



Inherent Optical Properties (IOPs)

Lecture 1

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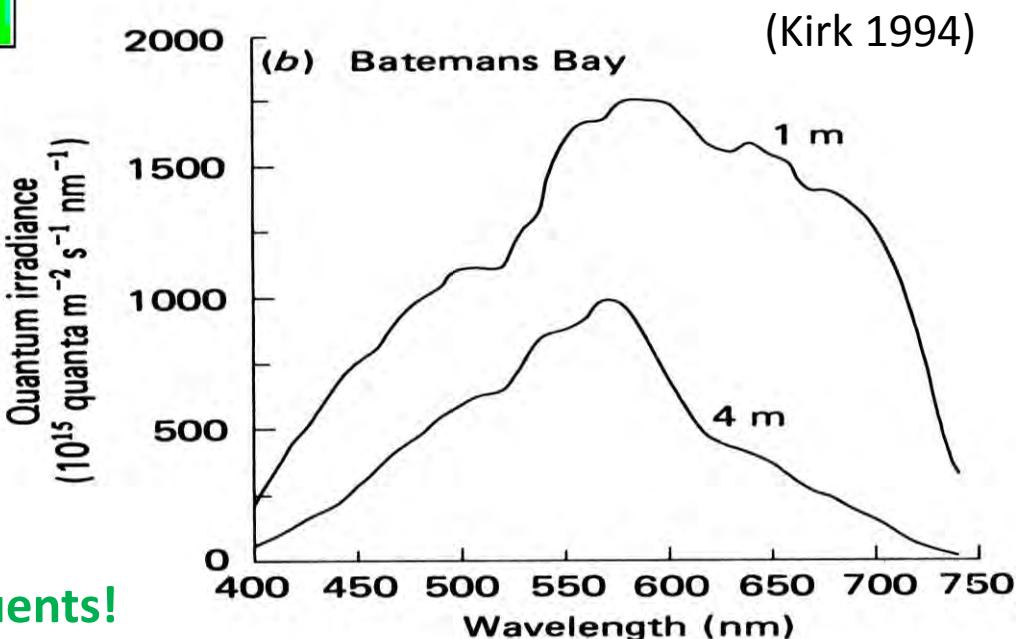
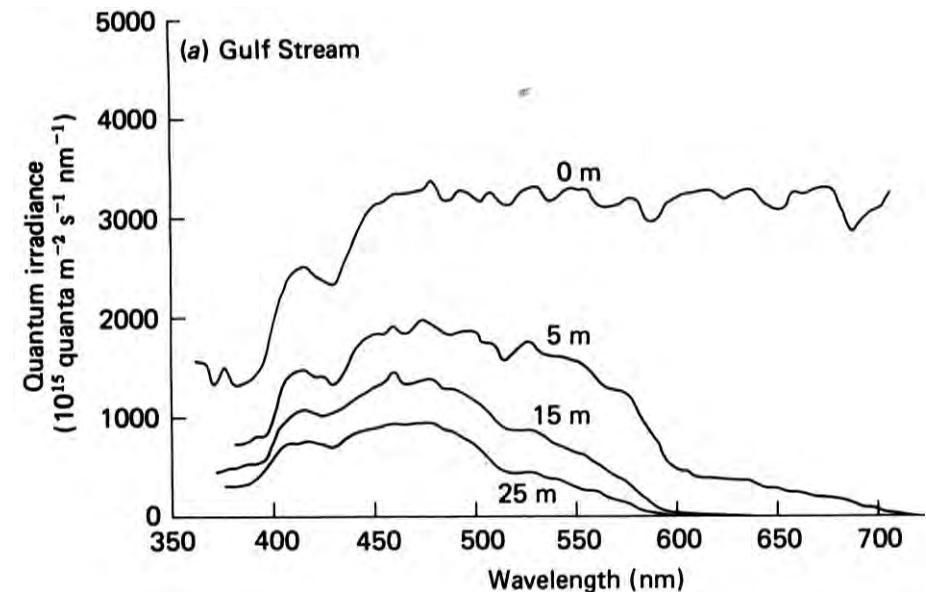
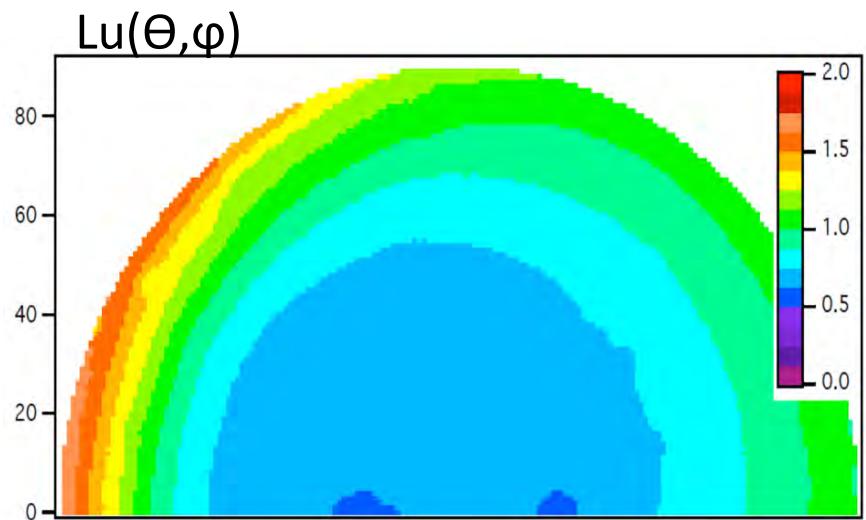
Absorption properties

Scattering properties

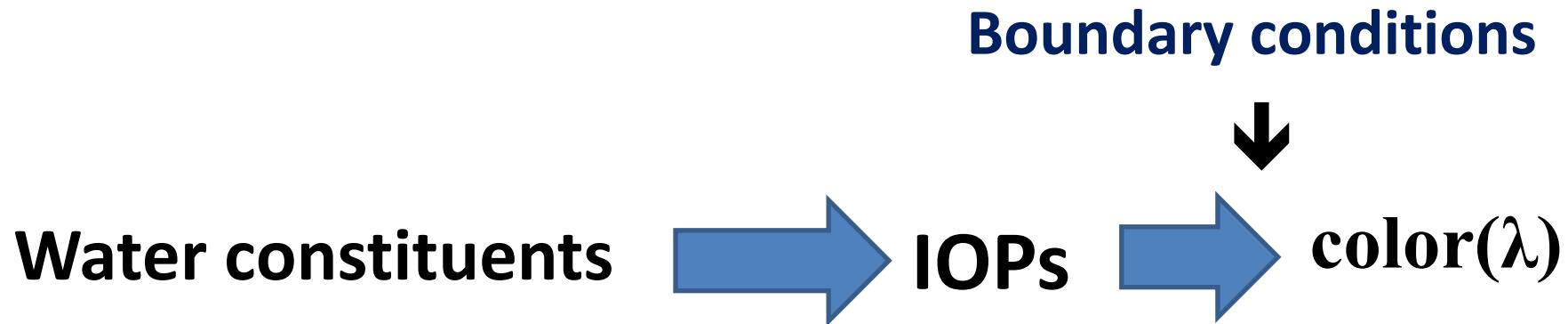
ocean (water) color



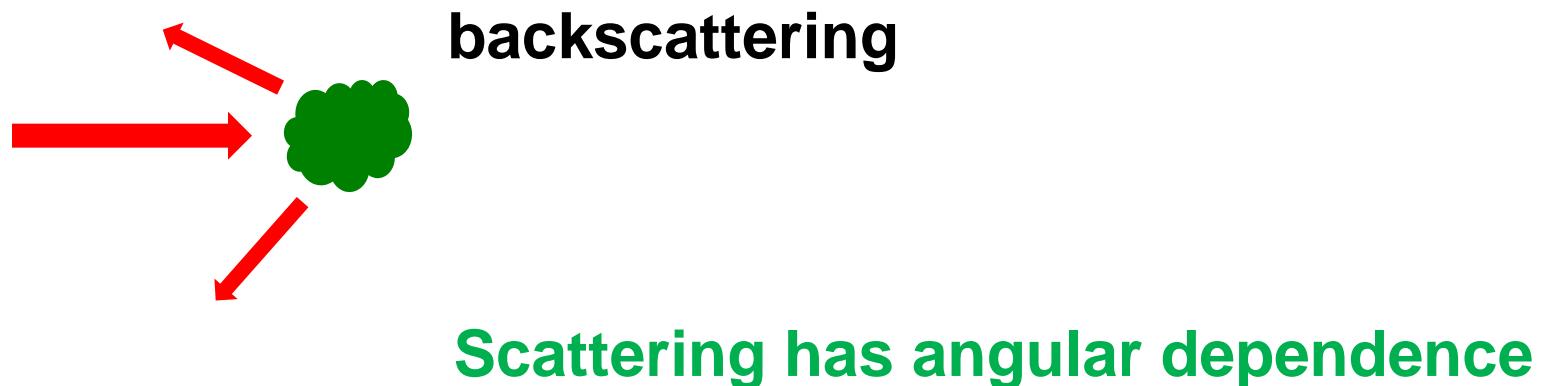
light within water medium



They are modulated by water constituents!



