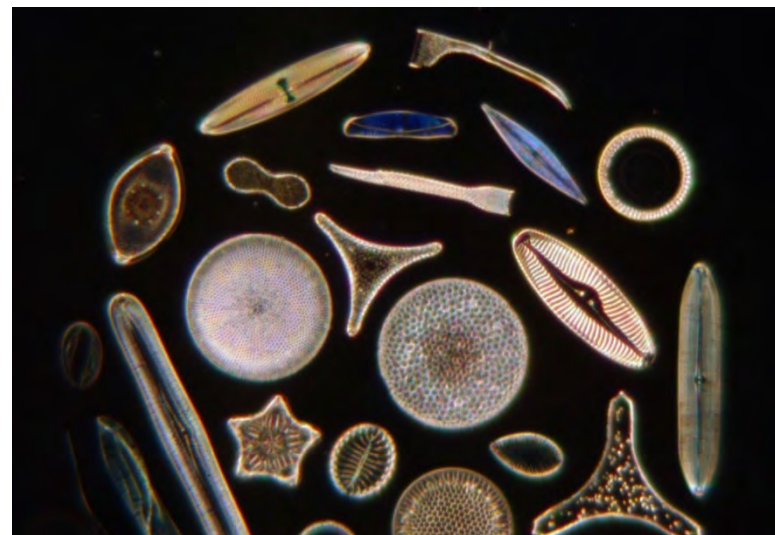
Suspended
Particulate
Matter

A great variety of biological and mineral particle types which absorb and scatter light differently

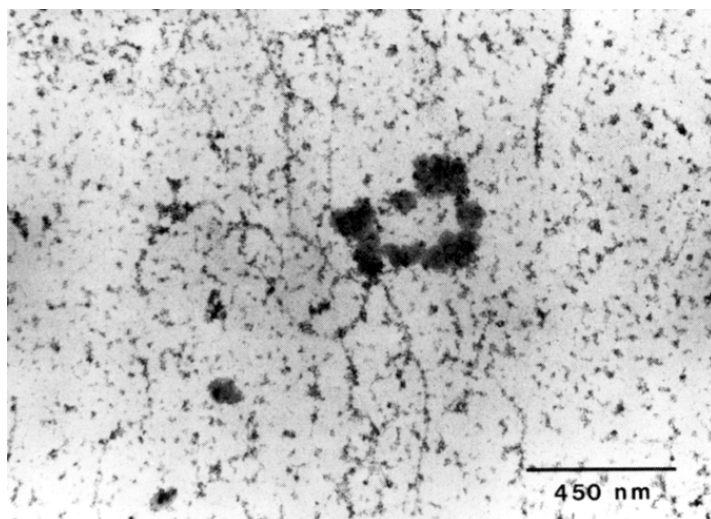
Mineral particles



Plankton microorganisms



Colloids / nanoparticles



**Fundamentals of single-particle optics
and the linkage between
the single-particle and bulk optical
properties of particle suspension**

Linkage between the single-particle optical properties and bulk optical properties of particle suspension

This is an example relationship for light absorption properties assuming that the bulk absorption coefficient represents a collection of identical particles (similar relationships can be written for light scattering and attenuation properties)

$$a = (N/V) Q_a G = (N/V) \sigma_a \quad \sigma_a = a / (N/V)$$

Bulk properties:

a is the bulk absorption coefficient of a collection of identical particles in aqueous suspension (units of m^{-1})

N/V is the number of particles per unit volume of water (units of m^{-3})

Single-particle properties:

Q_a is the absorption efficiency factor (dimensionless) – defined on the next slide

$\sigma_a (= Q_a G)$ is the absorption cross-section (units of m^2)

G is the area of geometric cross-section of particle (units of m^2)

For spherical particle $G = (\pi/4)D^2$ where D is a diameter

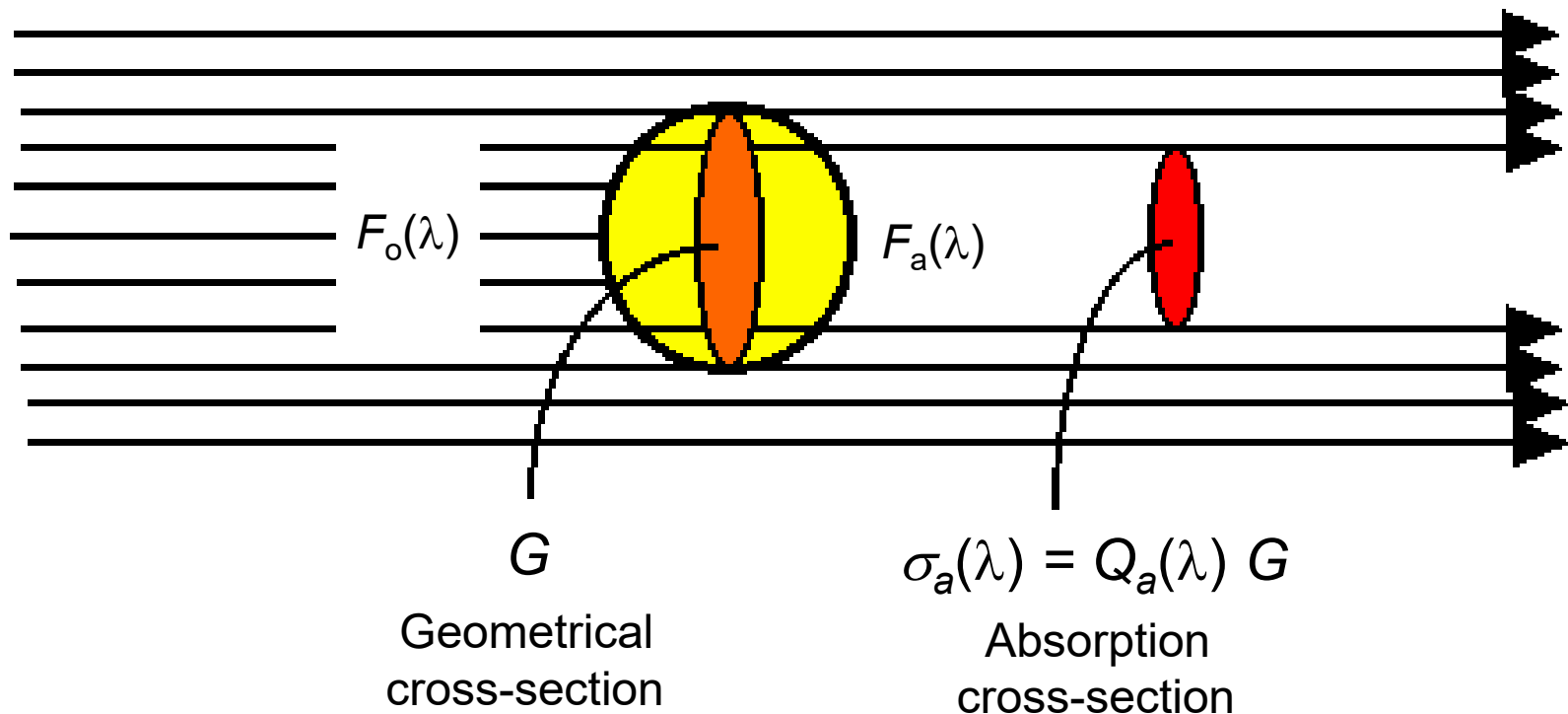
Note: a , Q_a , and σ_a are the spectral quantities (i.e., they are functions of light wavelength λ)

Absorption efficiency factor for a single particle

$$Q_a(\lambda) = F_a(\lambda) / F_o(\lambda)$$

$F_o(\lambda)$ - spectral radiant power intercepted by geometrical cross-section of particle

$F_a(\lambda)$ - spectral radiant power absorbed by particle



Theoretical dependence of absorption efficiency on particle properties parameterized in terms of “absorption thickness” ρ'

For a particle suspended in water

$$\rho' = 4 \alpha n' = a_s D$$

where the particle size parameter α is

$$\alpha = (\pi D n_w) / \lambda$$

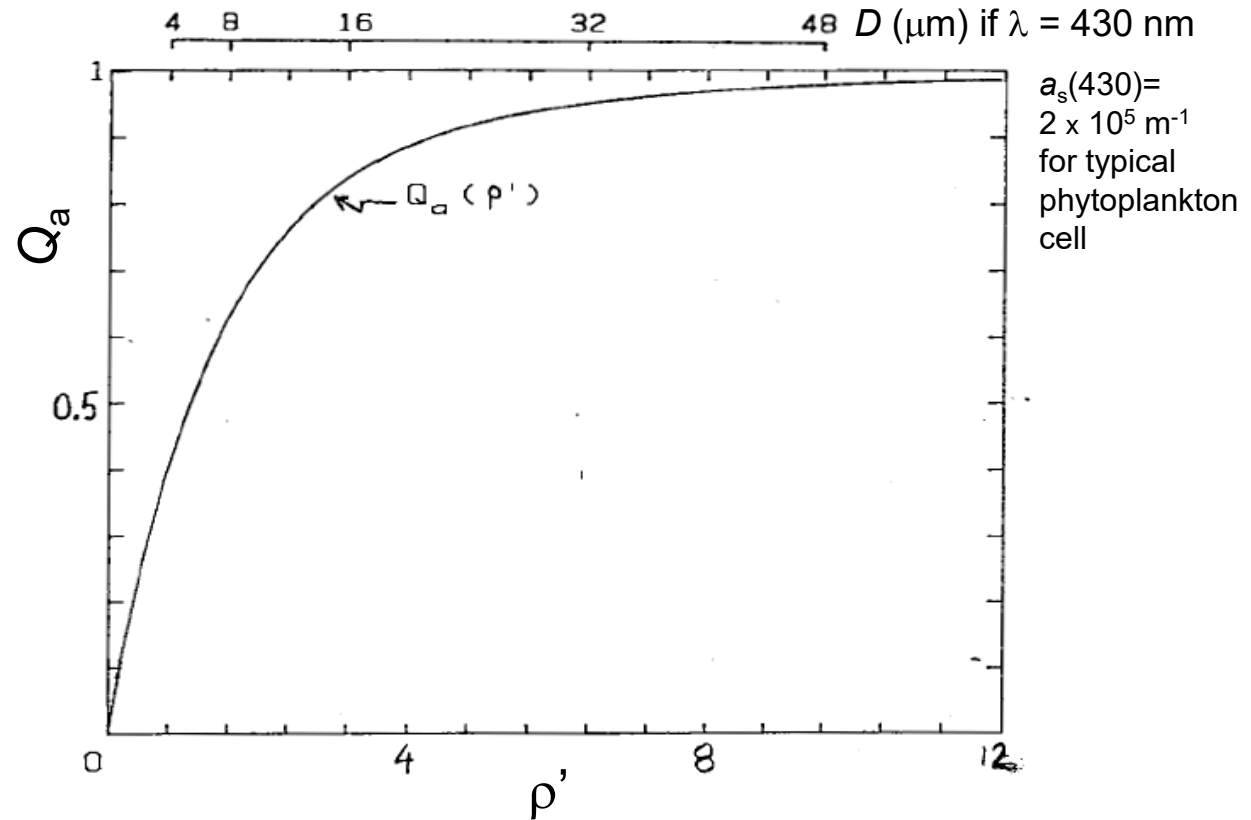
$$\lambda_w = \lambda / n_w$$

and the imaginary index of refraction of particle is

$$n' = (a_s \lambda) / (4 \pi n_w)$$

a_s (m^{-1}) is the absorption coefficient of substance forming the particle; D (m) is the particle diameter; and n_w is the refractive index of water

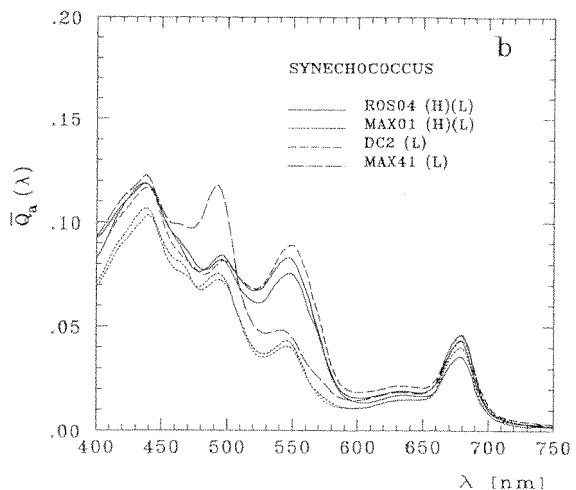
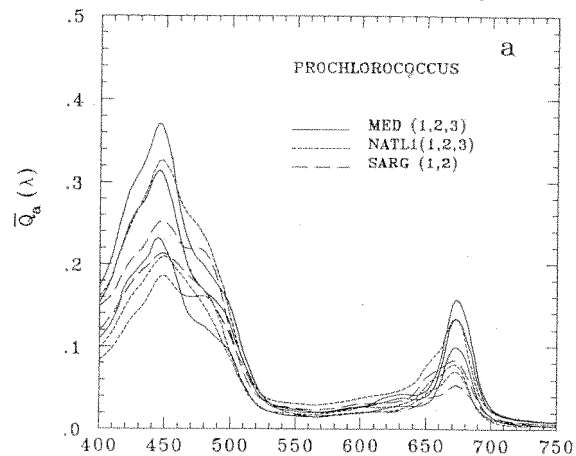
Note: ρ' , α and n' are dimensionless; symbol x is often used in literature instead of α
 Q_a , a_s , and n' are all functions of λ



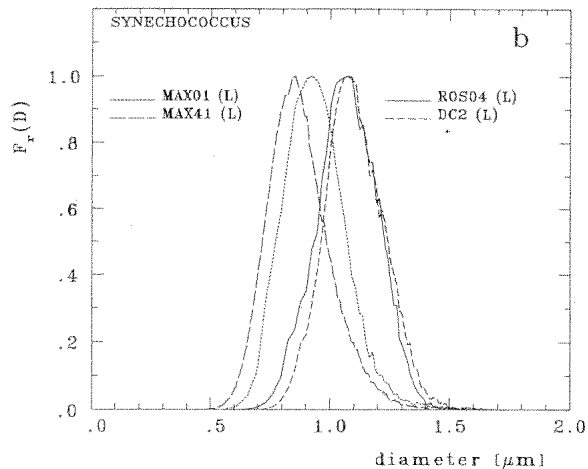
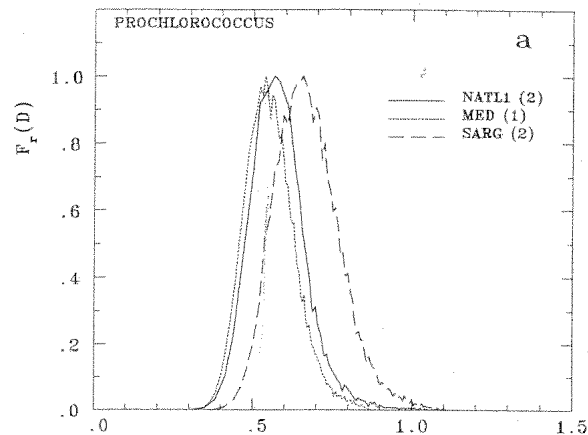
Example spectra of absorption efficiency factor for two phytoplankton species derived from laboratory measurements of $a(\lambda)$ and cell size distribution made on cultures

The *mean* efficiency factor, \overline{Q}_a , represents an “average” phytoplankton cell derived from the actual population of cells that exhibit a certain size distribution. Because the size distribution is narrow the mean is meaningful in a sense that it represents an “average” cell within a population of similar cells.

Mean absorption efficiency factor \overline{Q}_a



Size distribution $N(D)/V$



Comparison of experimental data of absorption efficiency for various phytoplankton and heterotrophic microorganisms with theoretical curve