Primary IOPs:



absorption coefficient: a (m⁻¹)

Volume Scattering Function (VSF): β (m⁻¹ sr⁻¹)



Scattering coefficient: b (m⁻¹)

forward-scattering coefficient: b_f (m⁻¹) → $b_f = 2\pi \int_0^{\pi/2} \beta \sin(\theta) d\theta$

$$b = 2\pi \int_0^{\pi} \beta \sin(\theta) d\theta$$

backward-scattering coefficient: b_b (m⁻¹) → $b_b = 2\pi \int_{\pi/2}^{\pi} \beta \sin(\theta) d\theta$

beam attenuation coefficient: $c = a + b$ (m-1)

IOPs are additive.

$$a = a_w + \sum a_{xi}$$

$$b = b_w + \sum b_{xi}$$

D. Stramski et al. / Progress in Oceanography 61 (2004) 27–56

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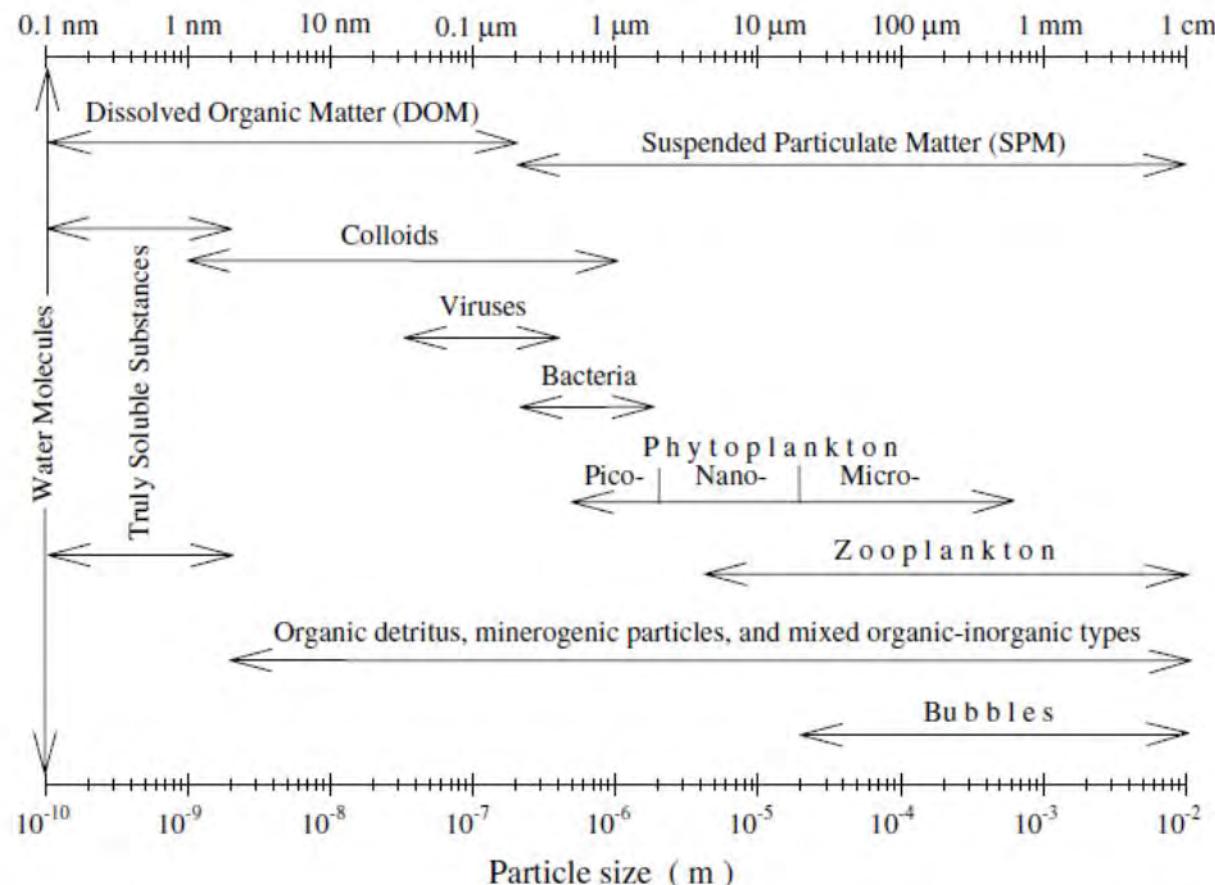


Fig. 1. Schematic diagram showing various seawater constituents in the broad size range from molecular size of the order of 10^{-10} m to large particles and bubbles of the order of 10^{-3} – 10^{-2} m in size. The arrow ends generally indicate approximate rather than sharp boundaries for different constituent categories.

1. absorption properties

$$a = a_w + \sum a_{xi}$$

Very detailed:

$$\begin{aligned} a(\lambda) &= a_w(\lambda) + \sum_{i=1}^{18} a_{\text{pla},i}(\lambda) + a_{\text{det}}(\lambda) + a_{\text{min}}(\lambda) + a_{\text{CDOM}}(\lambda) \\ &= a_w(\lambda) + \sum_{i=1}^{18} N_{\text{pla},i} \sigma_{a,\text{pla},i}(\lambda) + N_{\text{det}} \sigma_{a,\text{det}}(\lambda) \\ &\quad + N_{\text{min}} \sigma_{a,\text{min}}(\lambda) + a_{\text{CDOM}}(\lambda), \end{aligned} \tag{1}$$

(Stramski et al 2001)

Practical (and common) division:

$$a = a_w + a_p + a_g$$

$$a = a_w + a_{ph} + a_d + a_g$$

Pure water (seawater): a_w

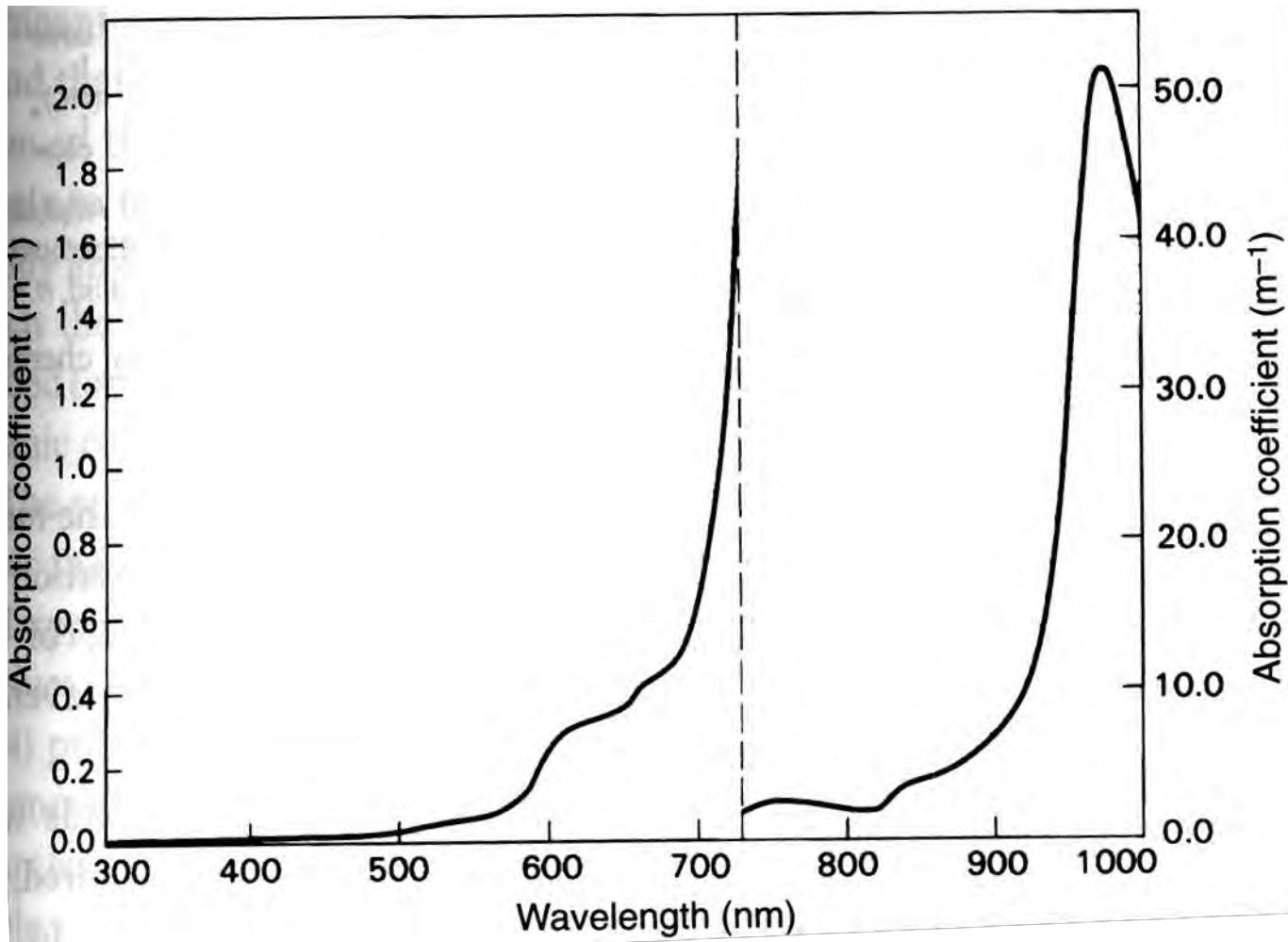
Particulate: $a_p = a_{ph} + a_d$

Pigments of living phytoplankton: a_{ph}

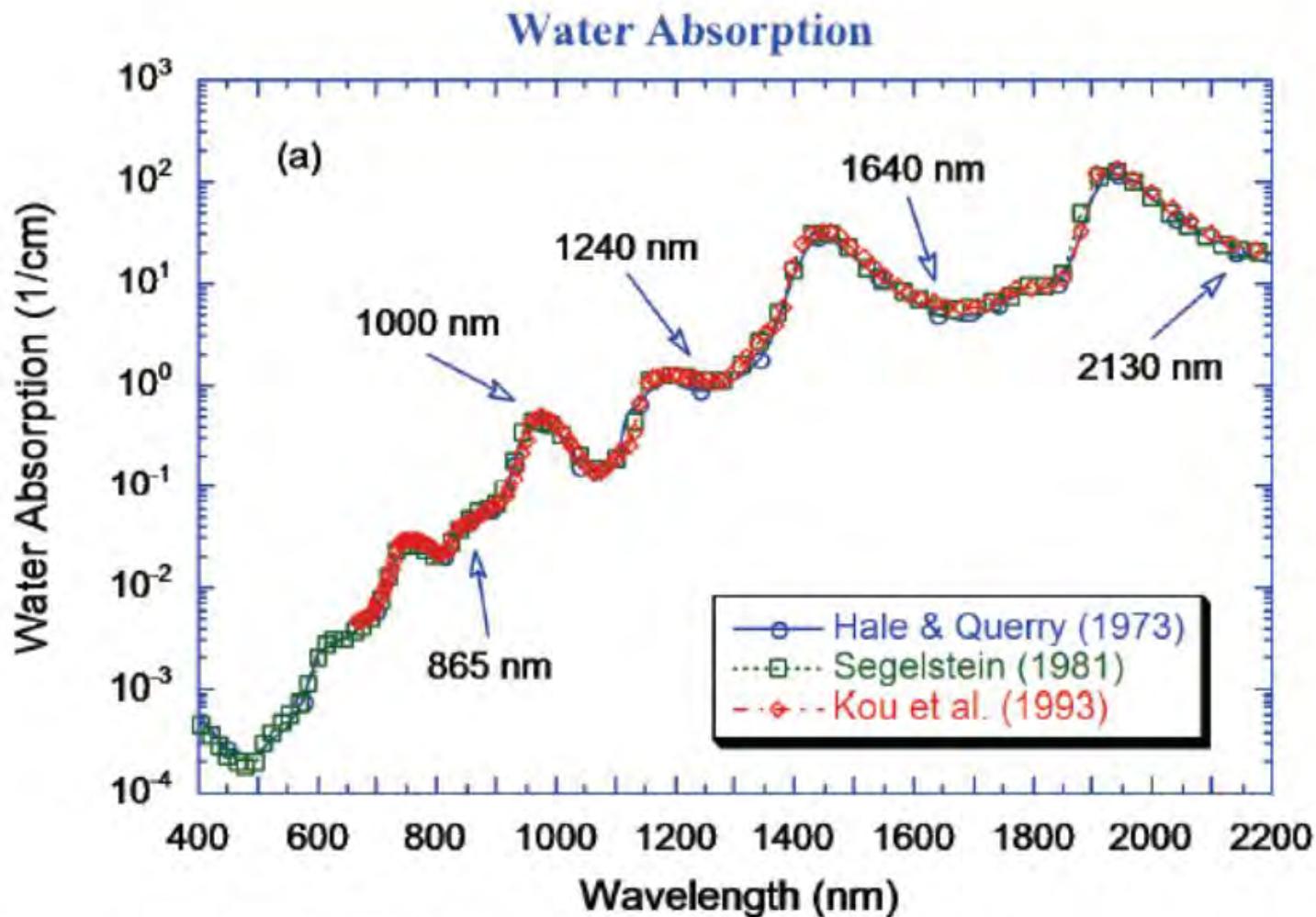
Detritus: a_d

Gelbstoff (yellow substance; colored dissolved organic matter): a_g

a_w spectrum



a_w



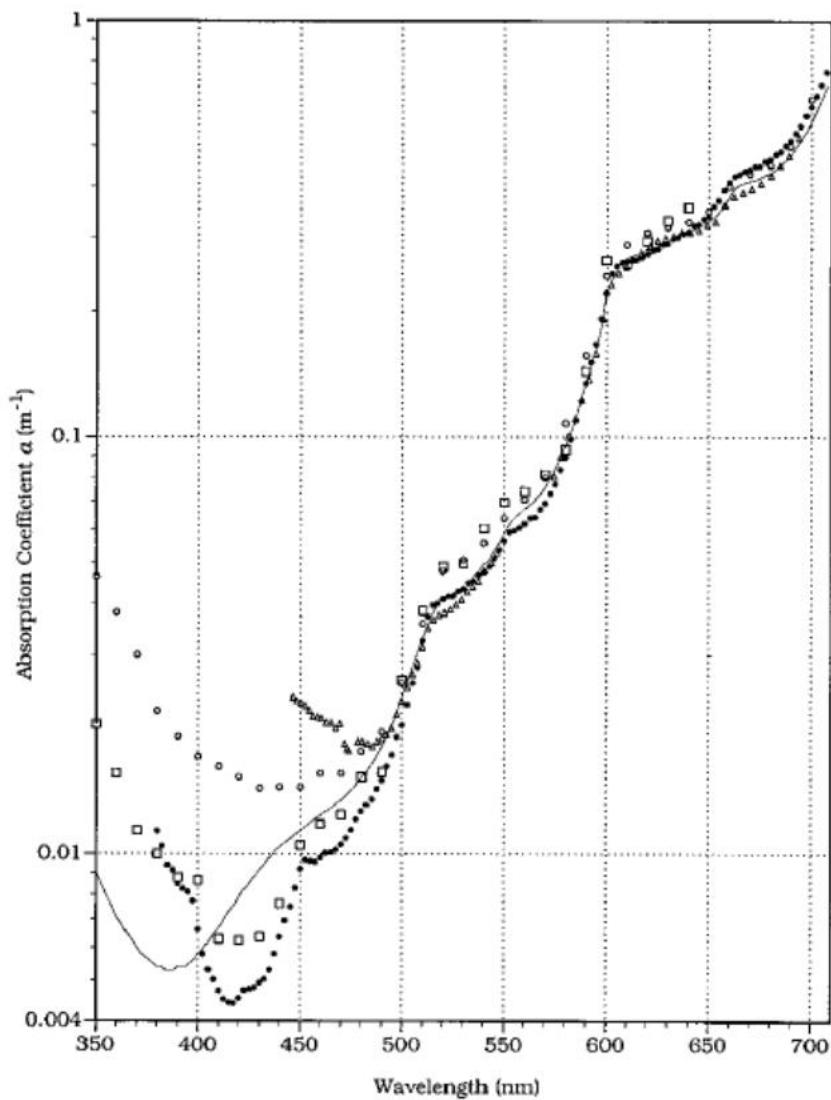
*a*_w

Table 3.1. Absorption coefficients for pure water: 280–320 nm, Quickenden & Irvin (1980); 366 nm, Boivin et al. (1986); 380–700 nm, Morel & Prieur (1977); 700–800 nm, Smith & Baker (1981)

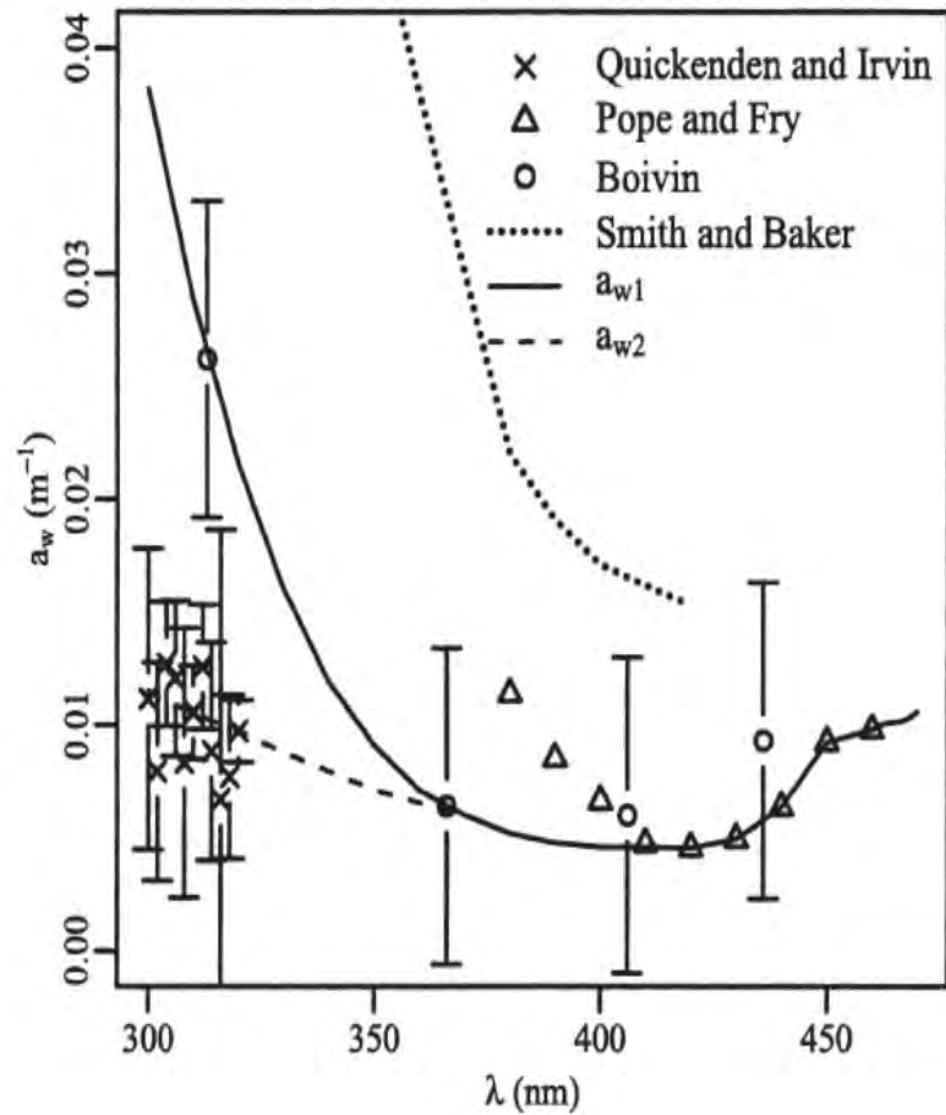
λ (nm)	a (m ⁻¹)	λ (nm)	a (m ⁻¹)
280	0.0239 ^{ab}	560	0.071
290	0.0140 ^{ab}	570	0.080
300	0.0085 ^{ab}	580	0.108
310	0.0082 ^{ab}	590	0.157
320	0.0077 ^{ab}	600	0.245
366	0.0055 ^a	610	0.290
380	0.023	620	0.310
390	0.020	630	0.320
400	0.018	640	0.330
410	0.017	650	0.350
420	0.016	660	0.410
430	0.015	670	0.430
440	0.015	680	0.450
450	0.015	690	0.500
460	0.016	700	0.650
470	0.016	710	0.839
480	0.018	720	1.169
490	0.020	730	1.799
500	0.026	740	2.38
510	0.036	750	2.47
520	0.048	760	2.55
530	0.051	770	2.51
540	0.056	780	2.36
550	0.064	790	2.16
		800	2.07

(Mobley 1994)