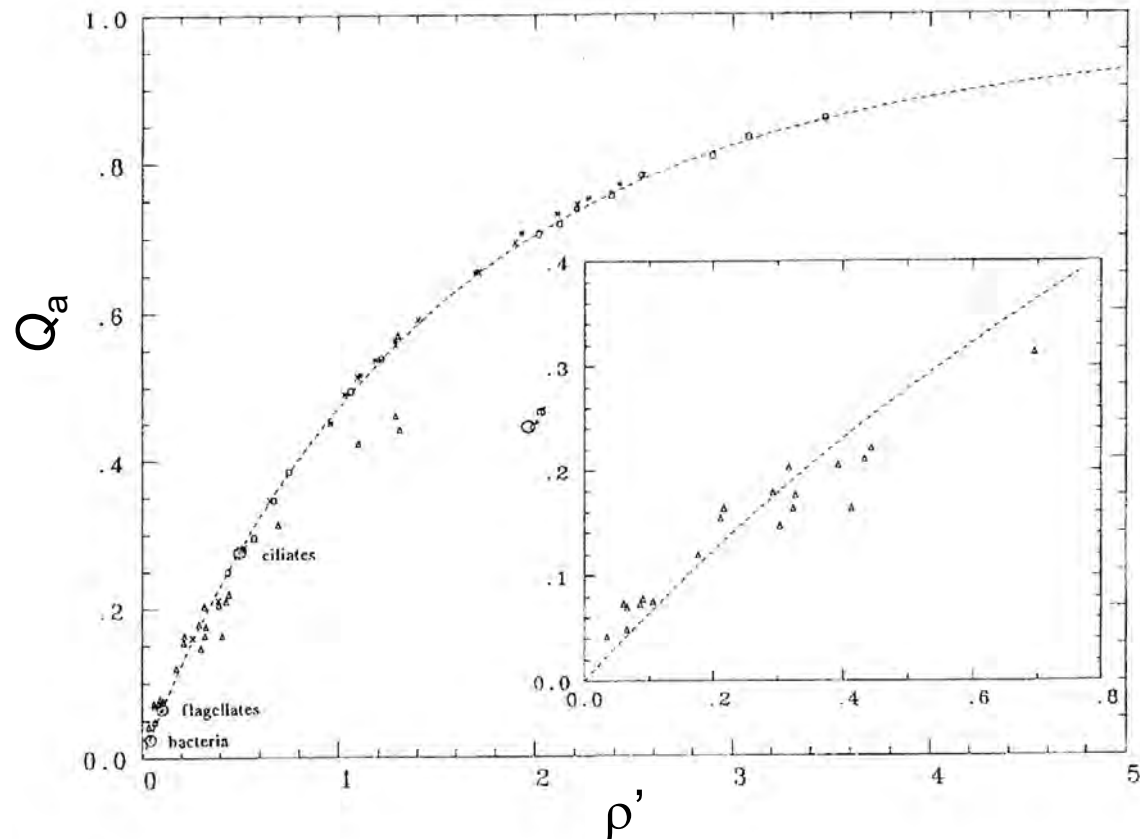


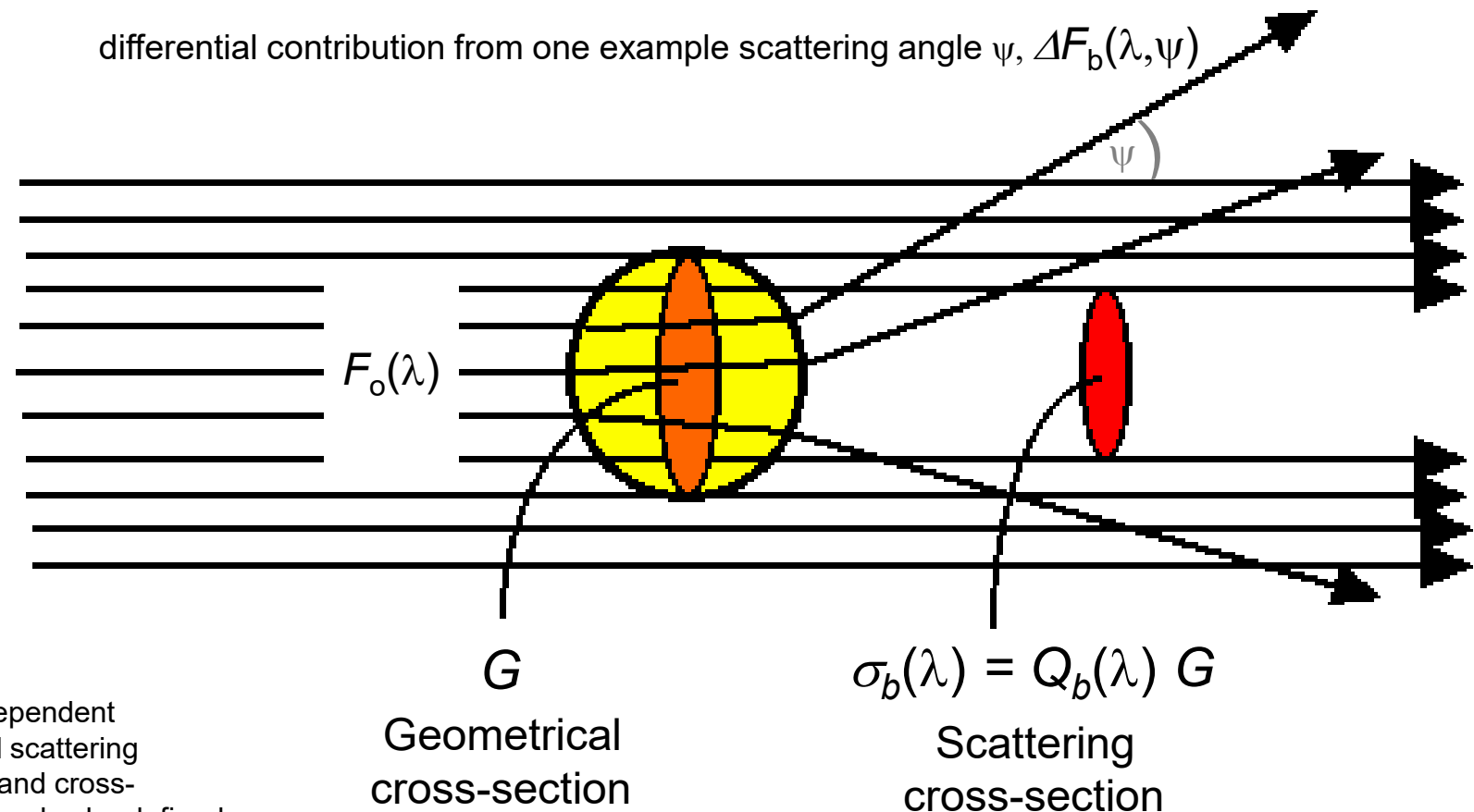
Figure 1. The theoretical variations of Q_a , the efficiency factor for absorption (dashed curves), as a function of the dimensionless parameter ρ' . The triangles are experimental determinations of Q_a (at 675 nm) for various algae (Morel and Bricaud, 1986; Ahn, 1990); other symbols are for determinations of 3 algal species studied by Sosik (1988). The values for heterotrophic organisms, as indicated, come from Morel and Ahn (1990, 1991). The inset is an enlargement of the initial part of the curve.

Scattering efficiency factor for a single particle

$$Q_b(\lambda) = F_b(\lambda) / F_o(\lambda)$$

$F_o(\lambda)$ - spectral radiant power intercepted by geometrical cross-section of particle

$F_b(\lambda)$ - spectral radiant power scattered by particle in all directions



Note: ψ -dependent differential scattering efficiency and cross-section can also be defined

Theoretical dependence of optical efficiency factors on particle properties parameterized in terms of phase shift parameter ρ

$$\rho = 2 \alpha (n - 1) \quad \text{where } n \text{ is the refractive index of particle relative to water}$$

Absorption efficiency

$$Q_a = F_a / F_o$$

Scattering efficiency

$$Q_b = F_b / F_o$$

Attenuation efficiency

$$Q_c = F_c / F_o$$

$$F_c = F_a + F_b$$

$$Q_c = Q_a + Q_b$$

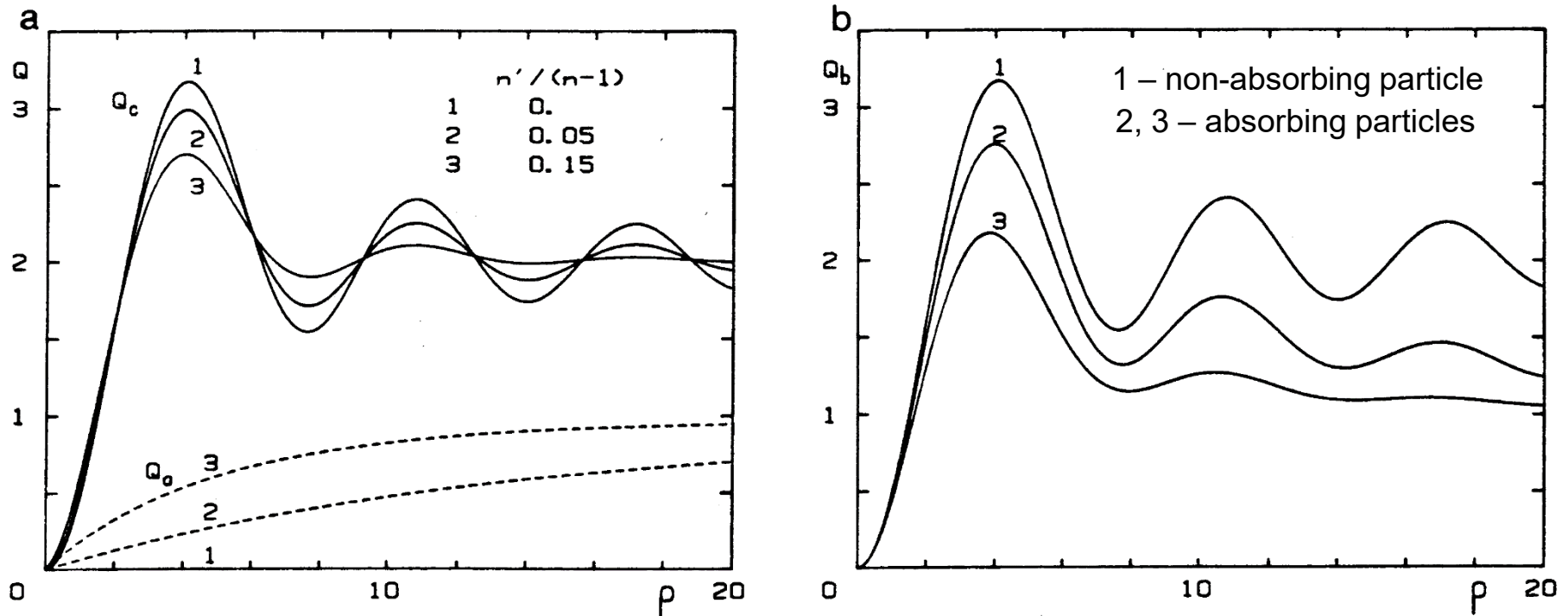


FIG. 3. Variations of the efficiency factors for attenuation, Q_c , for absorption, Q_a (a), and for scattering, Q_b (b) vs. the parameter $\rho = 2 \alpha (n - 1)$, for increasing values of the ratio $n'/(n - 1)$ where n and n' are the real and imaginary parts of the relative refractive index of the particles.

The effect of polydispersion on attenuation efficiency

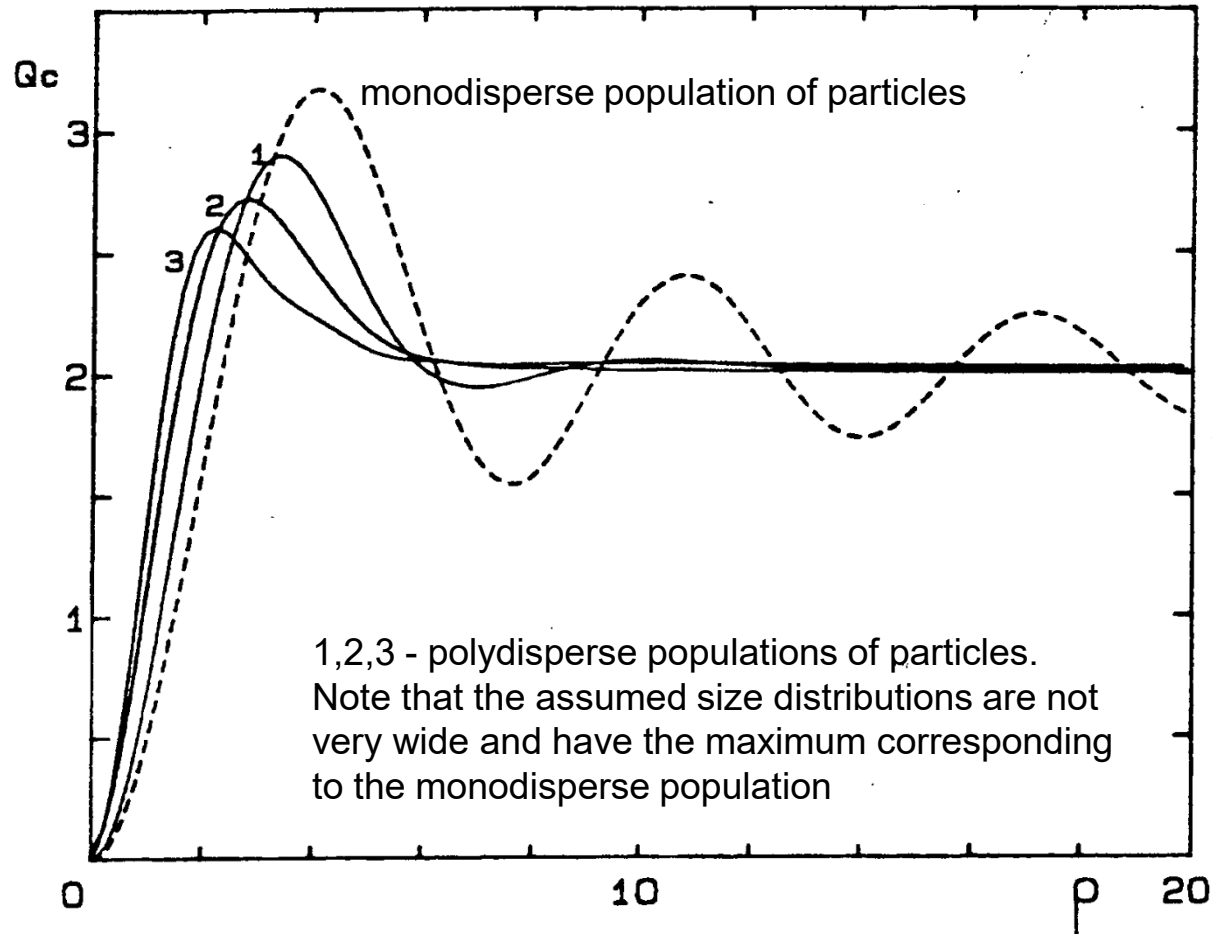


FIG. 4. Mean efficiency factor for attenuation Q_c of a "mean" particle representative of a polydispersed population, plotted as a function of q_m , the q value which corresponds to the maximum of the size distribution function $F(q)$ (see Equation 17). The index of refraction is real (no absorption) and the curves 1 and 3 correspond to log-normal distributions such as $F(q_M/2) = F(2q_M) =$ respectively 0.01, 0.1, 0.3 $F(q_M)$. The dashed curve, redrawn from Fig. 3 for $n' = 0$, represents the limiting case of a population of monosized particles.

Comparison of experimental data of scattering efficiency for various phytoplankton and heterotrophic microorganisms with theoretical curves

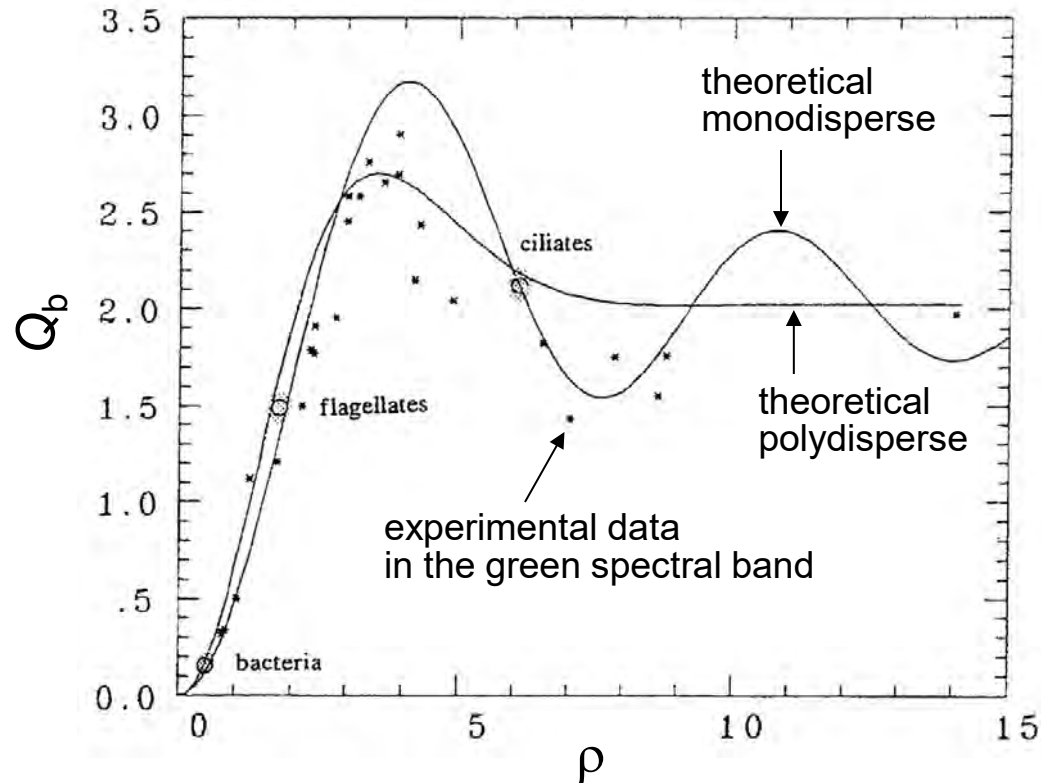
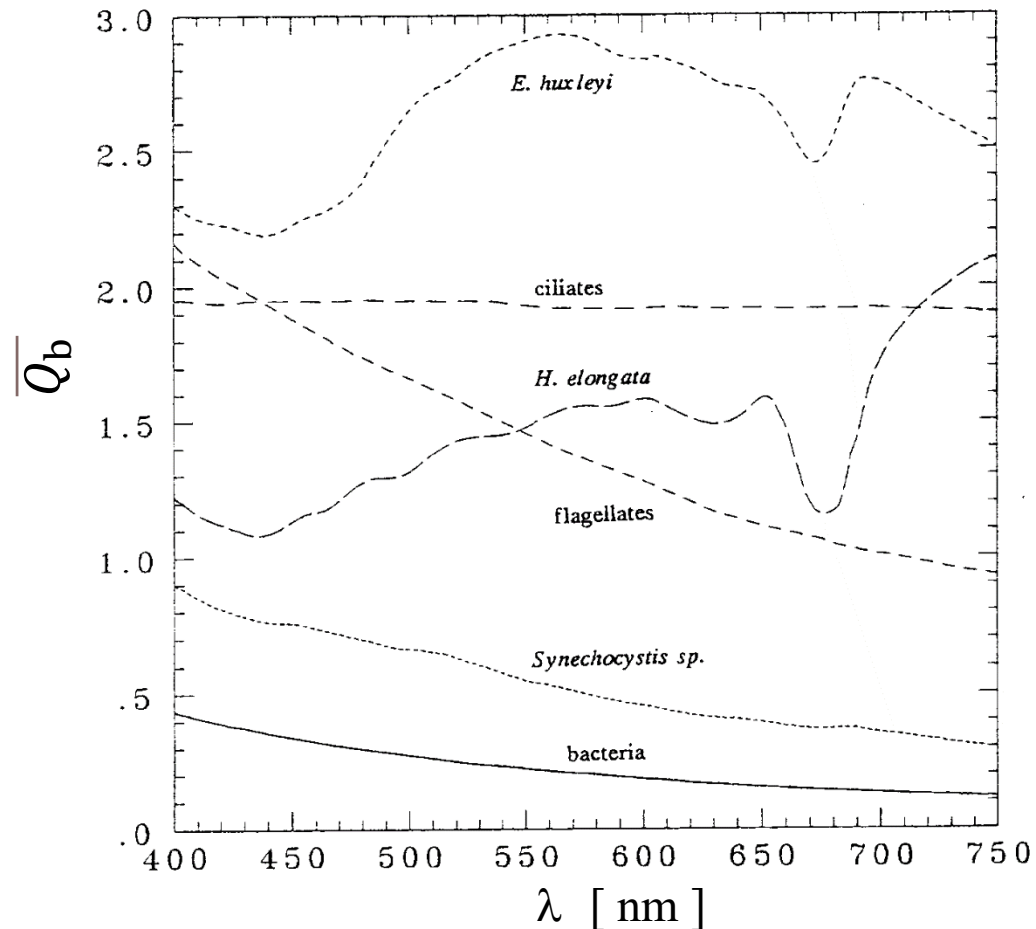


Figure 2. The theoretical variations of Q_b , the efficiency factor for scattering by non absorbing spheres (solid curve with marked oscillations) as a function of the dimensionless parameter ρ . The smoothed curve is for an averaged \bar{Q}_b to be applied for population with a log - normal size distribution. The crosses are the \bar{Q}_b values (at $\lambda \sim 580$ nm) determined for various phytoplankters grown in culture (see Table 1 in Morel and Bricaud, 1986); additional data for algal cells come from Ahn (1990). The circles indicate the \bar{Q}_b values (at $\lambda \sim 550$ nm) determined for free living marine bacteria, heterotrophic flagellates, and naked ciliates, (Morel and Ahn, 1990; 1991).