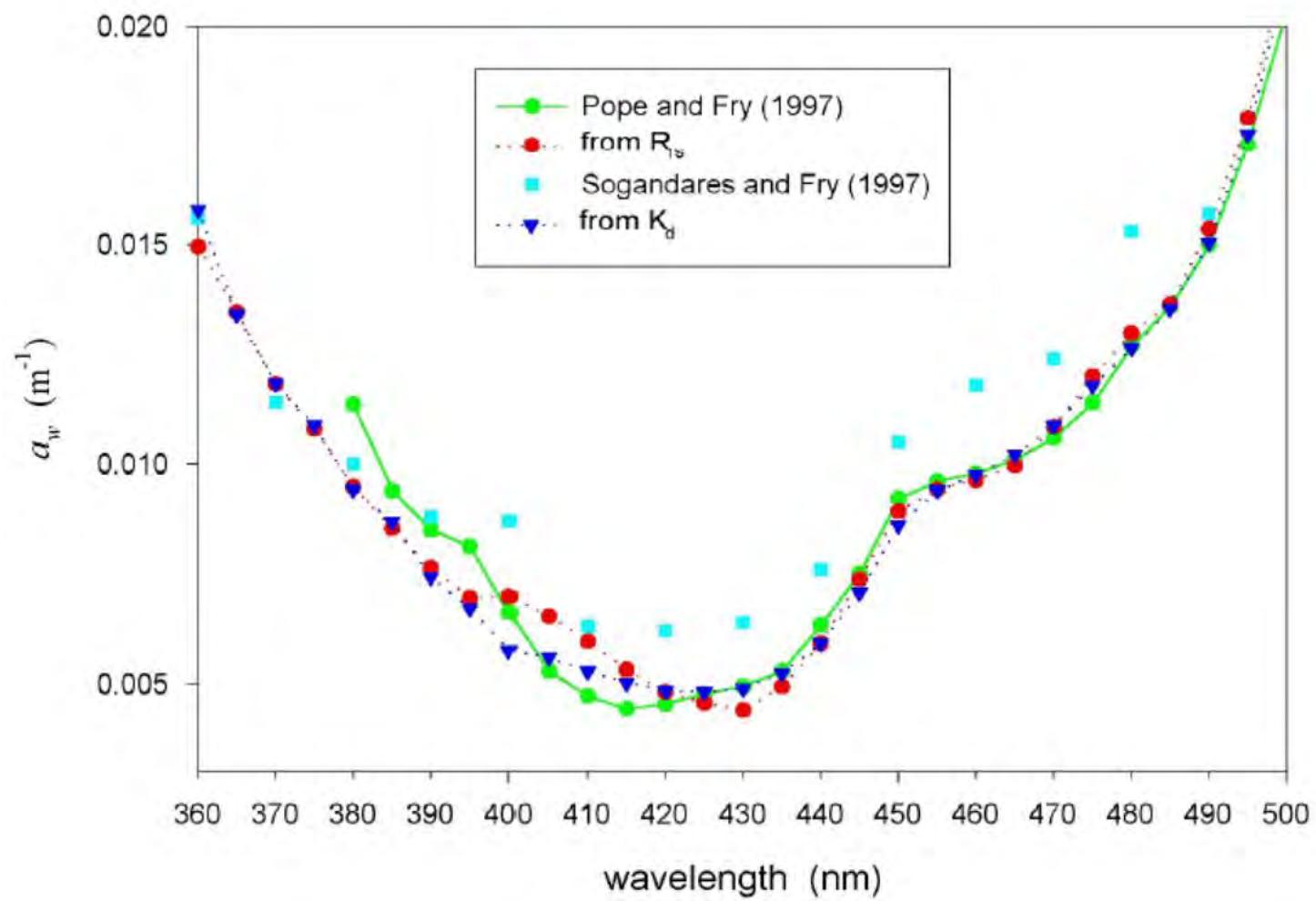
Uncertainties of a_w :



(Pope and Fry, 1997)



(Morel et al 2007)



(Lee et al 2000)

a_w is temperature and salinity dependent

$$a_w(\lambda, T, S) = a_w(\lambda, T_r, 0) + \Psi_T(T - T_r) + \Psi_S S, \quad (1)$$

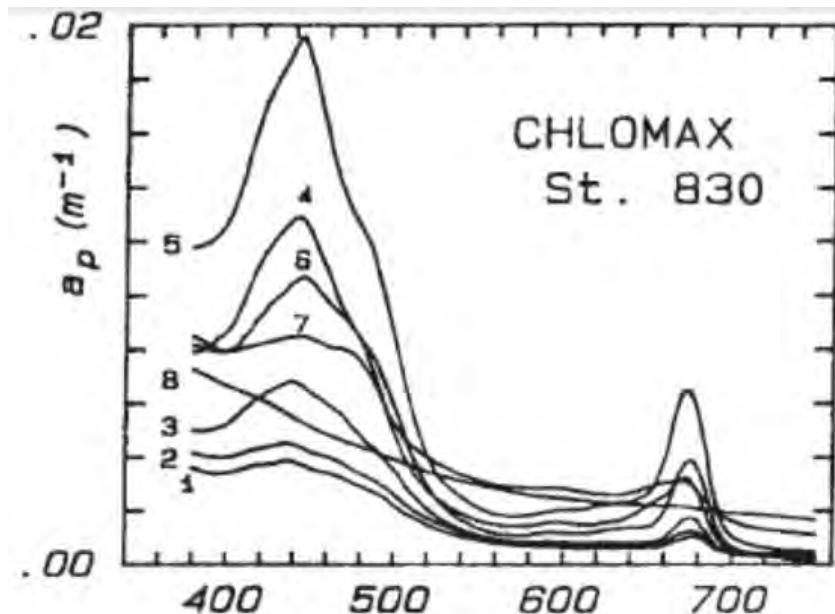
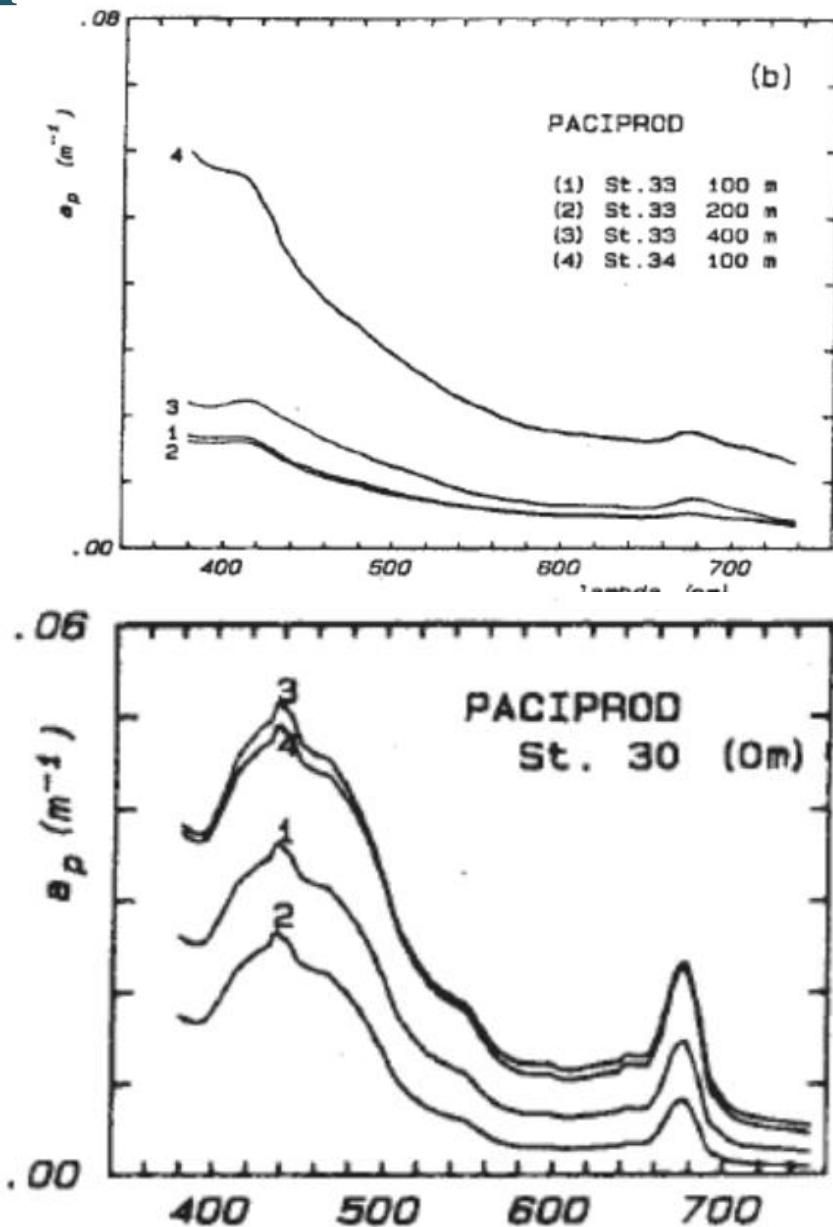
Table 2. Linear Slopes of the Temperature Dependence of the Absorption Coefficient Measured in the Laboratory^a

Wavelength	Ψ_T , Pure Water	Standard Deviation, Pure Water	Ψ_T , Saltwater	Standard Deviation, Saltwater
412	0.0001	0.0003	0.0003	0.0003
440	0.0000	0.0002	0.0002	0.0002
488	0.0000	0.0002	0.0001	0.0002
510	0.0002	0.0001	0.0003	0.0001
520	0.0001	0.0002	0.0002	0.0002
532	0.0001	0.0002	0.0001	0.0002
555	0.0001	0.0001	0.0002	0.0002
560	0.0000	0.0002	0.0000	0.0002
650	-0.0001	0.0001	-0.0001	0.0001
676	-0.0001	0.0001	-0.0001	0.0002
715	0.0029	0.0001	0.0027	0.0001
750	0.0107	0.0003	0.0106	0.0005
850	-0.0065	0.0001	-0.0068	0.0001
900	-0.0088	0.0001	-0.0090	0.0002
975	0.2272	0.0028	0.2273	0.0009

^aFor pure water the results of five tests are combined. The results of two tests were combined for the saltwater results. The absorption and attenuation meter results have been pooled together as well as pooling the common wavelengths between instruments. The standard deviations of the pooled values are provided.

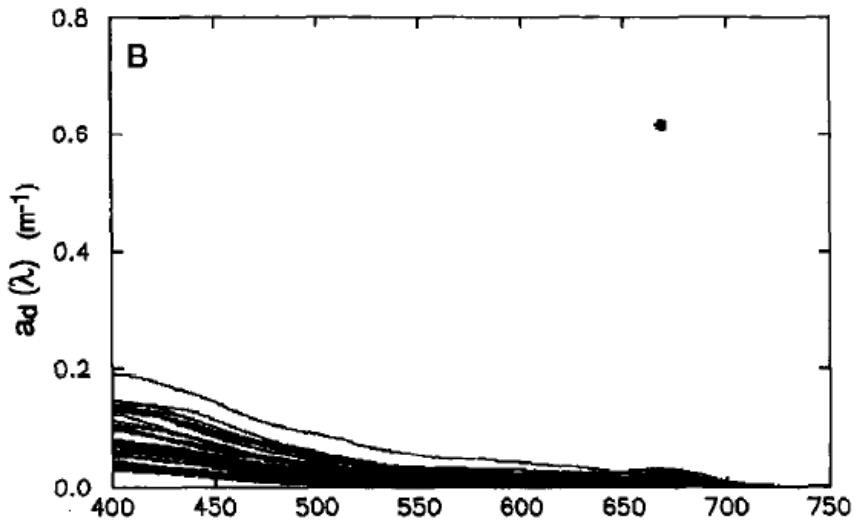
(Pegau et al 1997)

a_p spectrum



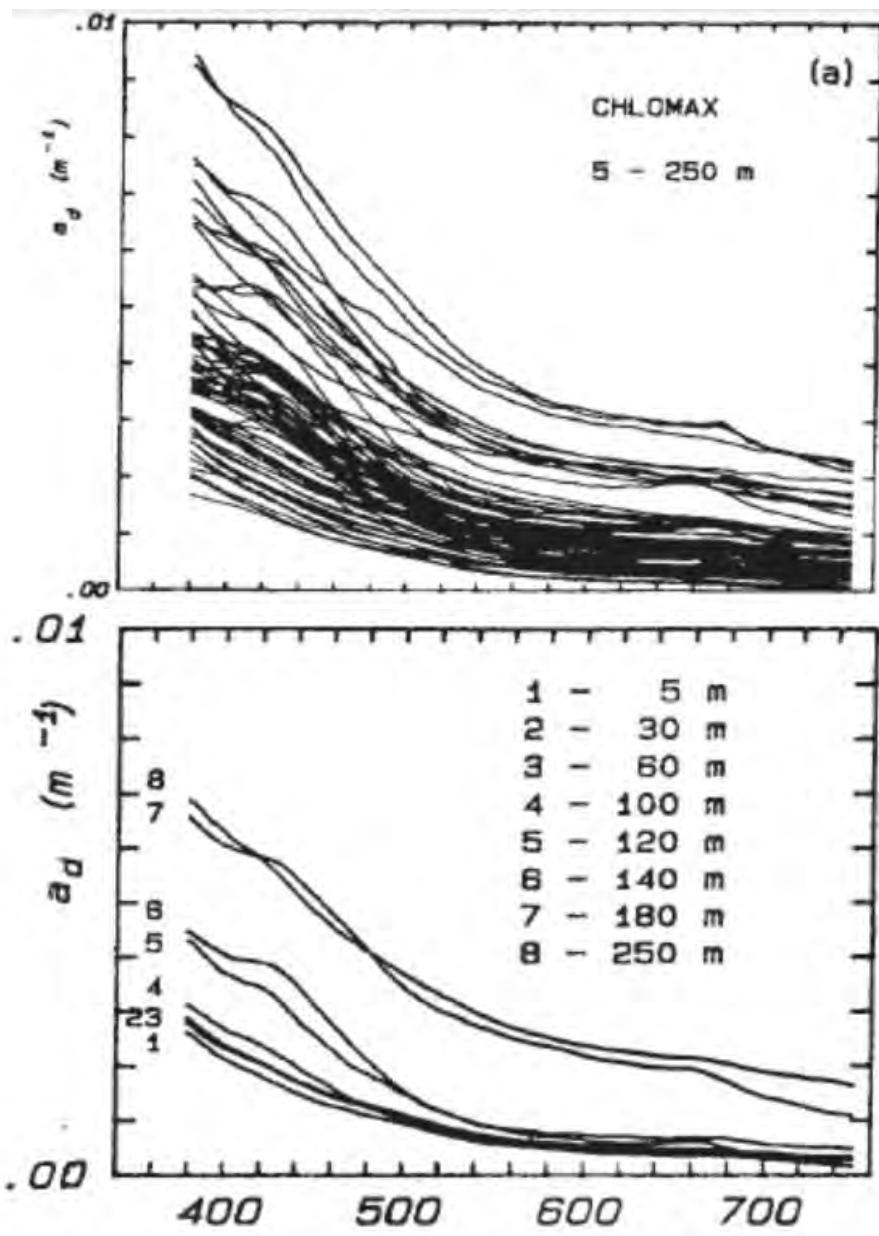
Bricaud and Stramski (1990)

a_d spectrum



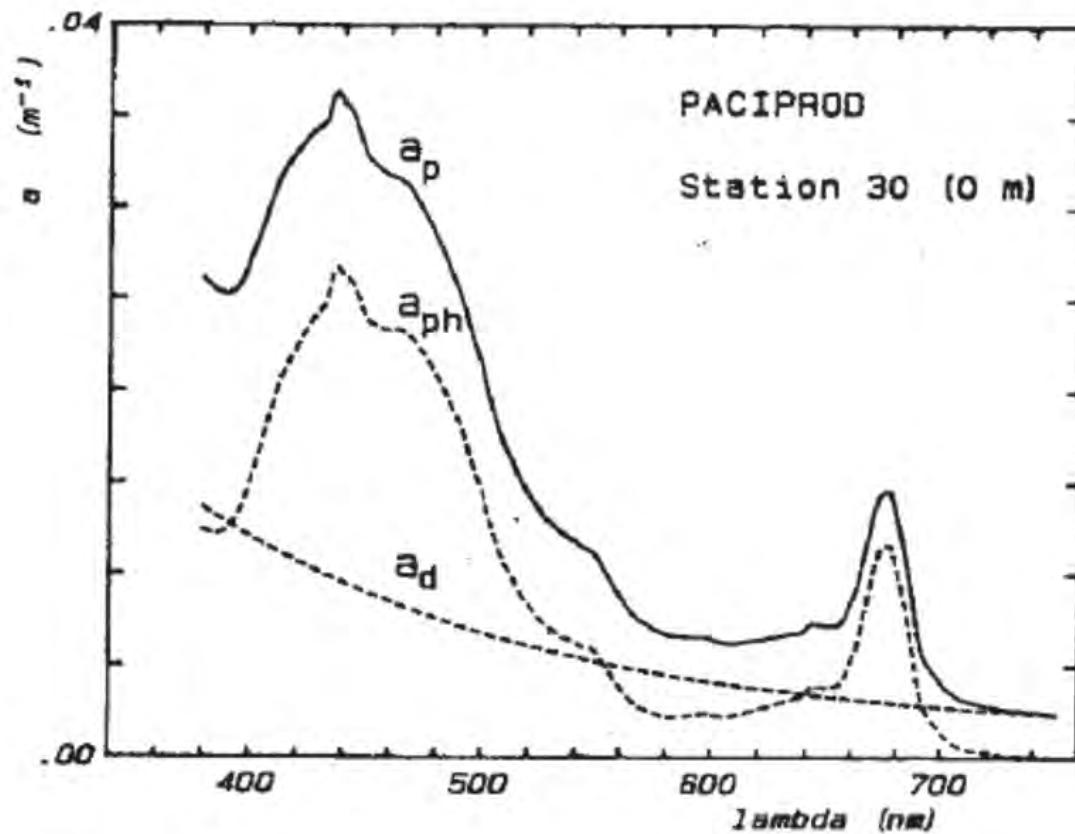
$$a_d = a_d(\lambda_0) e^{-S_d(\lambda - \lambda_0)}$$

S_d : ~0.005 – 0.015 nm^{-1}

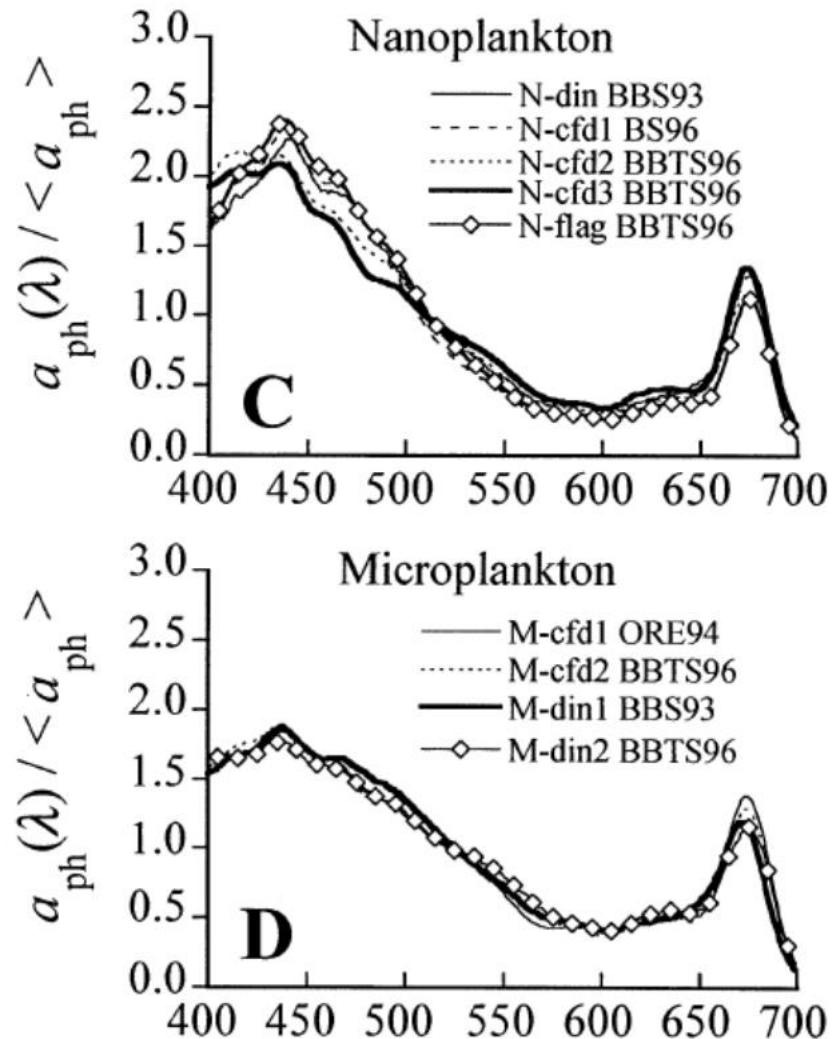
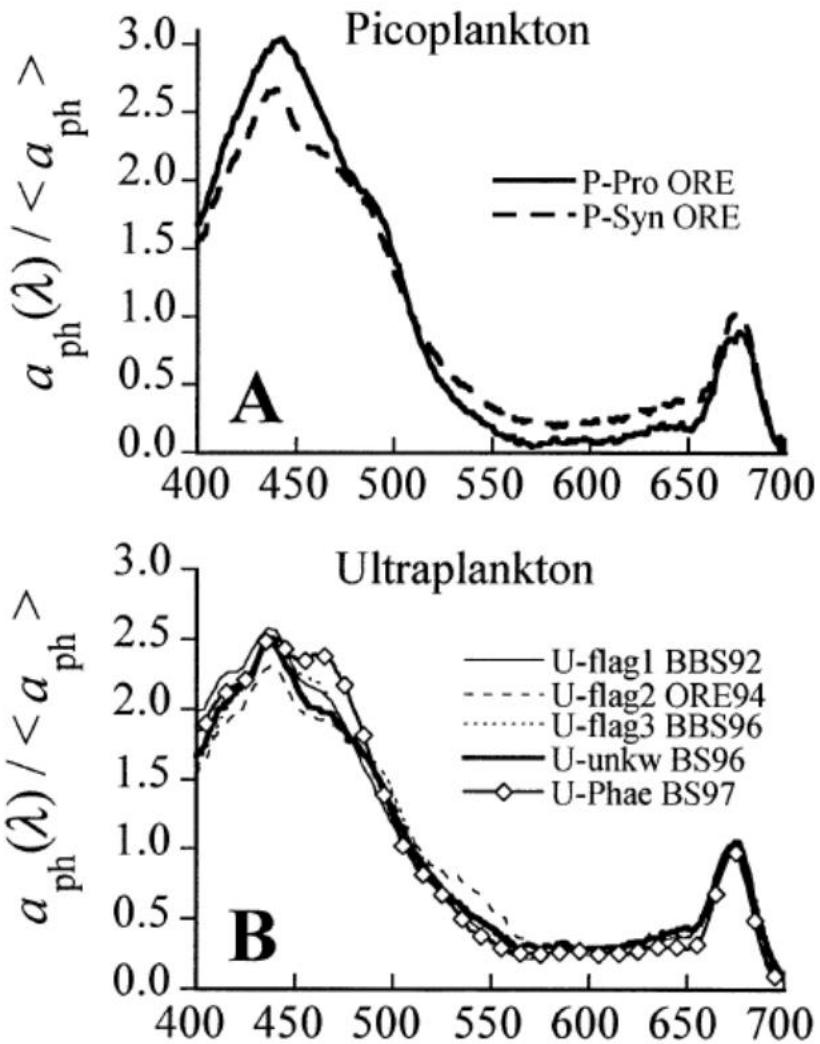