Spectra of scattering efficiency for various phototrophic and heterotrophic microorganisms derived from measurements



Note: Ciliates are the largest particles (~30 μm) in this comparison

Note: Bacteria are the smallest particles (~0.5 μm)

Figure 3. Spectral variations of \overline{Q}_b within the 400-750 nm range of various phototrophic and heterotrophic organisms as experimentally determined (Morel and Ahn, 1990, 1991).

$D = 3.4 \mu\text{m}$
 $n = 1.07$
 for visible light:
 $\alpha = 20 - 35$
 $\rho = 3 - 5$

Optical efficiency factors Q_c , Q_b , and Q_a :

Examples for monospecific
 cultures of phytoplankton cells
 (derived from laboratory
 measurements of absorption and
 attenuation coefficients, and size
 distribution made on cultures)

$D = 1.2 \mu\text{m}$
 $n = 1.0325$
 for visible light:
 $\alpha = 7 - 13$
 $\rho = 0.5 - 1.5$

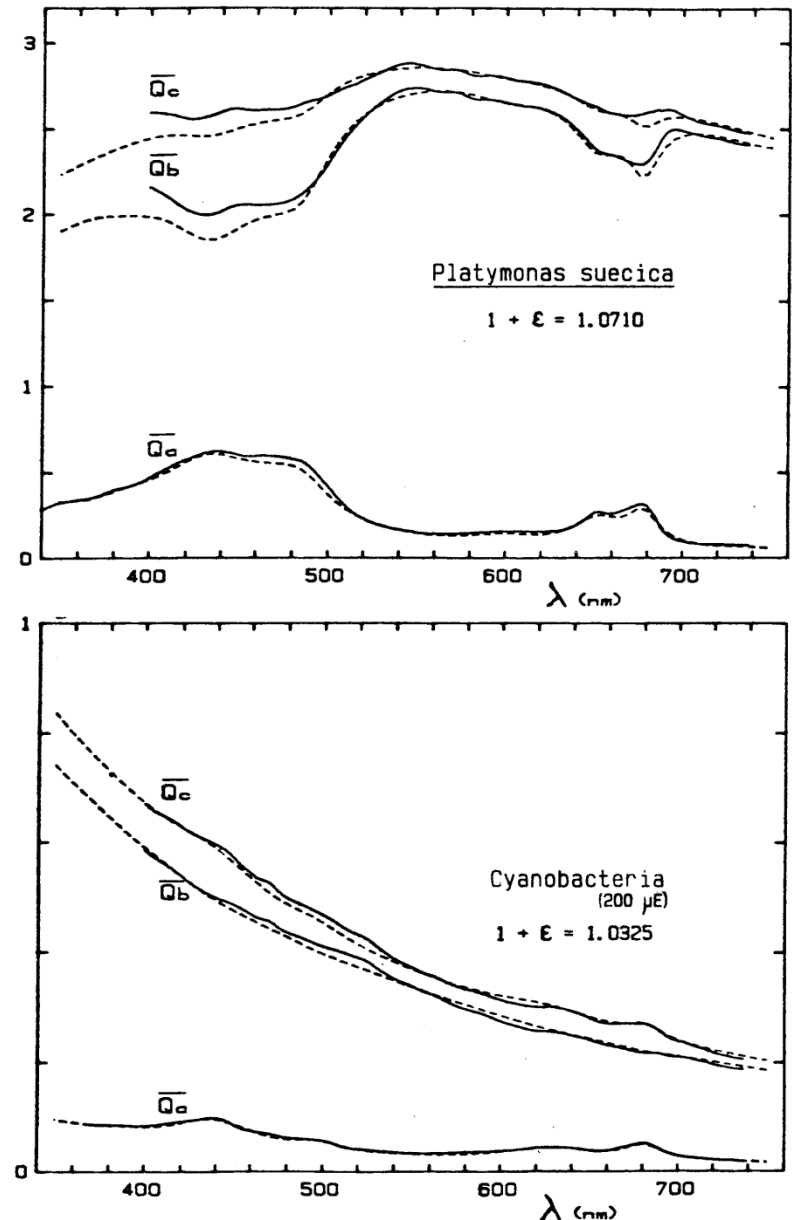


FIG. 14. Spectral variations of the mean efficiency factors for attenuation (\bar{Q}_c), scattering (\bar{Q}_b) and absorption (\bar{Q}_a), deduced from the attenuation and absorption coefficients experimentally determined (continuous lines), for two phytoplanktonic species. The variations of \bar{Q}_c , \bar{Q}_b and \bar{Q}_a obtained from a theoretical model (see text) are shown as dashed lines. The central value of the real part of the refractive index, $1 + \epsilon$, leading to the best theory/experiment agreement is indicated on the Figures.

Scattering phase function: Effects of particle size and refractive index

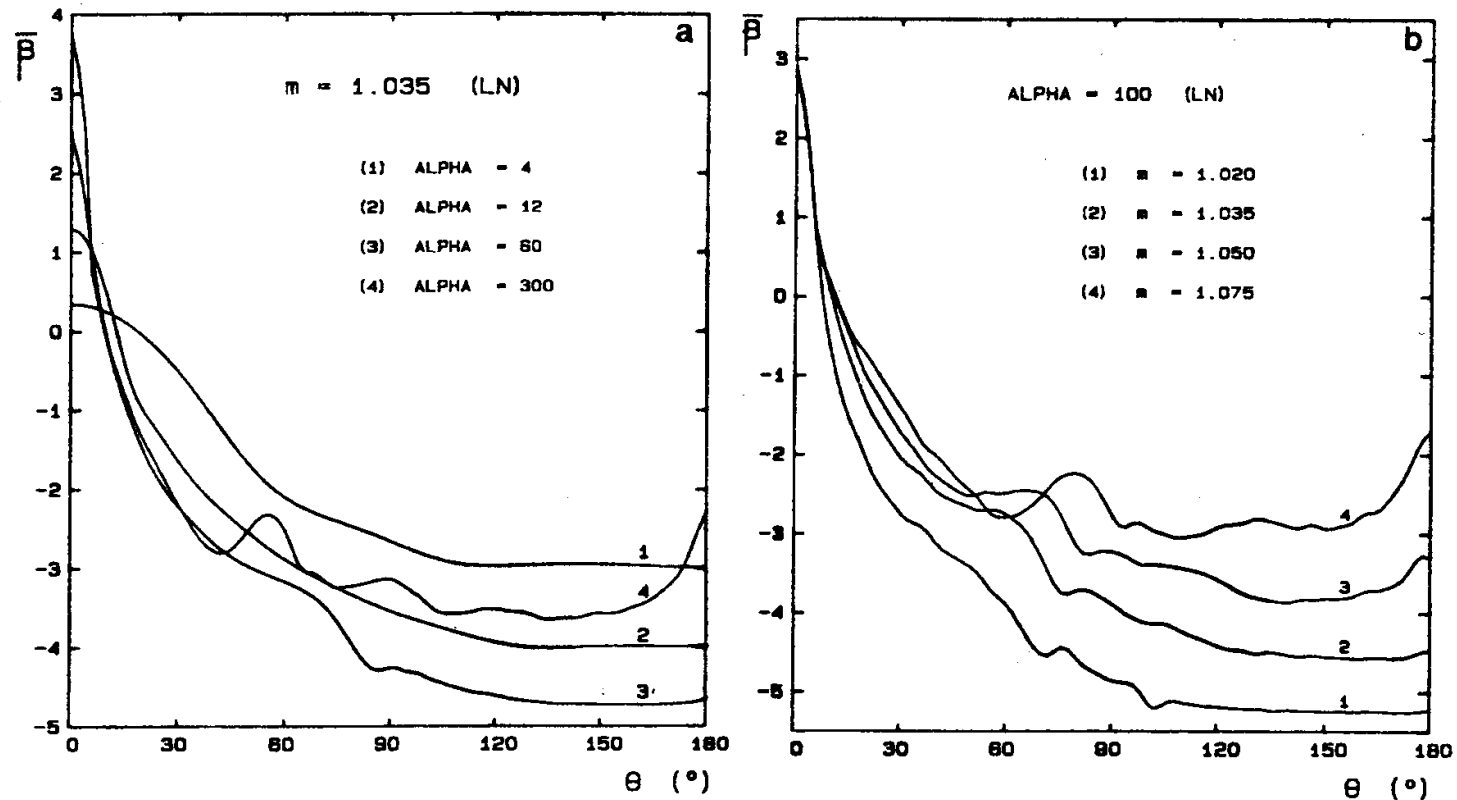


FIG. 6. (a) Normalized volume scattering function $\bar{\beta}(\theta)$ for increasing α_M values (increasing size) and for $m = 1.035$. (b) Normalized volume scattering function $\bar{\beta}(\theta)$ for increasing (real) index of refraction and for $\alpha_M = 100$. For Fig. 6a and b the log normal size distribution used is as in Fig. 5. The "bump" which occurs at about 75° for $m = 1.075$ and at smaller angles when the refractive index decreases (see also Fig. 6a) is the first "rainbow", at 138° for water droplets ($n = 1.33$). It appears for sufficiently large and perfect spheres. Thus it is unlikely that it can be observed for algal cells.

Normalized scattering function for various microorganisms (from Mie calculations)

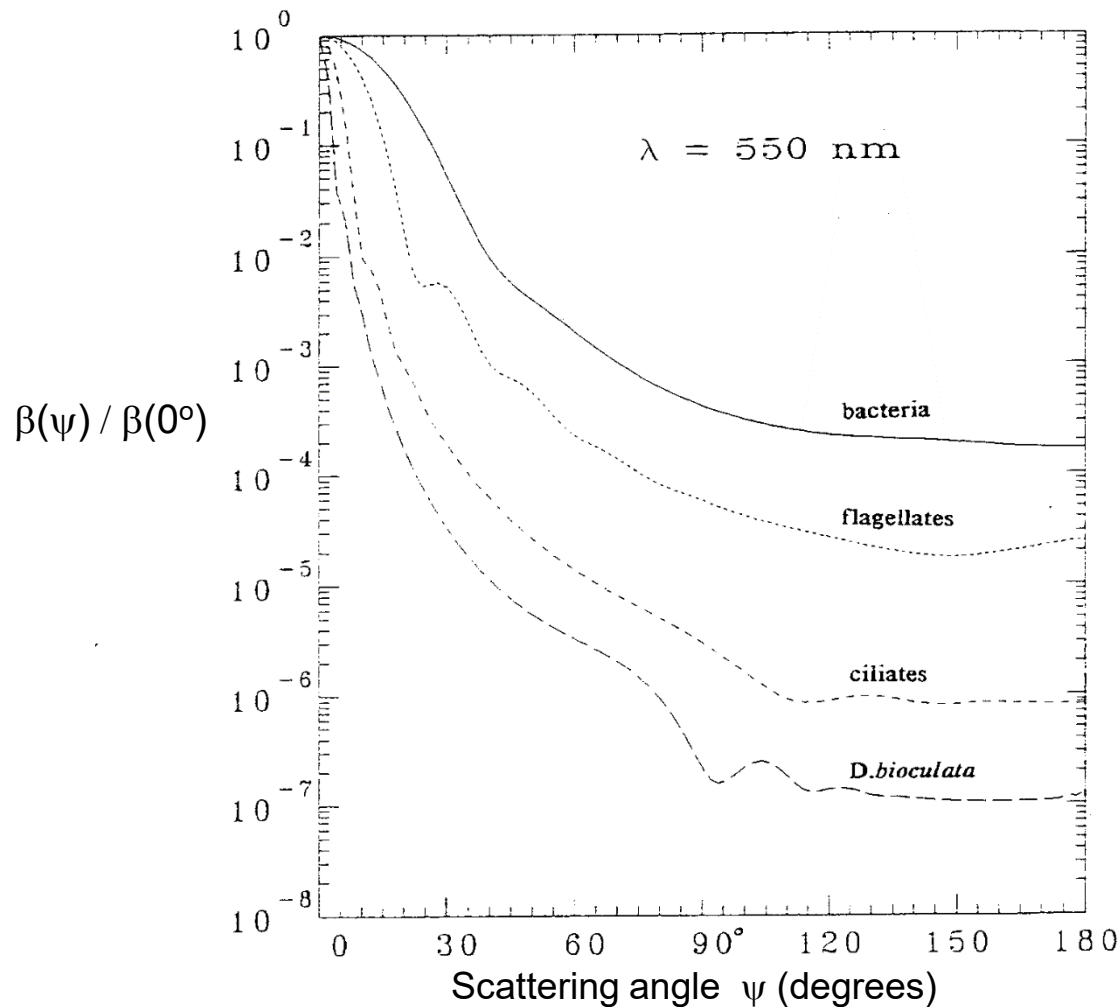


Figure 6. Volume scattering function (normalized at $\theta = 0^\circ$ and for $\lambda = 550$ nm) computed for various organisms by using their refractive index and size distribution as experimentally determined (see text).

Backscattering ratio versus particle size parameter

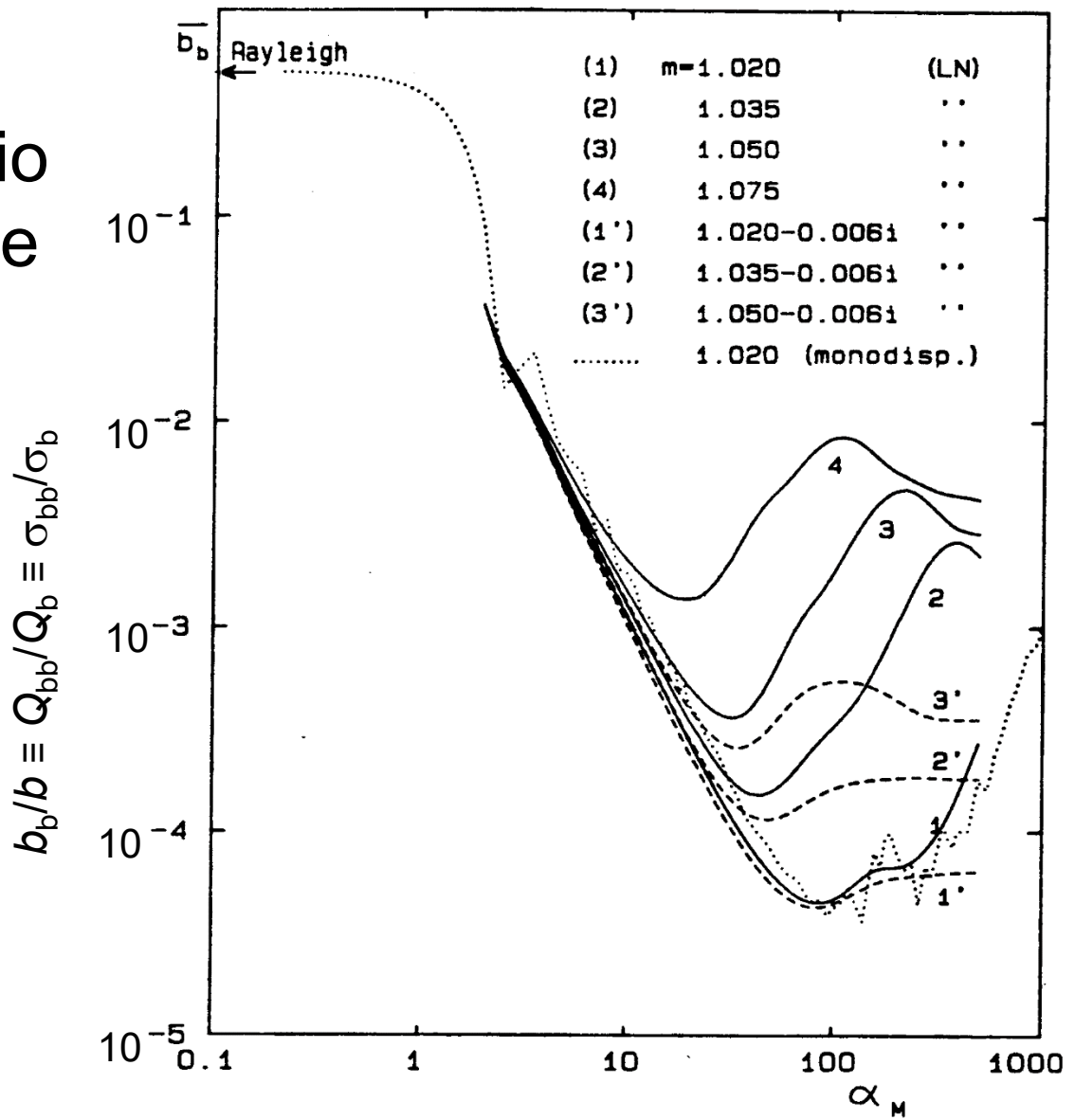


FIG. 8. Variations of the backscattering ratio $\bar{b}_b (= b_b/b)$ vs. the modal relative size α_M (same log-normal law as before in Fig. 5). The different curves correspond to various values of the refractive index given in inset. The curve for a monodispersed population (with $m = 1.02$) is also shown (dotted line). The arrow indicates the limiting value of $b_b/b (=0.5)$ when α tends toward 0 (Rayleigh domain).

INTERSPECIES OPTICAL VARIABILITY OF PLANKTON MICROORGANISMS

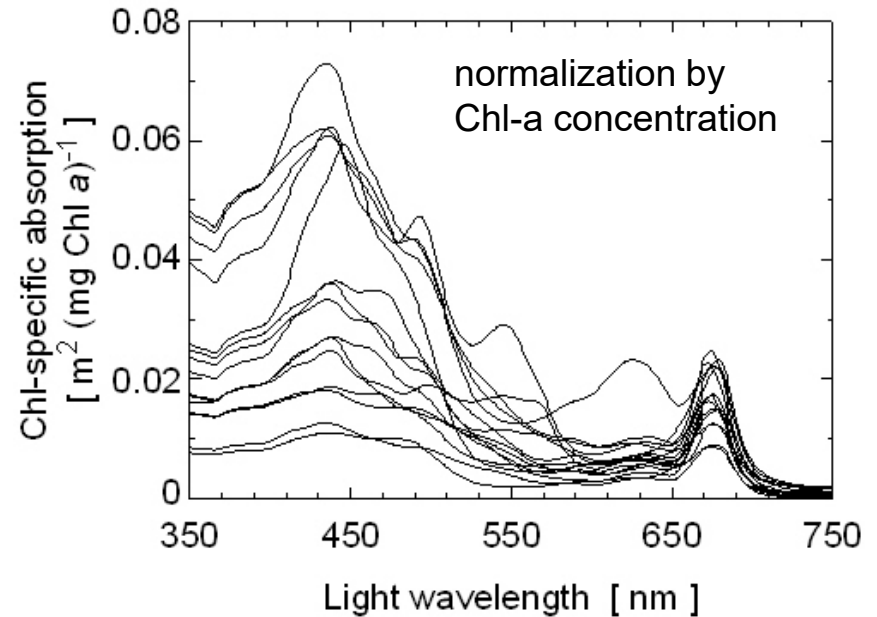
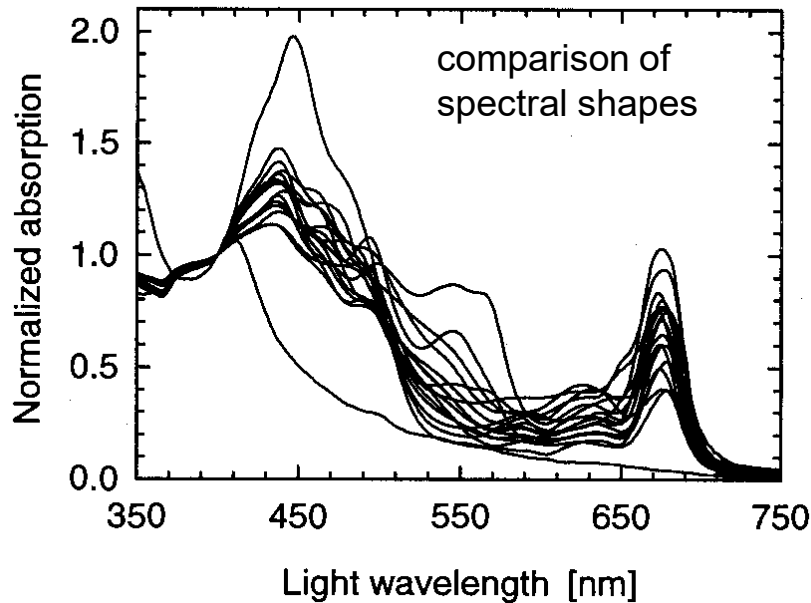
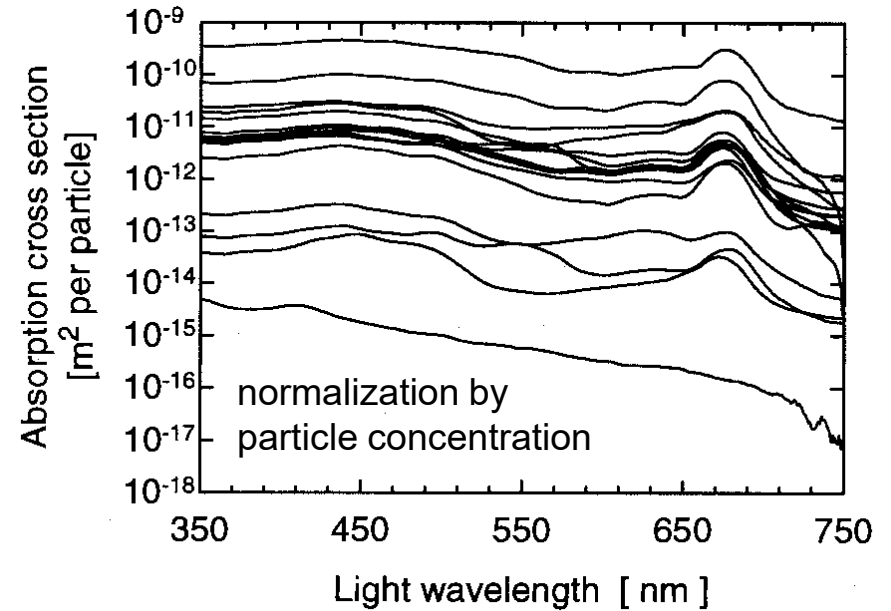
Particle size and complex refractive index are the first-order determinants of interspecies variability of single-particle optical properties

Plankton microorganisms

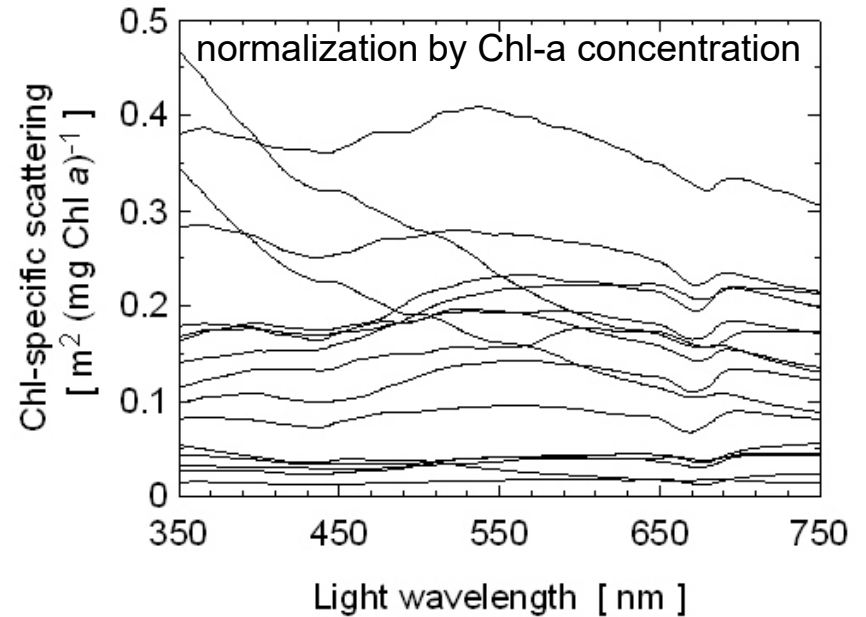
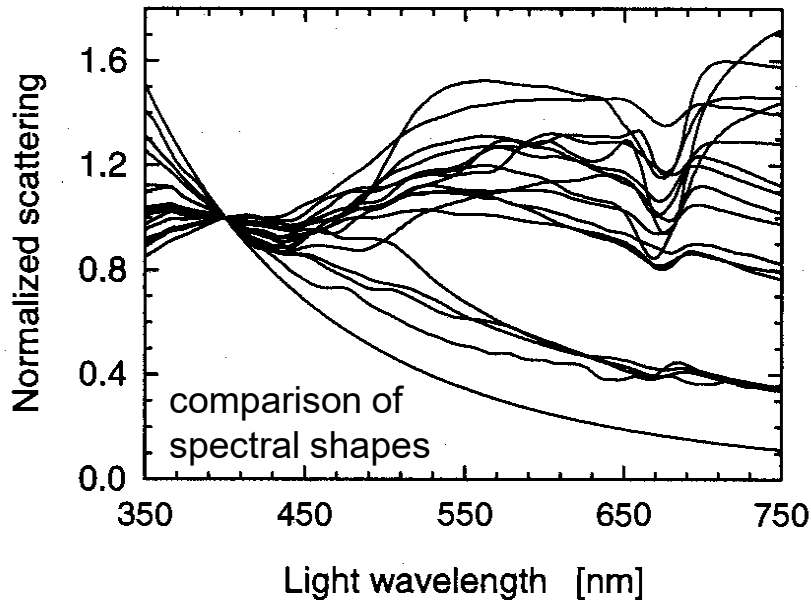
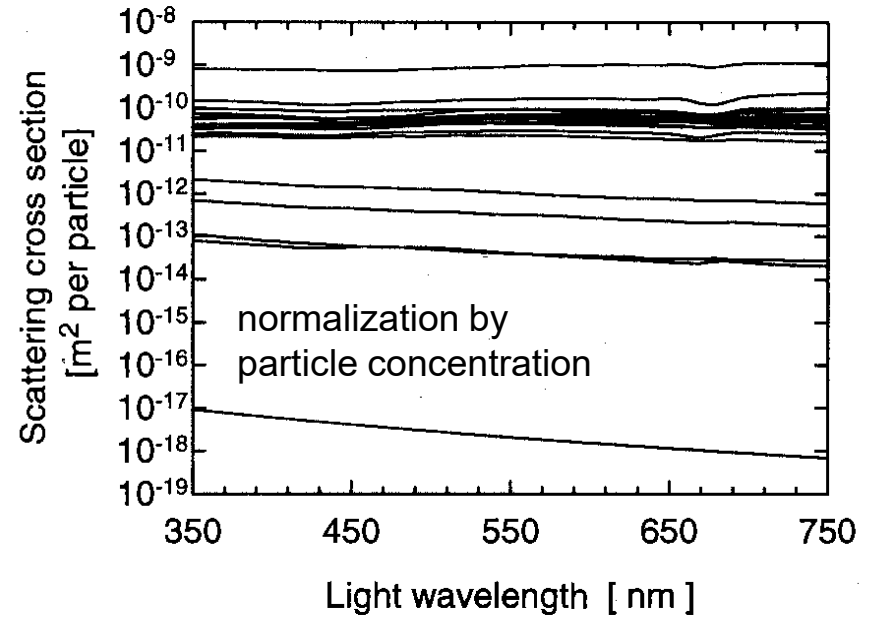
Table 1. Microbial components in the database and source of raw data. Values for the average equivalent spherical diameter (D), the real part of refractive index at 550 nm (n), and imaginary part of refractive index at 440 and 675 nm (n'') are also given for each component.

i	Label	Microbial species	D [μm]	n 550 nm	$n'' \cdot 10^3$ 440 nm	$n'' \cdot 10^3$ 675 nm	Source of raw data
1	VIRU	Viruses	0.07	1.050	0	0	Stramski and Kiefer, 1991
2	HBAC	Heterotrophic bacteria	0.55	1.055	0.509	0.057	Stramski and Kiefer, 1990
3	PROC	generic Prochlorophyte; the average of:	0.66	1.051	18.51	10.30	
		PMED - Prochlorococcus strain MED	0.59	1.055	23.25	13.77	Morel et al., 1993
		PNAS - average of Prochlorococcus strains NATL and SARG	0.70	1.046	13.78	6.687	Morel et al., 1993
4	SYNE	generic Synechococcus; the average of:	1.05	1.051	5.587	2.930	
		SM41 - Synechococcus strain MAX41 (Cyanophyceae)	0.92	1.047	5.415	2.905	Morel et al., 1993
		SM01 - Synechococcus strain MAX01 (Cyanophyceae)	0.94	1.049	4.505	2.547	Morel et al., 1993
		SROS - Synechococcus strain ROS04 (Cyanophyceae)	1.08	1.049	4.516	2.154	Morel et al., 1993
		SDC2 - Synechococcus strain DC2 (Cyanophyceae)	1.14	1.050	4.249	2.375	Morel et al., 1993
		SI03 - Synechococcus strain WH8103 (Cyanophyceae)	1.14	1.062	9.251	4.668	Stramski et al., 1995
5	SYMA	generic phycocyanin-rich picophytoplankton; the average of:	1.41	1.055	6.495	2.757	
		SCYS - Synechocystis (Cyanophyceae)	1.39	1.050	4.530	1.910	Ahn et al., 1992
		MARI - Anacystis marina (Cyanophyceae)	1.43	1.060	8.460	3.603	Ahn et al., 1992
6	PING	Pavlova pinguis (Haptophyceae)	3.97	1.046	4.177	2.709	Bricaud et al., 1988
7	PSEU	Thalassiosira pseudonana (Bacillariophyceae)	3.99	1.045	9.231	7.397	Stramski and Reynolds, 1993
8	LUTH	Pavlova lutheri (Haptophyceae)	4.26	1.045	5.767	2.403	Bricaud et al., 1988
9	GALB	Isochrysis galbana (Haptophyceae)	4.45	1.056	7.673	5.101	Ahn et al., 1992
10	HUXL	Emiliana huxleyi (Haptophyceae)	4.93	1.050	5.012	2.950	Ahn et al., 1992
11	CRUE	Porphyridium cruentum (Rhodophyceae)	5.22	1.051	3.351	2.443	Bricaud et al., 1988
12	FRAG	Chroomonas fragarioides (Cryptophyceae)	5.57	1.039	4.275	2.904	Ahn et al., 1993
13	PARV	Prymnesium parvum (Haptophyceae)	6.41	1.045	2.158	1.329	Bricaud et al., 1988
14	BIOC	Dunaliella bioculata (Chlorophyceae)	6.71	1.038	10.49	7.839	Ahn et al., 1993
15	TERT	Dunaliella tertiolecta (Chlorophyceae)	7.59	1.063	6.260	5.076	Stramski et al., 1993
16	CURV	Chaetoceros curvisetum (Bacillariophyceae)	7.73	1.024	2.877	1.480	Bricaud et al., 1988
17	ELON	Hymenomonas elongata (Haptophyceae)	11.77	1.046	13.87	7.591	Ahn et al., 1992
18	MICA	Prorocentrum micans (Dinophyceae)	27.64	1.045	2.466	1.710	Ahn et al., 1992

Interspecies variability of absorption properties



Interspecies variability of scattering properties

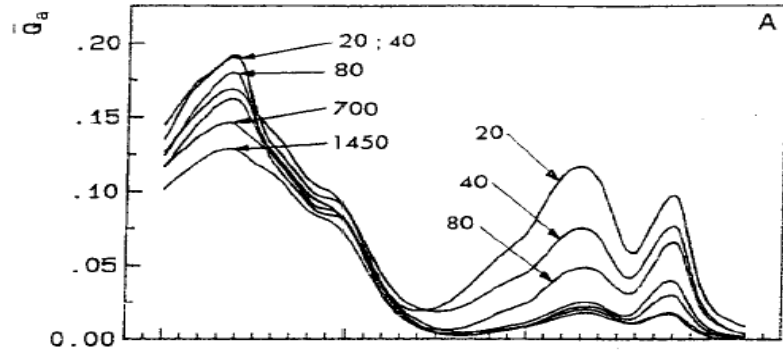


INTRASPECIES OPTICAL VARIABILITY OF PLANKTON MICROORGANISMS

Plankton optical properties vary in response
to varying growth conditions:
light, nutrients, temperature

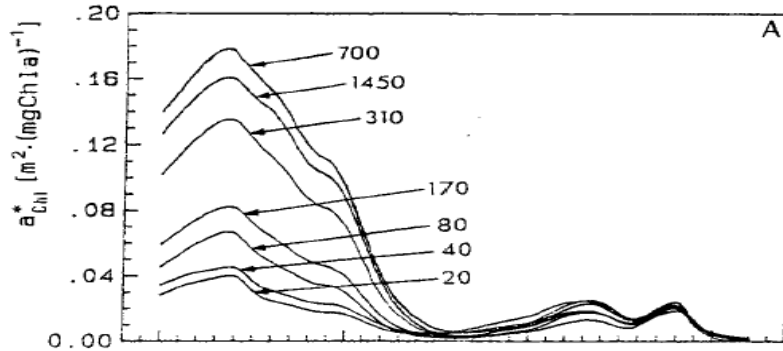
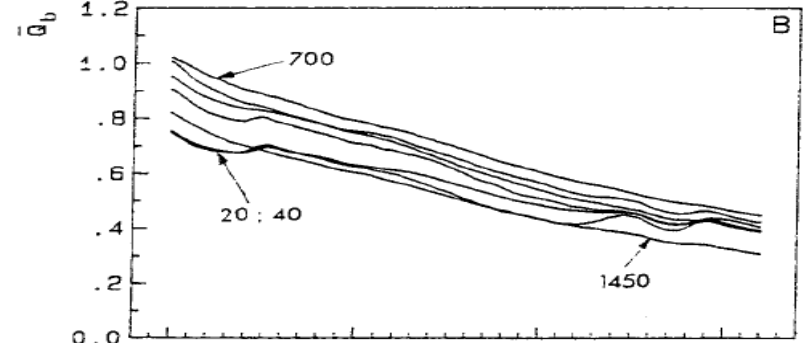
Intraspecies variability due to acclimation to growth irradiance cyanobacteria *Synechocystis*

Absorption

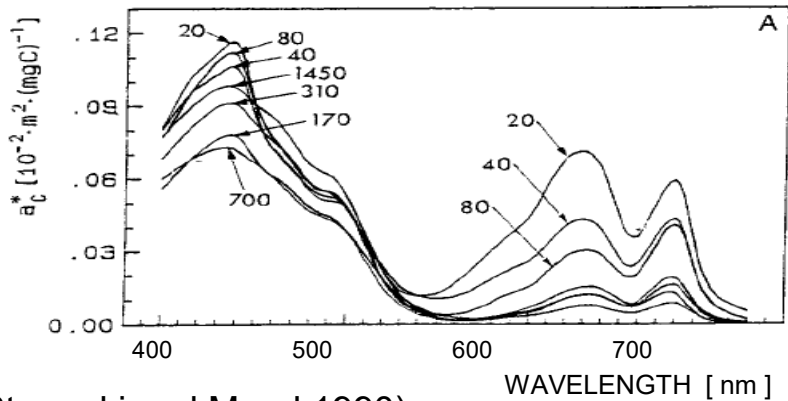
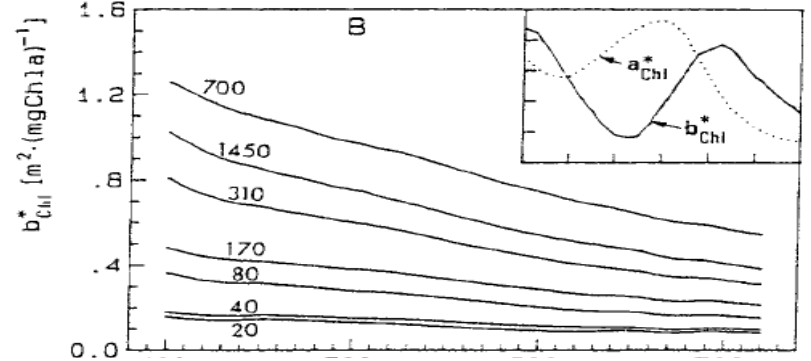