$$a_{ph} = a_p - a_d$$

spectrum

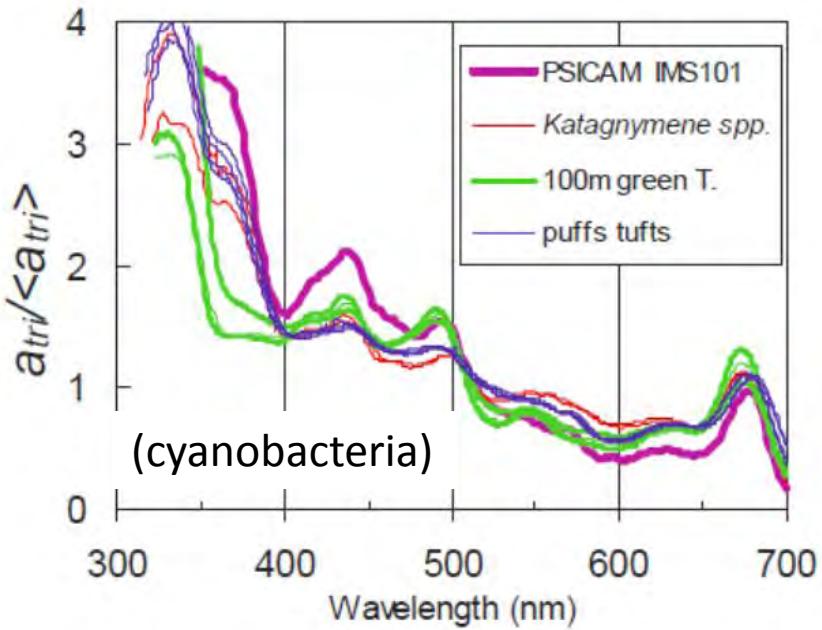
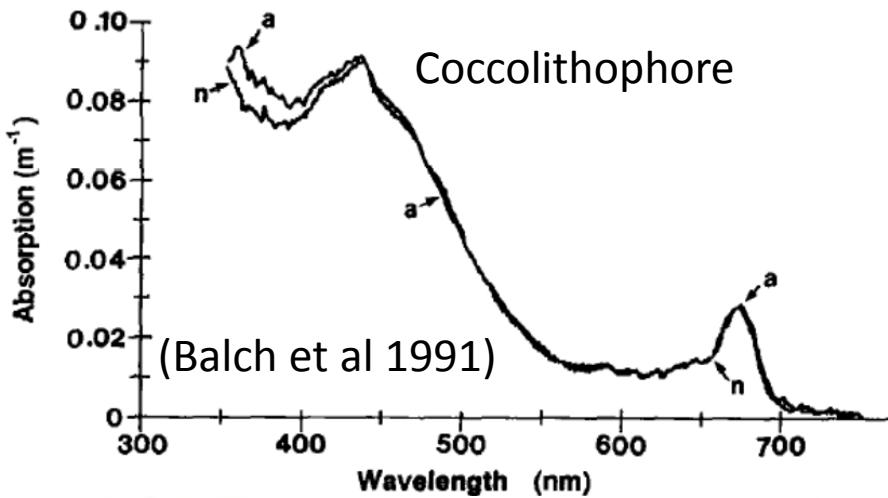


Separated by size

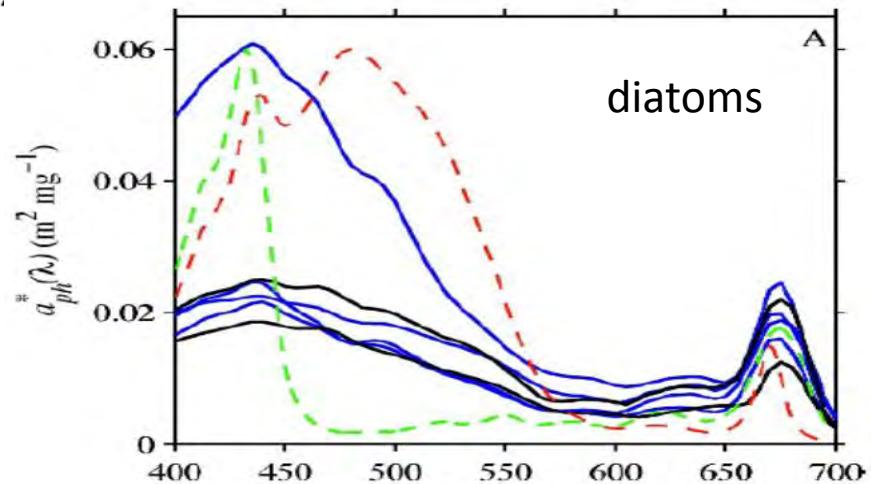


(Ciotti et al 2002)

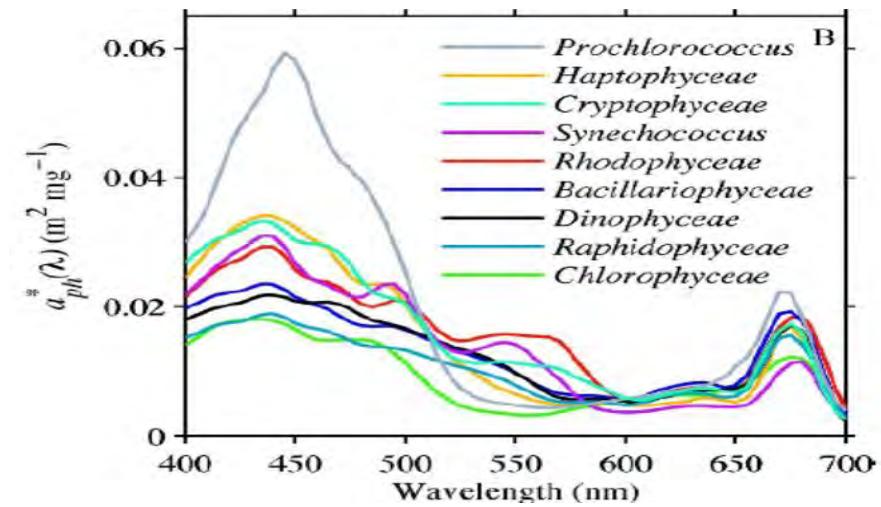
By species or groups



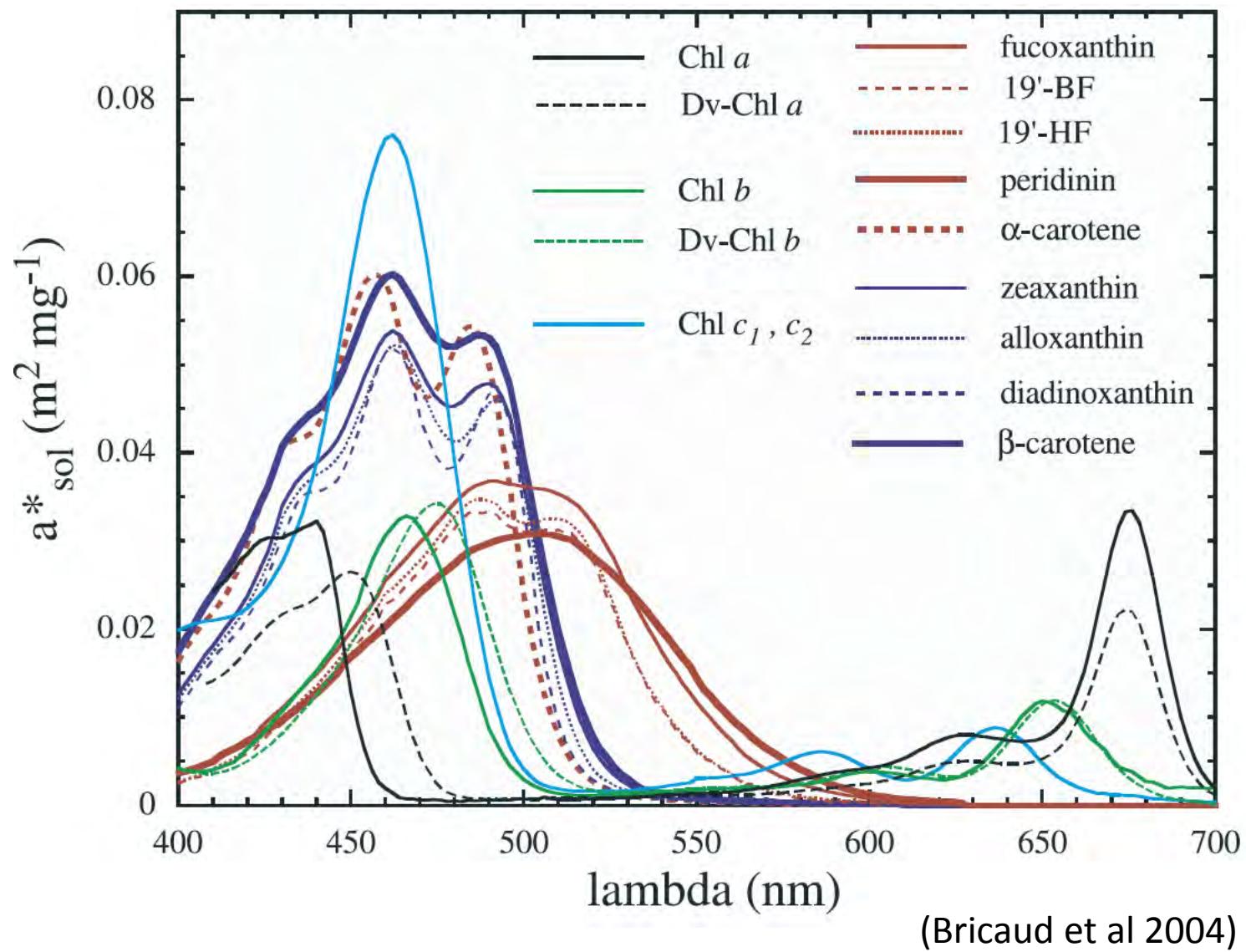
(Dupouy et al 2008)