Optical
efficiency

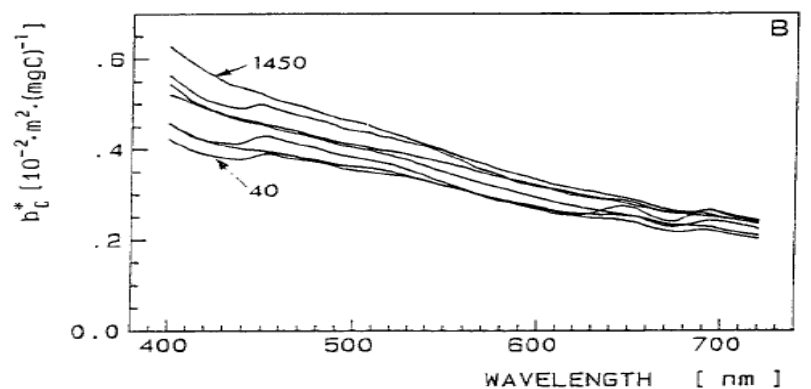
Scattering



Chla-
specific

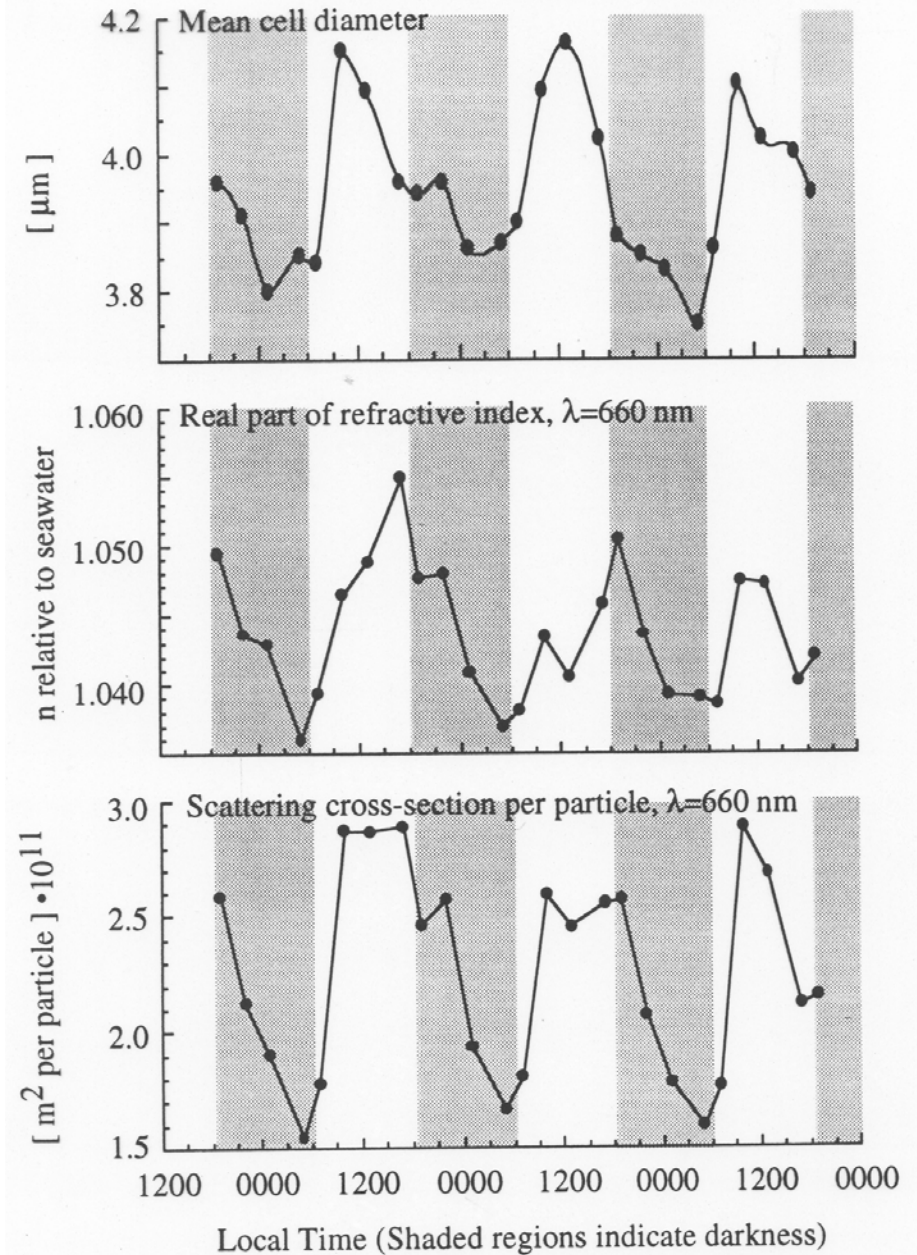


Carbon-
specific



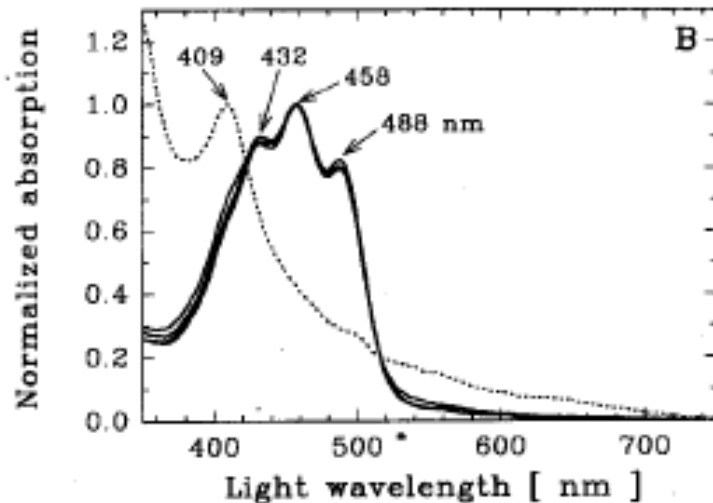
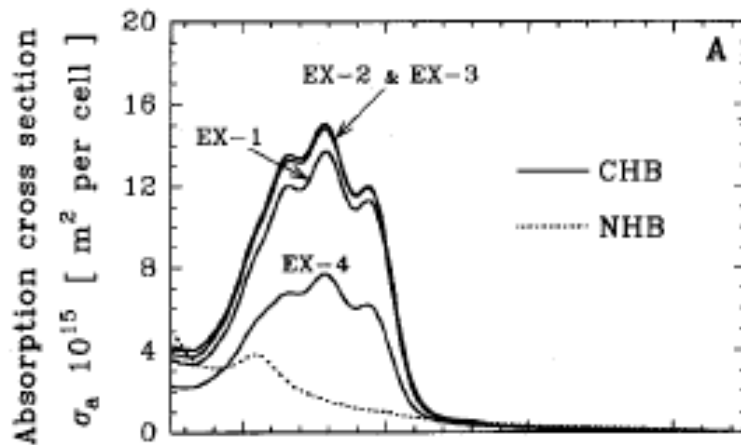
Intraspecies variability over a diel cycle

diatom
*Thalassiosira
pseudonana*

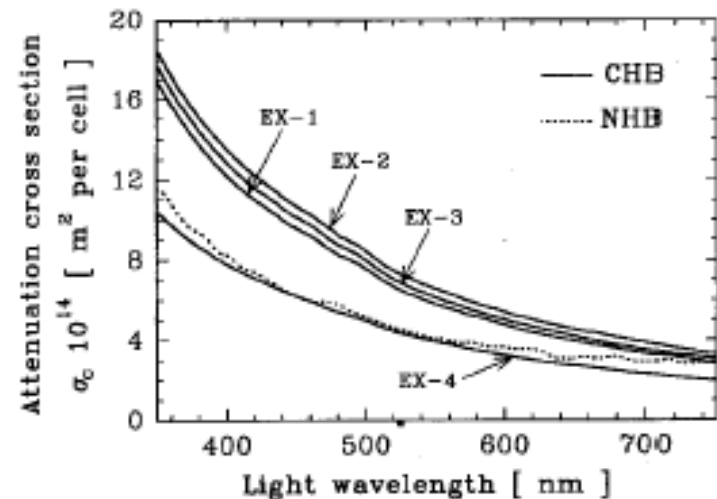


Optical properties of heterotrophic bacteria

Absorption



Beam attenuation



CHB Carotenoid-rich bacteria:
grown in nutrient-enriched seawater [EX-1
(light-dark cycle), EX-2 and EX-3 (dark)],
and in nutrient-poor seawater (EX-4)

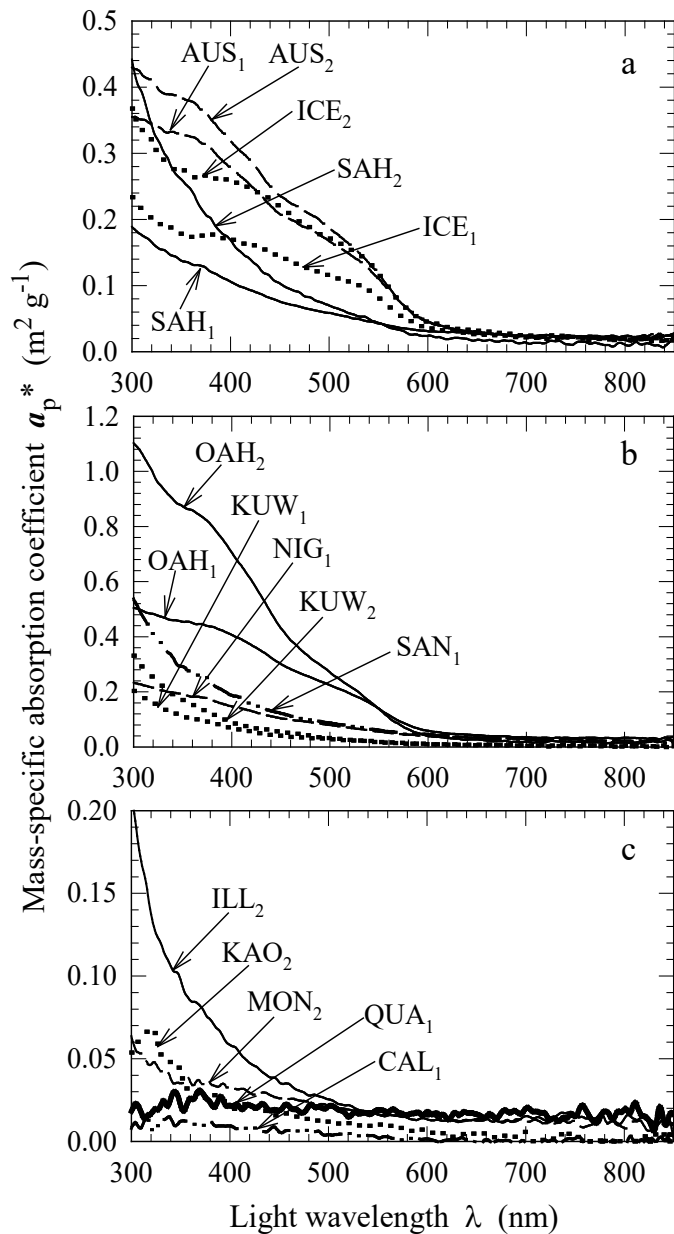
NHB Non-pigmented bacteria:
fast-growing in the absorption experiment
and starved in the attenuation experiment

Optical properties of mineral-rich particulate matter

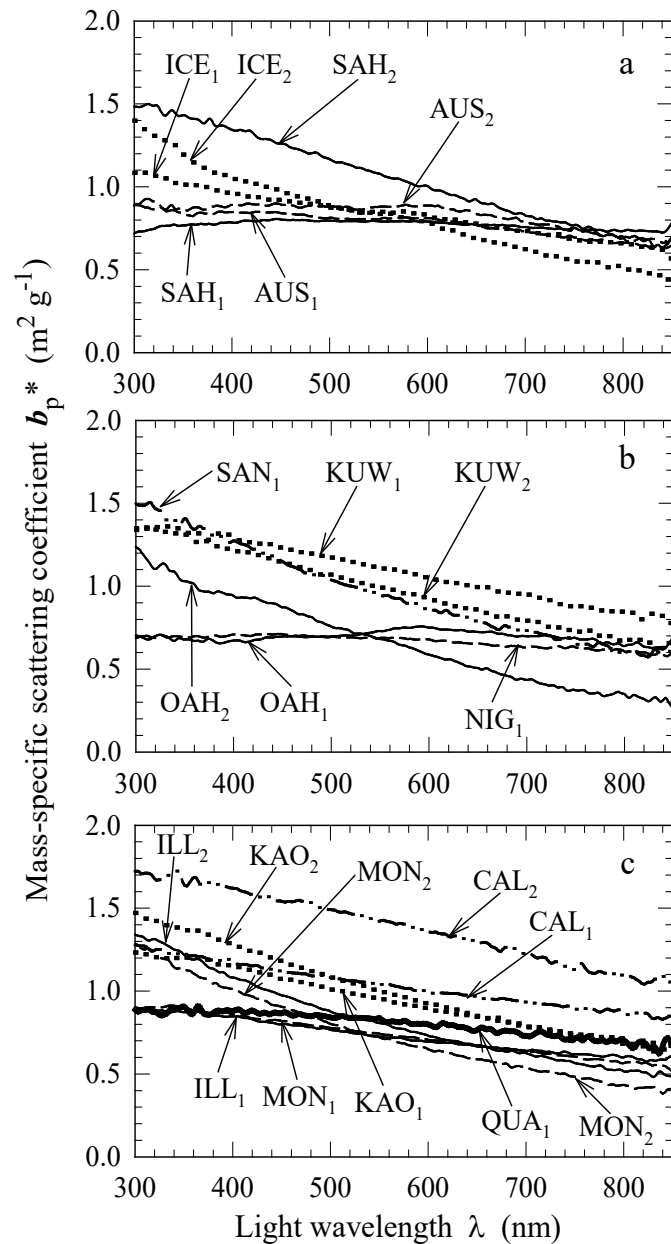
Sample ID	Description	Origin
ILL ₁	illite	Source Clay Minerals Repository, University of Missouri (ref. IMt-1)
ILL ₂	as above but different PSD	as above
KAO ₁	kaolinite (poorly crystallized)	as above (ref. KGa-2)
KAO ₂	as above but different PSD	as above
MON ₁	Ca-montmorillonite	as above (ref. SAz-1)
MON ₂	as above but different PSD	as above
CAL ₁	calcite	natural crystal
CAL ₂	as above but different PSD	as above
QUA ₁	quartz	natural crystal
SAH ₁	atmospheric dust from Sahara	red rain event, Villefranche-sur-Mer, France
SAH ₂	as above but different PSD	as above
AUS ₁	surface soil dust	cliff shore, Palm Beach near Sydney, Australia
AUS ₂	as above but different PSD	as above
ICE ₁	ice-rafted particles	glacier runoff, Kongsfjord, Spitsbergen
ICE ₂	as above but different PSD	as above
OAH ₁	surface soil dust	Oahu, Hawaii Islands
OAH ₂	as above but different PSD	as above
KUW ₁	surface soil dust	Kuwait (eastern part, close to ocean)
KUW ₂	as above but different PSD	as above
NIG ₁	surface soil dust	southwest Nigeria
SAN ₁	atmospheric dust	San Diego, California

(Stramski et al. 2007)

Mass-specific absorption



Mass-specific scattering



(Stramski et al. 2007)



ICE



R36



M1



ALG1



Fe(OH)₃



AUS



R37



M2



ALG2



SAH1



ILL2



R38



M3



NIG1



SAH2



CAL



R39



M4



NIG2



KUW

Particle size distributions in the submicrometer range (TEM data <math> < 0.2 \mu\text{m}</math>)

Transmission
electron
microscopy
(TEM)

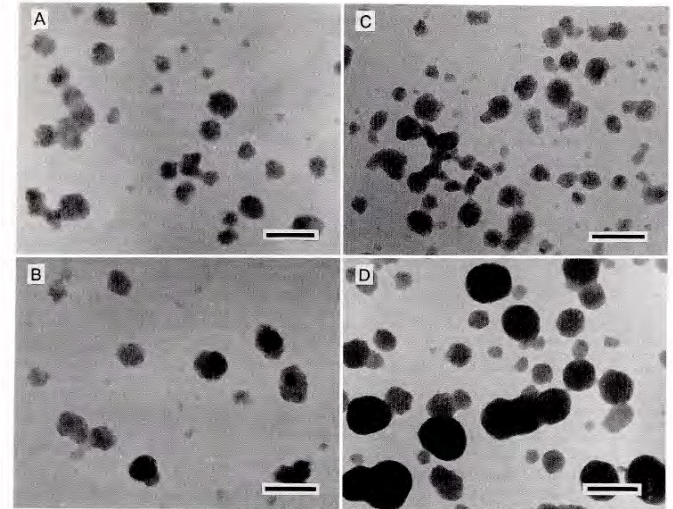
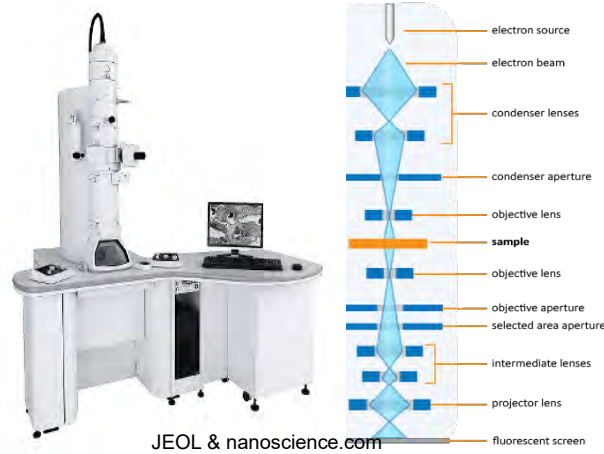
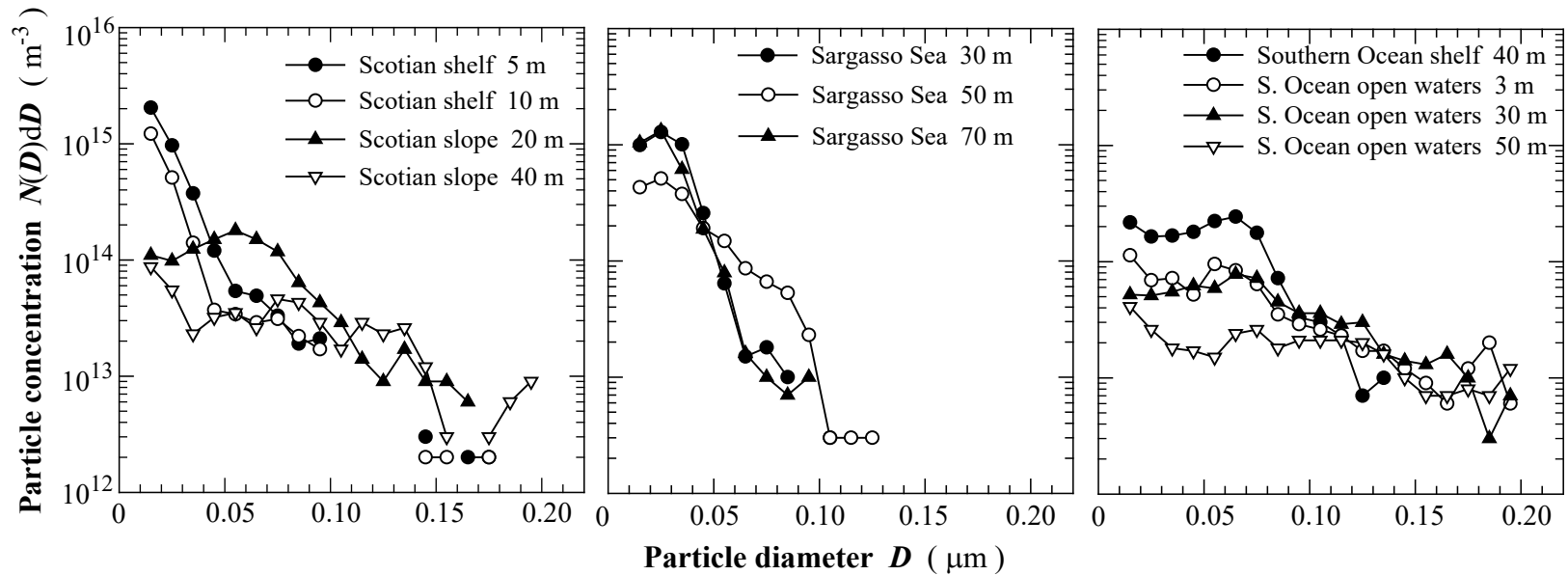
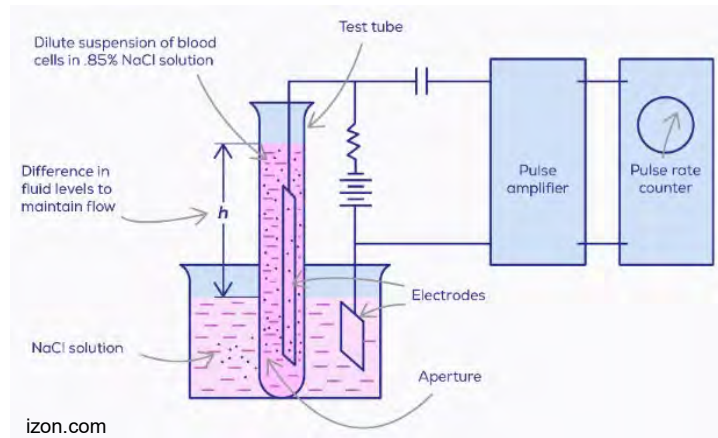


Fig. 3. TEM micrographs of small colloids: A—25-m depth, Scotian Shelf; B—300-m depth, Scotian slope; C—2,500-m depth, Sargasso Sea; D—30-m depth, open-water station in the Southern Ocean. Scale bars are 100 nm.



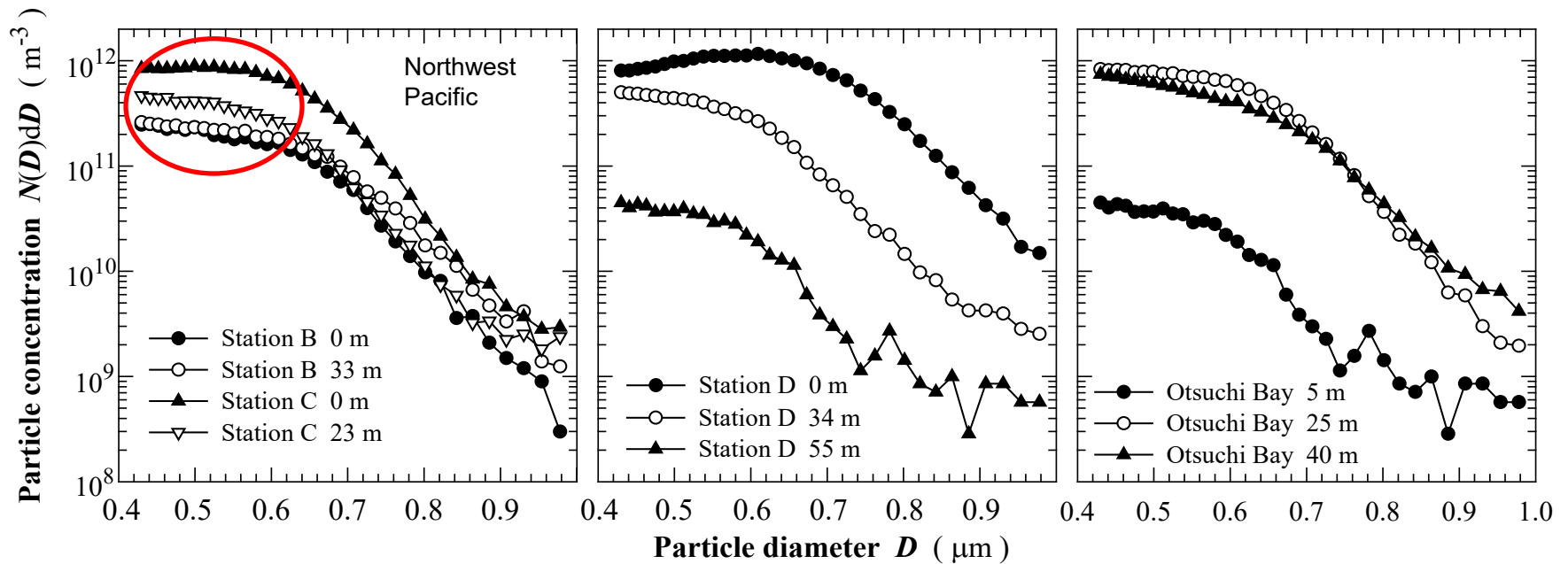
Particle size distributions in the submicrometer range (Coulter Counter data 0.4 – 1 μm)

Coulter Counter Principle

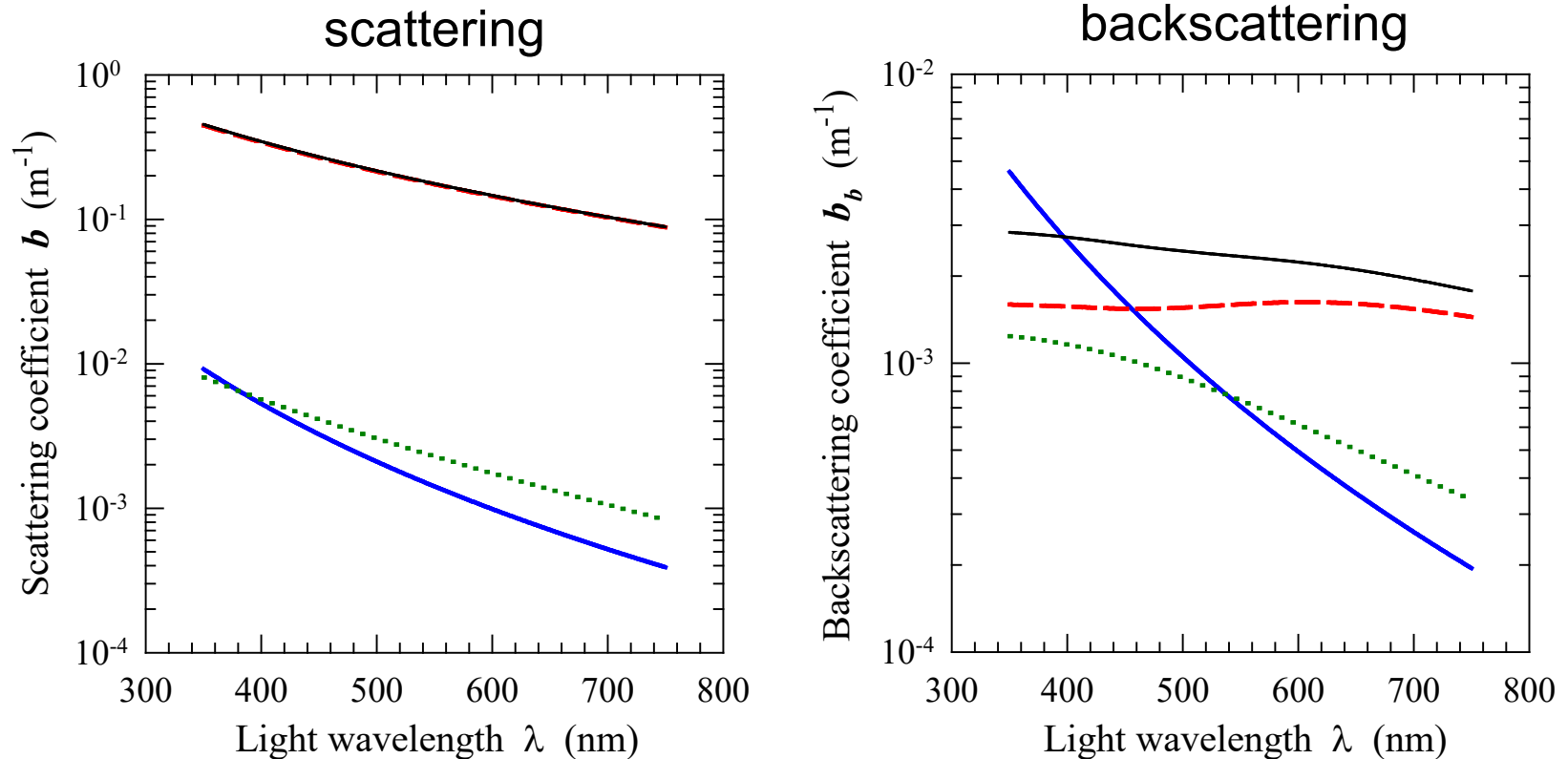


Multisizer 3

Beckmancoulter.com



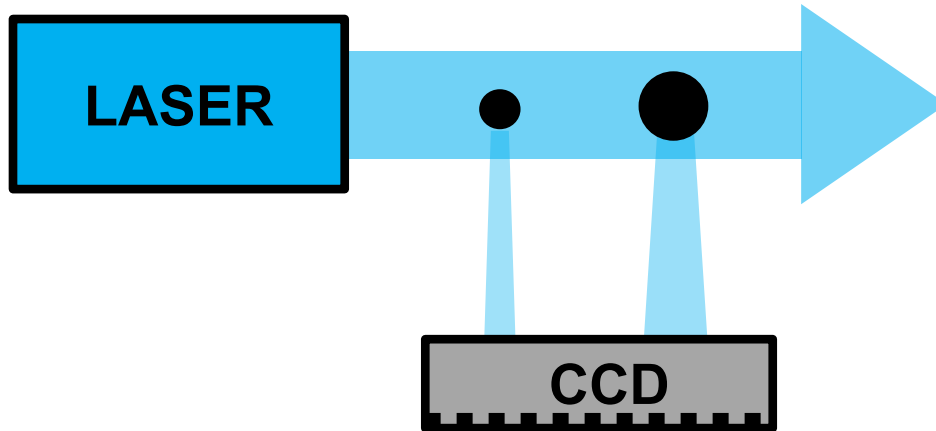
Assessment of scattering and backscattering coefficients of colloidal particles: Comparison with pure seawater



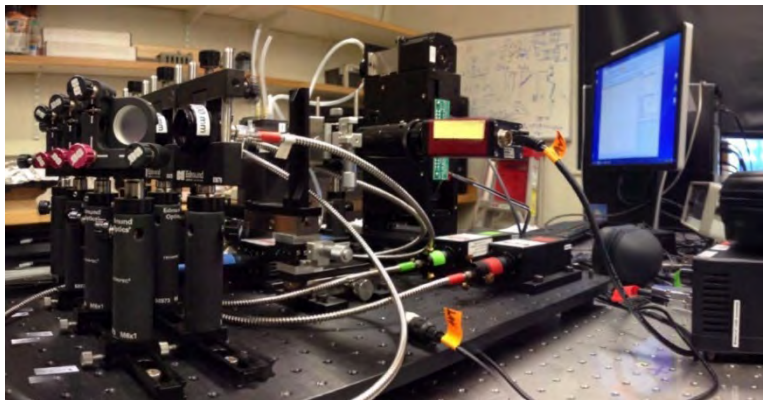
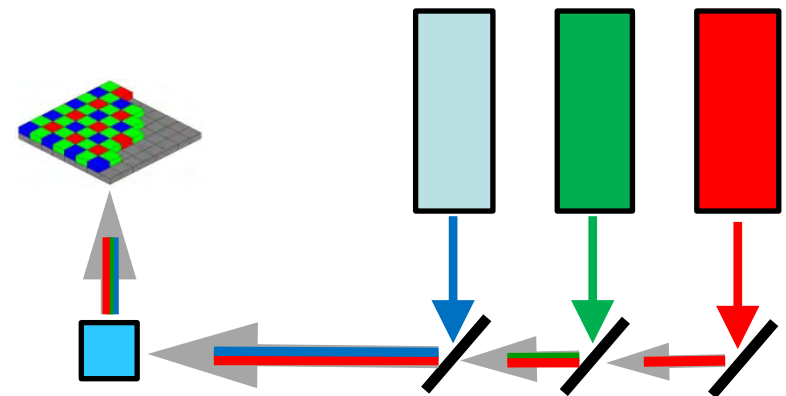
- pure seawater
- ⋯ "average" small colloids (nanoparticles $< 0.2 \mu\text{m}$)
- ⋯ "average" large colloids ($0.2 - 1 \mu\text{m}$)
- small + large colloids

Nanoparticle Tracking Analysis (NTA)

- Nanoparticles illuminated by a beam of light
- Scattered-light images produced by individual nanoparticles are recorded over time
- Individual particle counting yields nanoparticle concentration
- Individual nanoparticle sizes are derived from determinations of mean squared displacement and Brownian diffusion coefficient (Stokes-Einstein equation)