(Dierssen et al 2006)



Separated by pigments



Modeling a_{ph} spectrum

Example of one parameter hyperspectral $a_{ph}(\lambda)$ model:

Bricaud et al (1995):

$$a_{ph}(\lambda) = A_{ph}(\lambda) Chl^{1-B_{ph}(\lambda)}$$

Lee (1994); Lee et al (1998):

$$a_{ph}(\lambda) = (a_0(\lambda) + a_1(\lambda) \ln(P)) P$$

$$P = a_{ph}(440)$$

Table 2. Parameters for the Empirical $a_{ph}(\lambda)$ Simulation by Eq. (12)^a

Table 2. Spectral Values of the Constants Obtained When Fitting the Variations of $a_{ph}^*(\lambda)$ Versus the (chl a + di a) Concentration (Chl) to Power Laws of the Form $a_{ph}^*(\lambda) = A(\lambda) (\text{Chl})^{-B(\lambda)}$ and Determination Coefficients on the Log-Transformed Data r^2

λ , nm	A	B	r^2	λ , nm	A	B	Wavelength	$a_0(\lambda)$	$a_1(\lambda)$
400	0.0263	0.282	0.702	402	0.0271	0.281	390	0.5813	0.0235
404	0.0280	0.282	0.706	406	0.0290	0.281	400	0.6843	0.0205
408	0.0301	0.282	0.710	410	0.0313	0.283	410	0.7782	0.0129
412	0.0323	0.286	0.718	414	0.0333	0.291	420	0.8637	0.006
416	0.0342	0.293	0.725	418	0.0349	0.296	430	0.9603	0.002
420	0.0356	0.299	0.733	422	0.0359	0.306	440	1.0	0
424	0.0362	0.313	0.746	426	0.0369	0.316	450	0.9634	0.006
428	0.0376	0.317	0.749	430	0.0386	0.314	460	0.9311	0.0109
432	0.0391	0.318	0.750	434	0.0395	0.324	470	0.8697	0.0157
436	0.0399	0.328	0.757	438	0.0401	0.332	480	0.789	0.0152
440	0.0403	0.332	0.762	442	0.0398	0.339	490	0.7558	0.0256
444	0.0390	0.348	0.774	446	0.0383	0.355	500	0.7333	0.0559
448	0.0375	0.360	0.783	450	0.0371	0.359	510	0.6911	0.0865
452	0.0365	0.362	0.783	454	0.0358	0.366	520	0.6327	0.0981
456	0.0354	0.367	0.789	458	0.0351	0.368	530	0.5681	0.0969
460	0.0350	0.365	0.789	462	0.0347	0.366	540	0.5046	0.09
464	0.0343	0.368	0.792	466	0.0339	0.369	550	0.4262	0.0781
468	0.0335	0.369	0.793	470	0.0332	0.368	560	0.3433	0.0659
472	0.0325	0.371	0.792	474	0.0318	0.375	570	0.295	0.06
476	0.0312	0.378	0.793	478	0.0306	0.379	580	0.2784	0.0581
480	0.0301	0.377	0.791	482	0.0296	0.377	590	0.2575	0.0571
484	0.0290	0.376	0.788	486	0.0285	0.373	600	0.2375	0.0551
488	0.0279	0.369	0.783	490	0.0274	0.361	610	0.2175	0.0531
492	0.0267	0.356	0.774	494	0.0258	0.349	620	0.1975	0.0511
496	0.0249	0.341	0.763	498	0.0240	0.332	630	0.1775	0.0491
500	0.0230	0.321	0.747	502	0.0220	0.311	640	0.1575	0.0471
504	0.0209	0.300	0.722	506	0.0199	0.288	650	0.1375	0.0451
508	0.0189	0.275	0.686	510	0.0180	0.260	660	0.1175	0.0431
512	0.0171	0.249	0.641	514	0.0163	0.237	670	0.0975	0.0411
516	0.0156	0.224	0.578	518	0.0149	0.211	680	0.0775	0.0391
520	0.0143	0.196	0.498	522	0.0137	0.184	690	0.0575	0.0371
524	0.0131	0.173	0.417	526	0.0126	0.162	700	0.0375	0.0351
528	0.0121	0.151	0.332	530	0.0117	0.139	710	0.0175	0.0331
532	0.0113	0.129	0.248	534	0.0108	0.119	720	0.0075	0.0311
536	0.0104	0.109	0.176	538	0.0100	0.100	730	0.0075	0.0301
540	0.0097	0.090	0.116	542	0.0093	0.081	740	0.0075	0.0291

Example of two parameter model:

(Ciotti et al 2002)

$$a_{\phi}(\lambda) = a_{\phi}(505) \cdot [S_f \cdot \bar{a}_{<pico>}(\lambda)] + [(1 - S_f) \cdot \bar{a}_{<micro>}(\lambda)]$$

Table 3. Basis vectors representing the normalized absorption for the smallest ($\bar{a}_{\text{(pico)}}(\lambda)$, *Prochlorococcus*) and biggest ($\bar{a}_{\text{(micro)}}(\lambda)$, average microplankton) cell sizes in our data set. Wavelength (λ) in nm. Basis vectors for $a_{\text{ph}}^*(\lambda)$ can be constructed by setting $\bar{a}_{\text{(pico)}}(676)$ to 0.023 m² mg⁻¹, $\bar{a}_{\text{(micro)}}(674)$ to 0.0086 m² mg⁻¹, and scaling for the other wavelengths accordingly.

λ	Pico	Micro												
400	1.682	1.574												
402	1.734	1.584	462	2.526	1.623	522	0.544	1.013	582	0.111	0.459	642	0.191	0.528
404	1.800	1.600	464	2.455	1.616	524	0.522	0.992	584	0.072	0.452	644	0.174	0.526
406	1.890	1.617	466	2.402	1.606	526	0.486	0.977	586	0.073	0.452	646	0.197	0.528
408	1.978	1.633	468	2.331	1.592	528	0.448	0.959	588	0.073	0.449	648	0.176	0.538
410	2.057	1.654	470	2.281	1.568	530	0.391	0.944	590	0.099	0.443	650	0.168	0.549
412	2.162	1.669	472	2.205	1.542	532	0.375	0.927	592	0.070	0.433	652	0.160	0.574
414	2.269	1.674	474	2.136	1.509	534	0.336	0.909	594	0.095	0.424	654	0.217	0.605
416	2.327	1.684	476	2.063	1.481	536	0.305	0.888	596	0.085	0.416	656	0.244	0.655
418	2.398	1.697	478	2.049	1.459	538	0.292	0.868	598	0.090	0.406	658	0.286	0.720
420	2.457	1.708	480	1.998	1.437	540	0.288	0.847	600	0.086	0.401	660	0.381	0.798
422	2.533	1.710	482	1.930	1.415	542	0.261	0.826	602	0.068	0.400	662	0.437	0.889
424	2.614	1.716	484	1.918	1.399	544	0.245	0.806	604	0.078	0.403	664	0.520	0.979
426	2.663	1.737	486	1.897	1.387	546	0.214	0.785	606	0.069	0.408	666	0.660	1.068
428	2.749	1.763	488	1.867	1.377	548	0.194	0.764	608	0.090	0.416	668	0.716	1.147
430	2.804	1.793	490	1.812	1.367	550	0.187	0.737	610	0.096	0.429	670	0.824	1.207
432	2.840	1.812	492	1.776	1.349	552	0.138	0.711	612	0.094	0.443	672	0.846	1.243
434	2.915	1.827	494	1.701	1.338	554	0.137	0.682	614	0.084	0.458	674	0.816	1.249
436	2.947	1.830	496	1.648	1.319	556	0.111	0.653	616	0.105	0.473	676	0.891	1.227
438	2.978	1.834	498	1.522	1.301	558	0.094	0.626	618	0.128	0.487	678	0.869	1.174
440	3.014	1.824	500	1.439	1.271	560	0.095	0.604	620	0.119	0.495	680	0.812	1.096
442	3.032	1.800	502	1.373	1.242	562	0.070	0.580	622	0.126	0.499	682	0.741	1.004
444	3.011	1.771	504	1.270	1.222	564	0.053	0.555	624	0.138	0.504	684	0.605	0.893

Multiple parameter model:

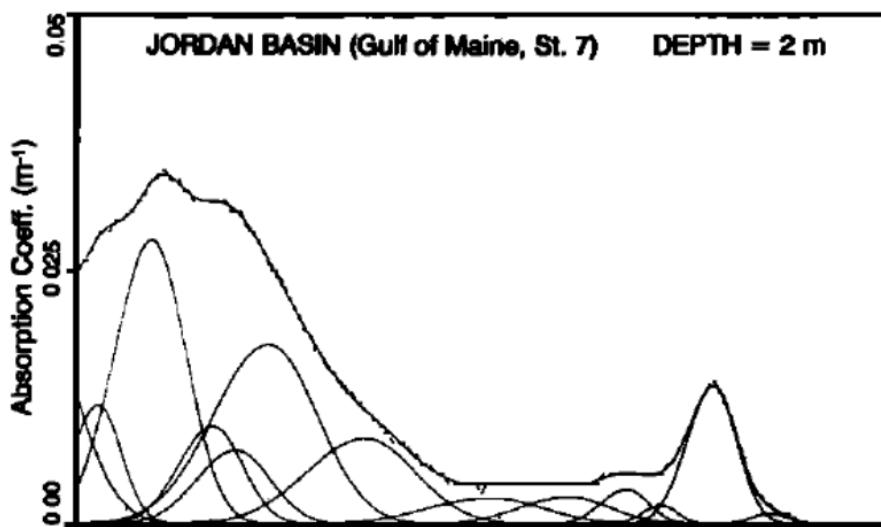
$$a_{ph}(\lambda) = \sum_{j=1}^l C_j a_j^*(\lambda_{mj}) \exp \left[\frac{(\lambda - \lambda_{mj})^2}{2\sigma_j^2} \right] \quad (7)$$