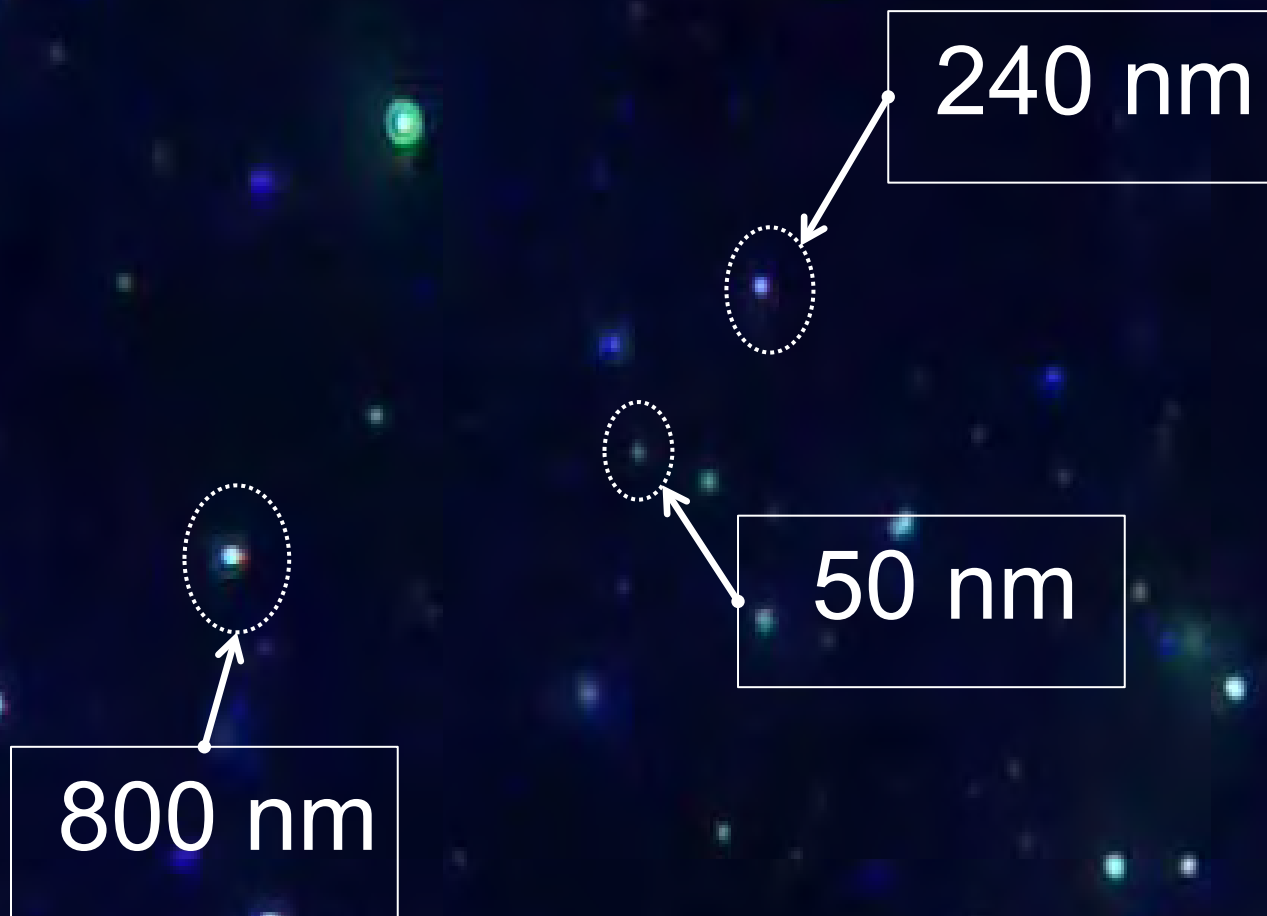


MANTA Multispectral Advanced Nanoparticle Tracking Analysis



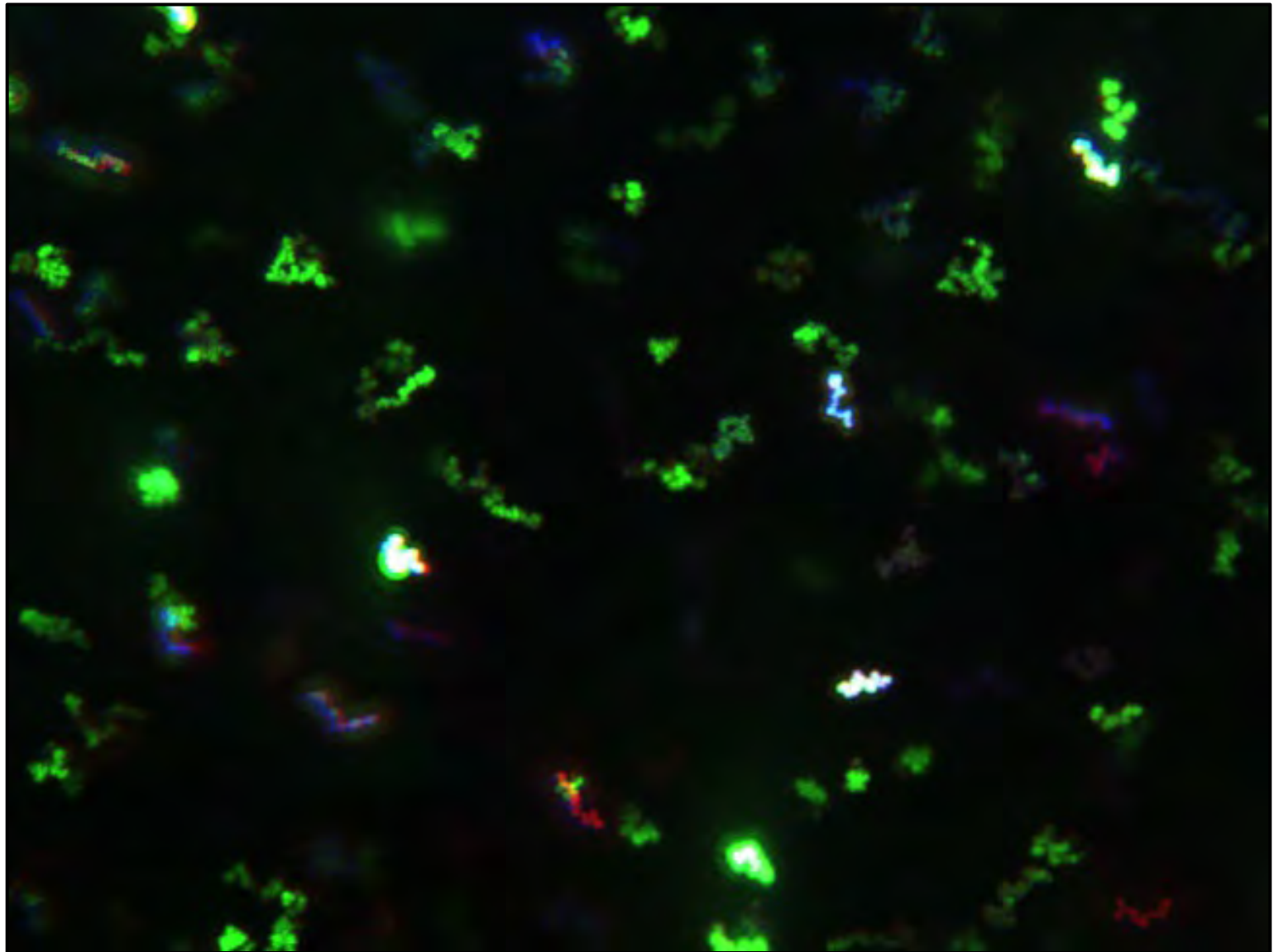
HORIBA
Scientific

ViewSizer 3000



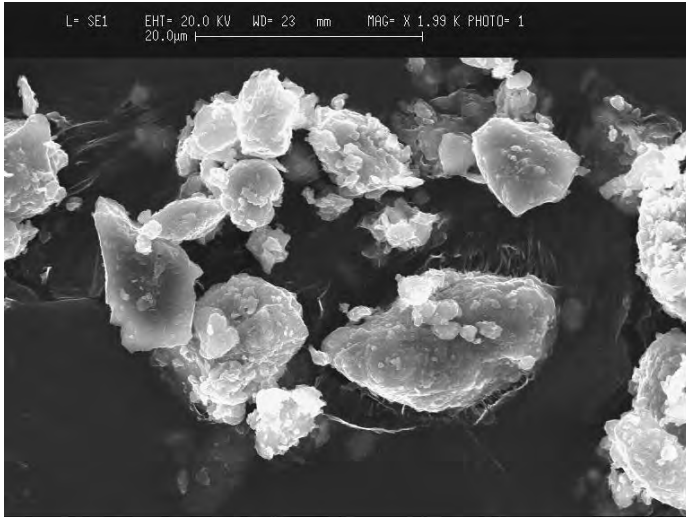
Measurement of a wide range of nanoparticle sizes simultaneously using novel MANTA technology (Multispectral Advanced Nanoparticle Tracking Analysis)

A superposition of 300 video frames acquired during 10 seconds illustrating trajectories of individual nanoparticles through time

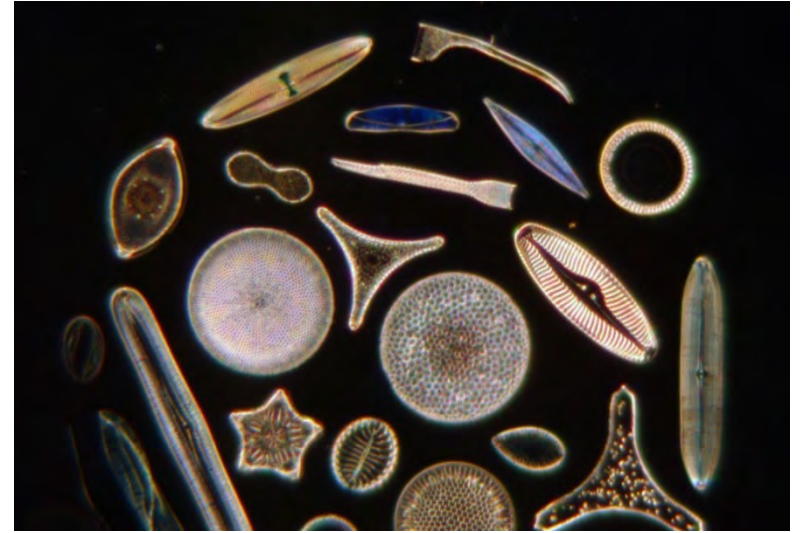


How can we account for large complexity of seawater composition?

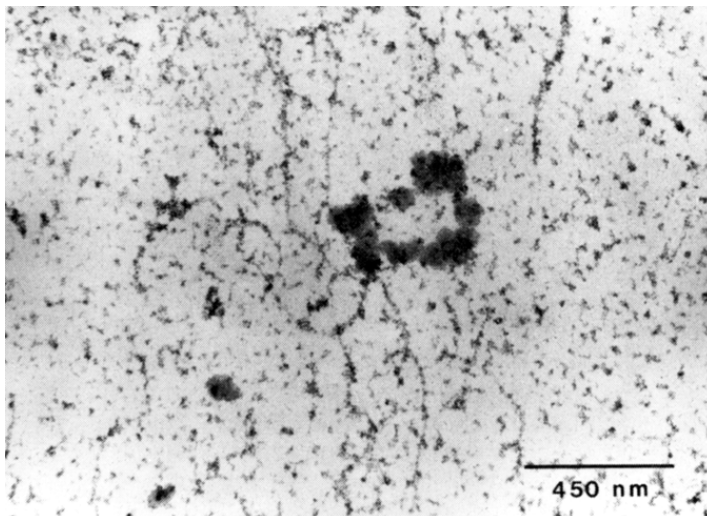
Mineral particles



Plankton microorganisms



Colloids / nanoparticles



Linkage between single-particle and bulk optical properties of particle suspension

$$IOP(\lambda) = \sum_{i=1}^{\gg 1} (NV)_i Q_i(\lambda) G_i = \sum_{i=1}^{\gg 1} (NV)_i \sigma_i(\lambda) \quad \sigma_i(\lambda) = Q_i(\lambda) G_i = IOP_i(\lambda) / (NV)_i$$

Bulk properties:

IOP – a , b , b_b , c , VSF

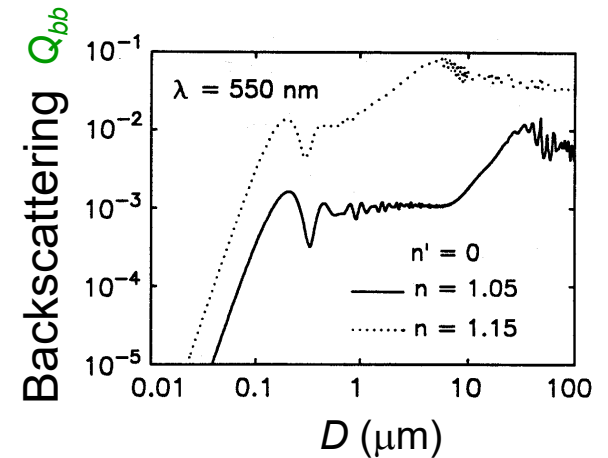
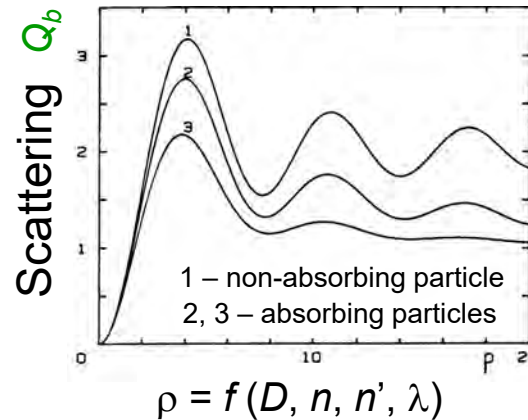
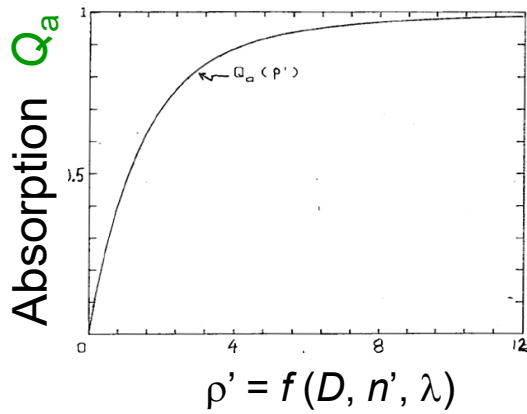
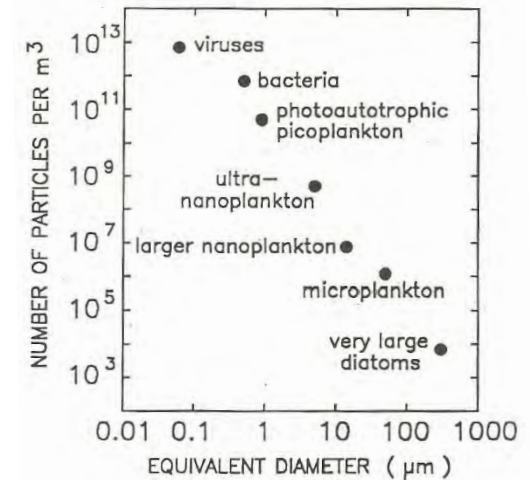
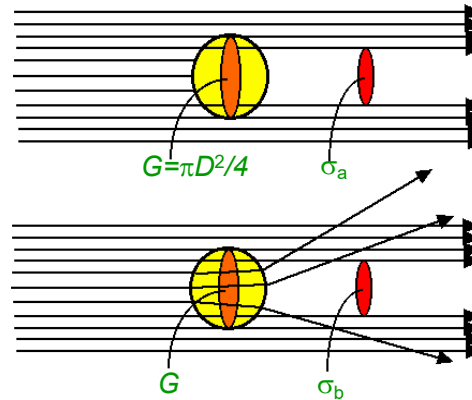
NV – number concentration of particles

Single-particle properties:

Q – optical efficiency factor

G – geometric cross-section

σ – optical cross-section



Chlorophyll-based approach

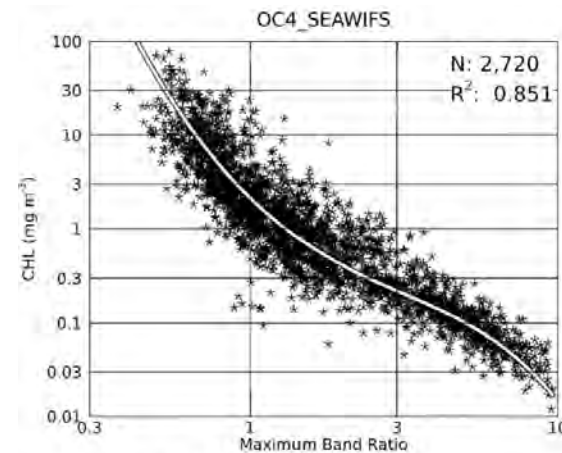
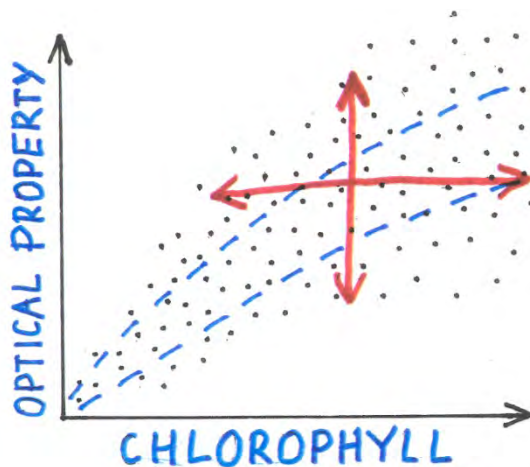
Parameterization of seawater composition in terms of chlorophyll-a concentration alone

$$IOP(\lambda) = IOP_w(\lambda) + f [Chla]$$

$$\text{for example } a_{ph}(\lambda) = f [Chla]$$

$$a_p(\lambda) = f [Chla]$$

$$AOP(\lambda) \text{ (e.g., ocean reflectance)} = f [Chla]$$



MBR=Rrs(443>490>510)/555

a=[0.32814, -3.20725, 3.22969, -1.36769, -0.81739]

(O'Reilly & Werdell 2019)

Traditional approach with a few IOP components

Inherent Optical Properties (IOPs) are described in terms of a few broadly-defined categories of seawater constituents amenable to measurements

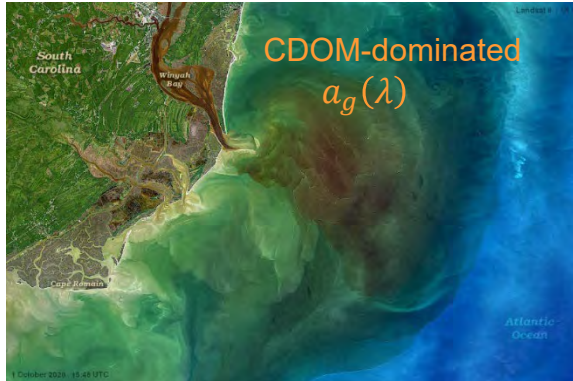
$$IOP(\lambda) = IOP_w(\lambda) + IOP_p(\lambda) + IOP_{CDOM}(\lambda)$$

$$IOP_p(\lambda) = IOP_{ph}(\lambda) + IOP_{NAP}(\lambda)$$

pure water (w), all particles (p), phytoplankton (ph), non-algal/detrital particles (NAP or d), chromophoric dissolved organic matter (CDOM or g)

Basic IOPs: absorption, scattering, and beam attenuation coefficients, volume scattering function

$$\text{Ocean Color } R_{rs}(\lambda) \propto \frac{b_b(\lambda)}{a(\lambda)} = \frac{b_{bw}(\lambda) + b_{bp}(\lambda)}{a_w(\lambda) + a_{ph}(\lambda) + a_d(\lambda) + a_g(\lambda)}$$

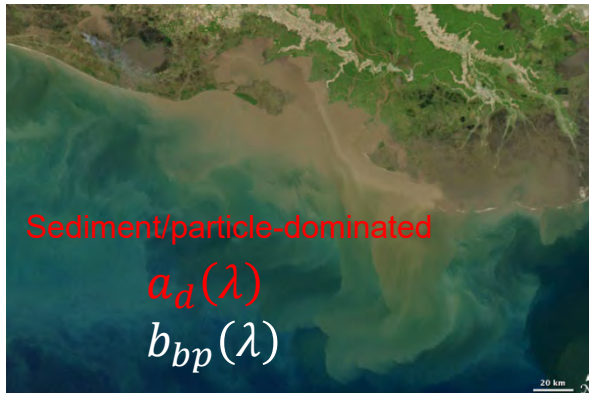


South Carolina coast, Landsat 8, October 1, 2020 (NASA Ocean Color Web)

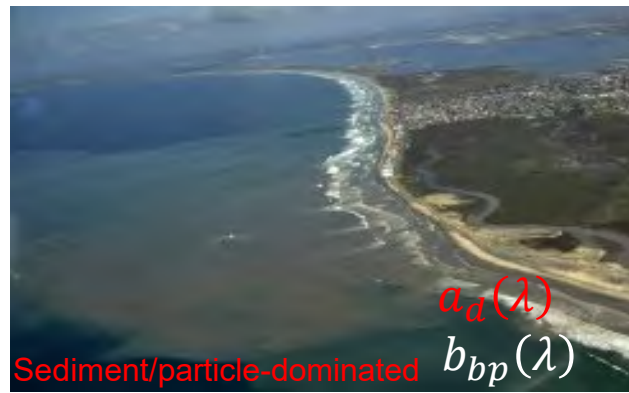


Halifax Bay, Eastern Australia
<https://blogs.ntu.edu.sg/science/2020/06/24/lights-out-for-muddy-water-coral-reefs-as-global-sea-level-rises/>

Phytoplankton-dominated



Atchafalaya River plume, Gulf of Mexico, MODIS-Aqua, April 7, 2009
<https://earthobservatory.nasa.gov/images/38273/sediment-in-the-gulf-of-Mexico>



Tijuana River plume, Imperial Beach, California
<https://giddingslab.ucsd.edu/research/coastal-ocean/small-plume-dispersion/>



Lingulodinium bloom, off California coast (<http://oceandatacenter.ucsc.edu/PhytoGallery/harmful-algae.html>)

Reductionist approach

To develop an understanding and assemble a model of the whole, from the reductionist study of its parts

$$\begin{aligned} IOP_p(\lambda) &= \sum_k IOP_{k, pla}(\lambda) && \text{plankton} \\ &+ \sum_m IOP_{m, min}(\lambda) && \text{minerals} \\ &+ \sum_n IOP_{n, det}(\lambda) && \text{detritus} \end{aligned}$$

Example IOP model with detailed description of plankton community

For example, for absorption we have:

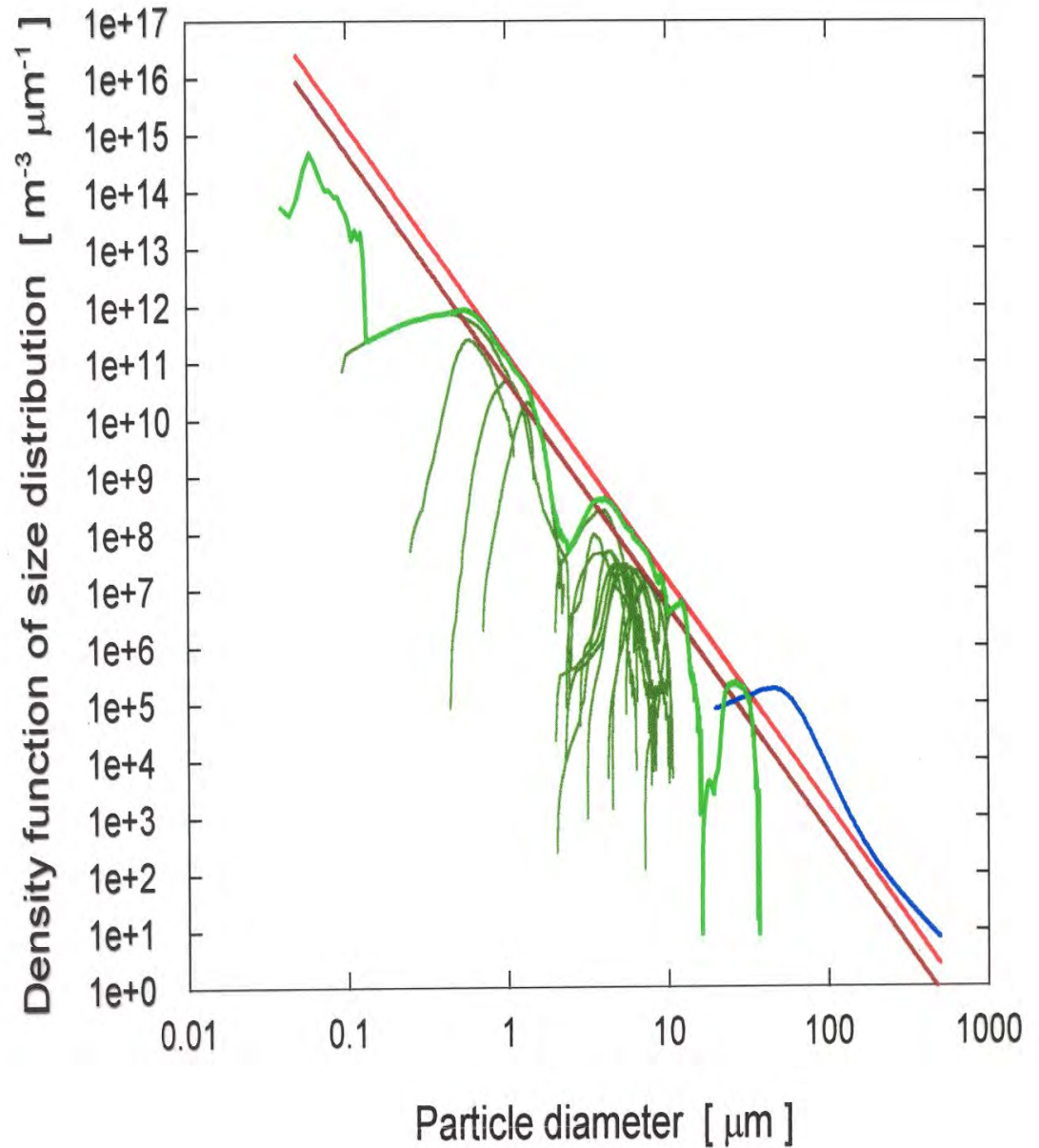
$$a_{all} = \sum (N/V)_i (\sigma_a)_i$$

where the sum includes *all* species/groups of microorganisms and other particles, each denoted by subscript *i*

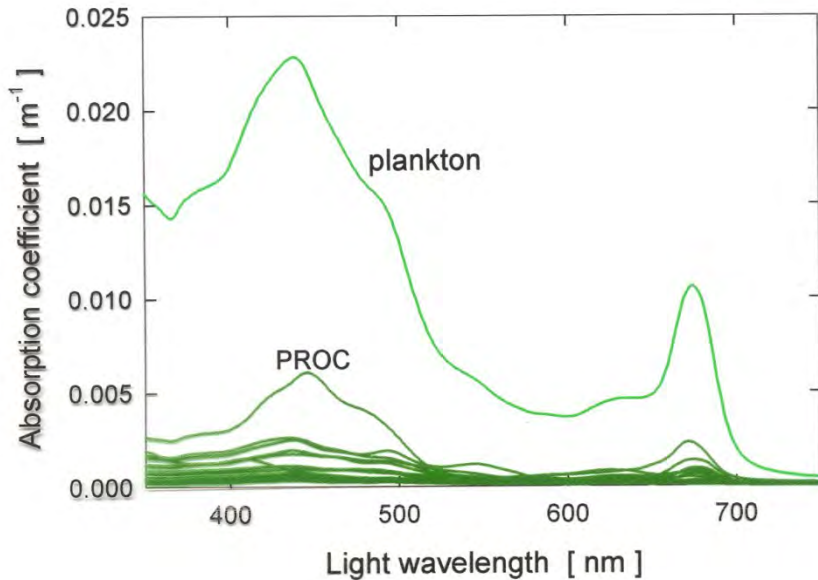
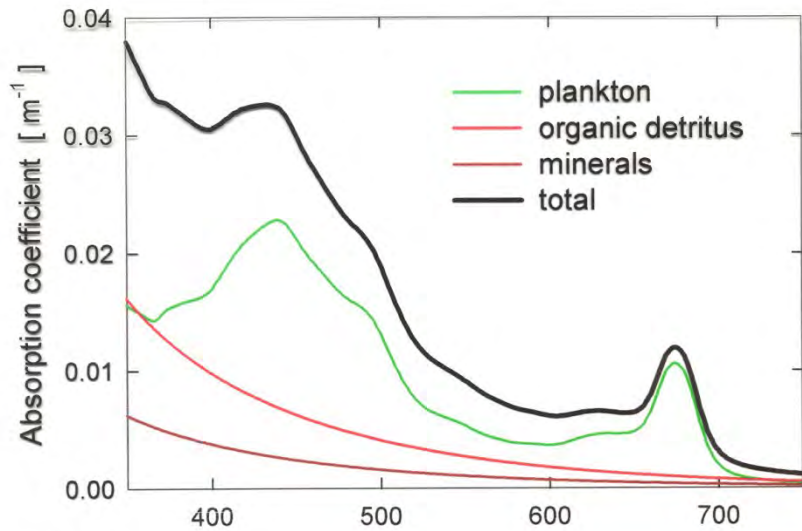
<i>i</i>	Component	Concentration [particles/m ³]	<i>Chl</i> [mg m ⁻³]
1	VIRU	1.0 · 10 ¹³	0
2	HBAC	4.0 · 10 ¹¹	0
3	PROC	7.0 · 10 ¹⁰	0.1026
4	SYNE	2.0 · 10 ¹⁰	0.0403
5	SYMA	8.0 · 10 ⁹	0.0360
Σ	Picoplankton	4.98 · 10 ¹¹	0.1789
6	PING	4.5056 · 10 ⁸	0.0540
7	PSEU	0.9808 · 10 ⁸	0.0303
8	LUTH	0.9924 · 10 ⁸	0.0107
9	GALB	0.4839 · 10 ⁸	0.0155
10	HUXL	0.4339 · 10 ⁸	0.0104
11	CRUE	0.4496 · 10 ⁸	0.0129
12	FRAG	0.4768 · 10 ⁸	0.0157
13	PARV	0.6247 · 10 ⁸	0.0181
14	BIOC	0.3966 · 10 ⁸	0.0900
15	TERT	0.3570 · 10 ⁸	0.0609
16	CURV	0.2987 · 10 ⁸	0.0099
Σ	Small Nanoplankton	1.0 · 10 ⁹	0.3284
17	ELON	1.7 · 10 ⁷	0.1595
18	MICA	2.0 · 10 ⁶	0.0508
Σ	Total Plankton	1.0499019 · 10 ¹³	0.7176
19	DET	3.3 · 10 ¹⁴	0
20	MIN	1.1 · 10 ¹⁴	0
Σ	Total Non-living Particles	4.4 · 10 ¹⁴	0
21	BUB	7.1 · 10 ⁶	0

Size distribution

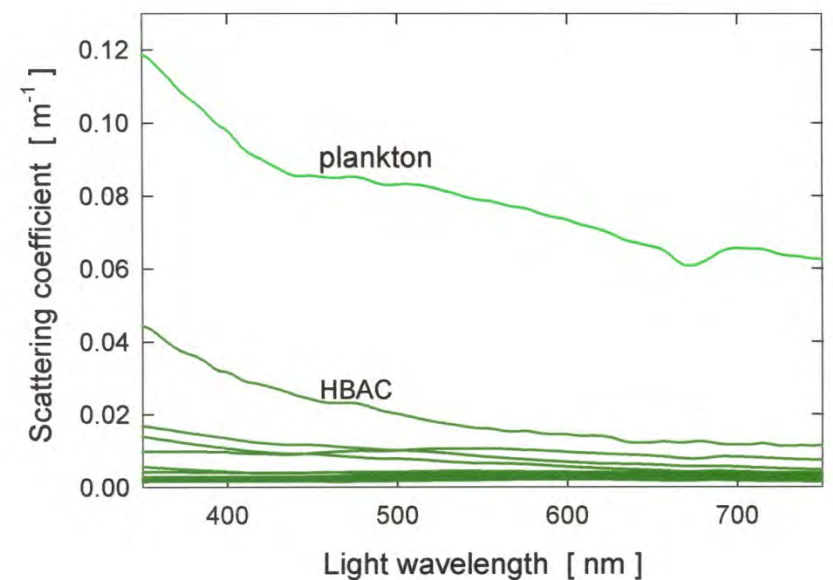
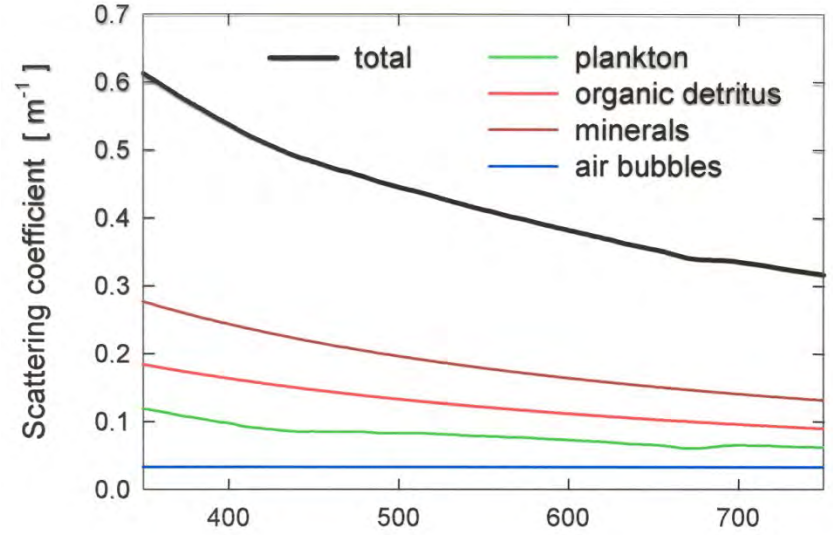
- 18 planktonic components
- composite plankton
- mineral particles
- organic detritus
- air bubbles



Absorption budget



Scattering budget



Reductionist radiative transfer/reflectance model

Input to radiative transfer model

$$IOP(\lambda) = \sum_{i=1}^j IOP_i(\lambda) = \sum_{i=1}^j N_i \overline{\sigma}_i(\lambda)$$

Output, e.g. ocean reflectance

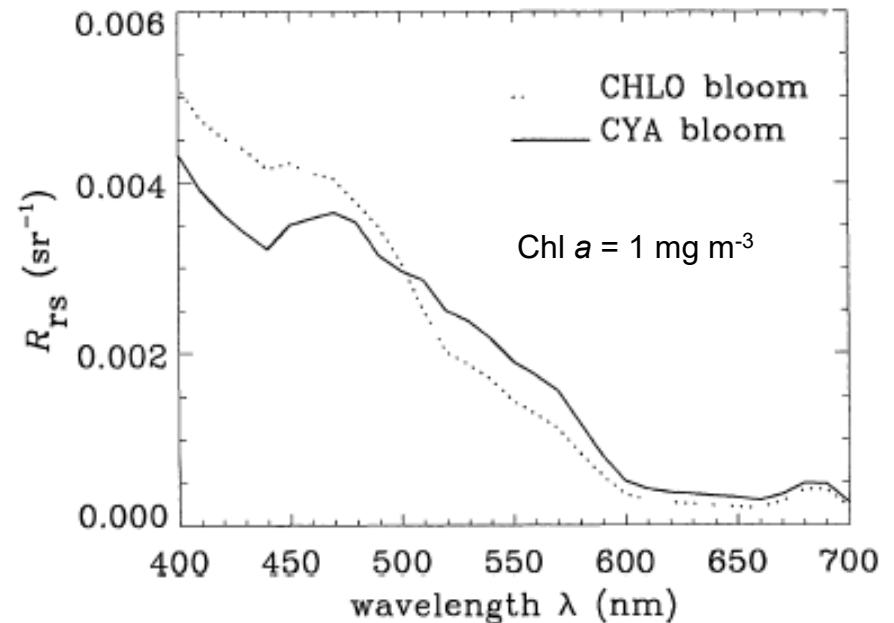
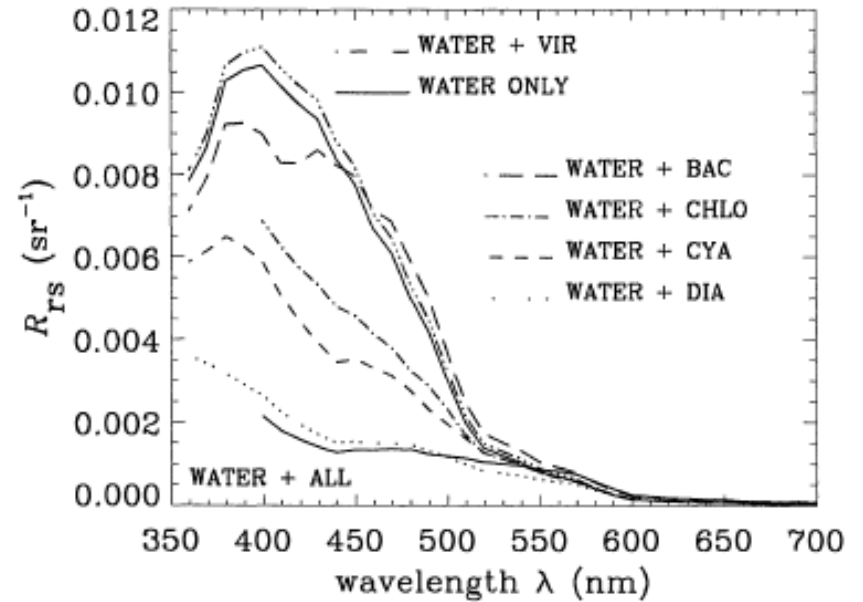
$$R(\lambda) = f \left[\sum_{i=1}^j N_i \overline{\sigma}_{i,a}(\lambda), \sum_{i=1}^j N_i \overline{\sigma}_{i,b}(\psi, \lambda) \right]$$

- In what ways does variability in detailed seawater composition determine variability in ocean reflectance?
- What information about water constituents and optical properties can we hope to extract from remotely sensed reflectance with acceptable accuracy?

Example combination of reductionist IOP model and radiative transfer model for simulating ocean color

- Viruses ($\sim 0.07 \mu\text{m}$ in size)
- Heterotrophic bacteria ($\sim 0.5 \mu\text{m}$)
- Cyanobacteria ($\sim 1 \mu\text{m}$)
- Small diatoms ($\sim 4 \mu\text{m}$)
- Chlorophytes ($\sim 8 \mu\text{m}$)
- Detritus
- CDOM

Stramski and Mobley (1997)
Mobley and Stramski (1997)



The complexity of seawater as an optical medium should not deter us from pursuing the proper course in future research

“The reductionist worldview has to be accepted as it is, not because we like it, but because that is the way the world works”

Steven Weinberg, Dreams of a Final Theory (1992)
1979 Nobel Prize in Physics