TABLE 2. Input Values and Mean Characteristics of Gaussian Bands Reflecting Absorption by Chlorophylls and Carotenoids

Characteristic	Gaussian Band Number and Associated Pigment Species												
	1 chl a	2 chl a	3 chl a	4 chl c	5 chl b	6 carot.	7 carot.	8 chl c	9 chl a	10 chl c	11 chl b	12 chl a	13 chl a
Input parameters													
Half width, nm	53.8	21.3	32.1	27.2	45.0	45.4	45.9	46.3	35.0	28.9	24.4	21.6	33.5
Center, nm	384	413	435	461	464	490	532	583	623	644	655	676	700
Output parameters													
Half width, nm	43.2	22.7	34.5	29.3	36.0	46.8	49.6	46.8	38.0	24.9	25.4	24.7	29.0
Center, nm	381.5	410.8	433.5	459.2	466.6	487.8	532.0	585.6	620.6	640.7	652.9	675.6	699.8
Specific absorption coefficient, m^2 (mg pigment) $^{-1}$	0.042	0.019	0.047	0.110	0.115	0.035	0.019	0.044	0.005	0.044	0.029	0.021	0.002

Specific absorption coefficients of each pigment are ratios of Gaussian band absorptions (in reciprocal meters) to high-performance liquid chromatography concentrations (in milligrams per cubic meter) of that pigment. Chl, chlorophyll; Carot., carotenoid.

(Hoepffner and Sathyendranath, 1993)



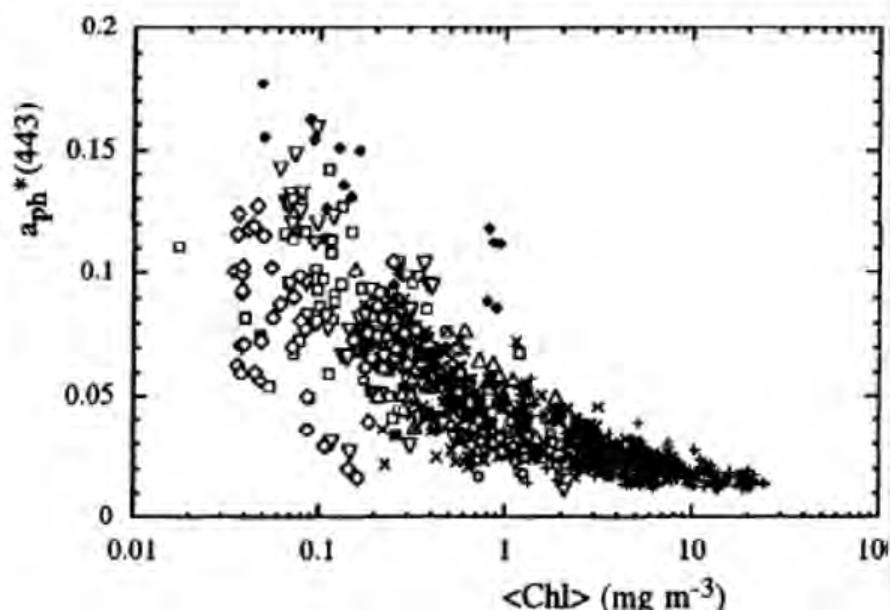
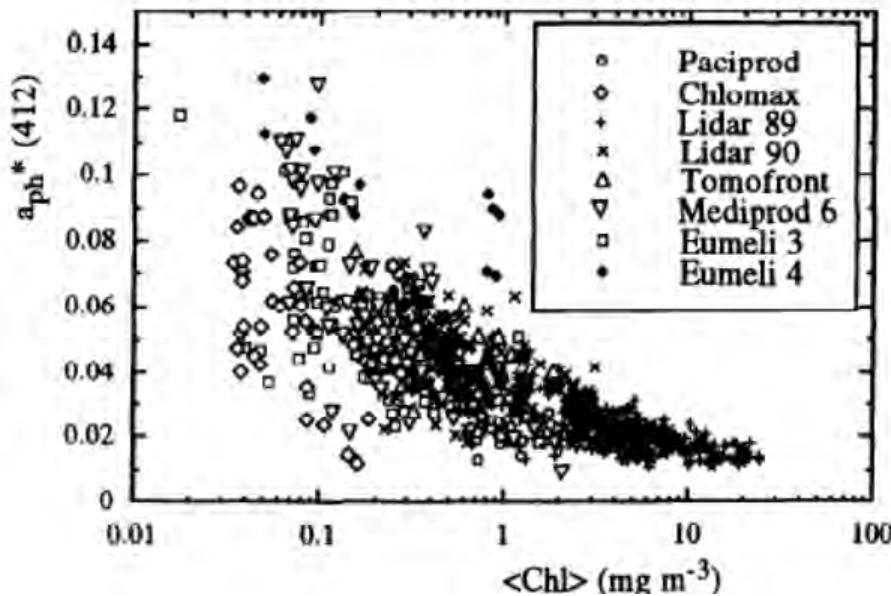
Package effect

Specific absorption/scattering coefficient =

Concentration normalized absorption/scattering coefficient

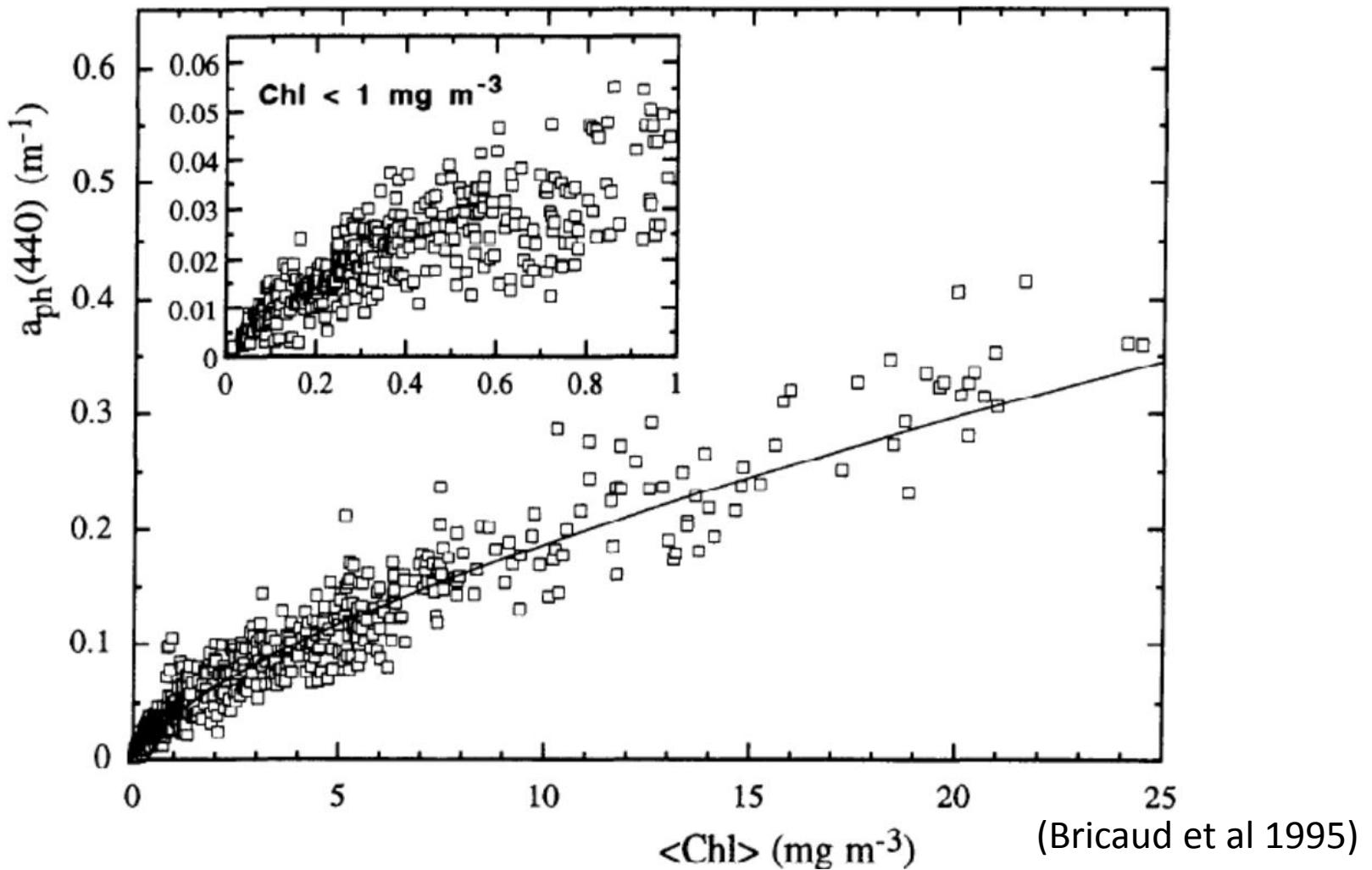
$$a_{ph}^* = \frac{a_{ph}}{Chl}$$

$$b_{ph}^* = \frac{b_{ph}}{Chl}$$



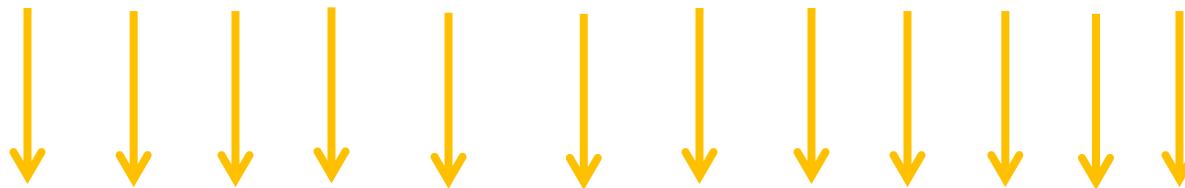
$Chl \uparrow \rightarrow$ specific optical property \downarrow

(Bricaud et al 1995)



Increase of absorption is NOT linearly proportionally to Chl concentration!

Simplified case:



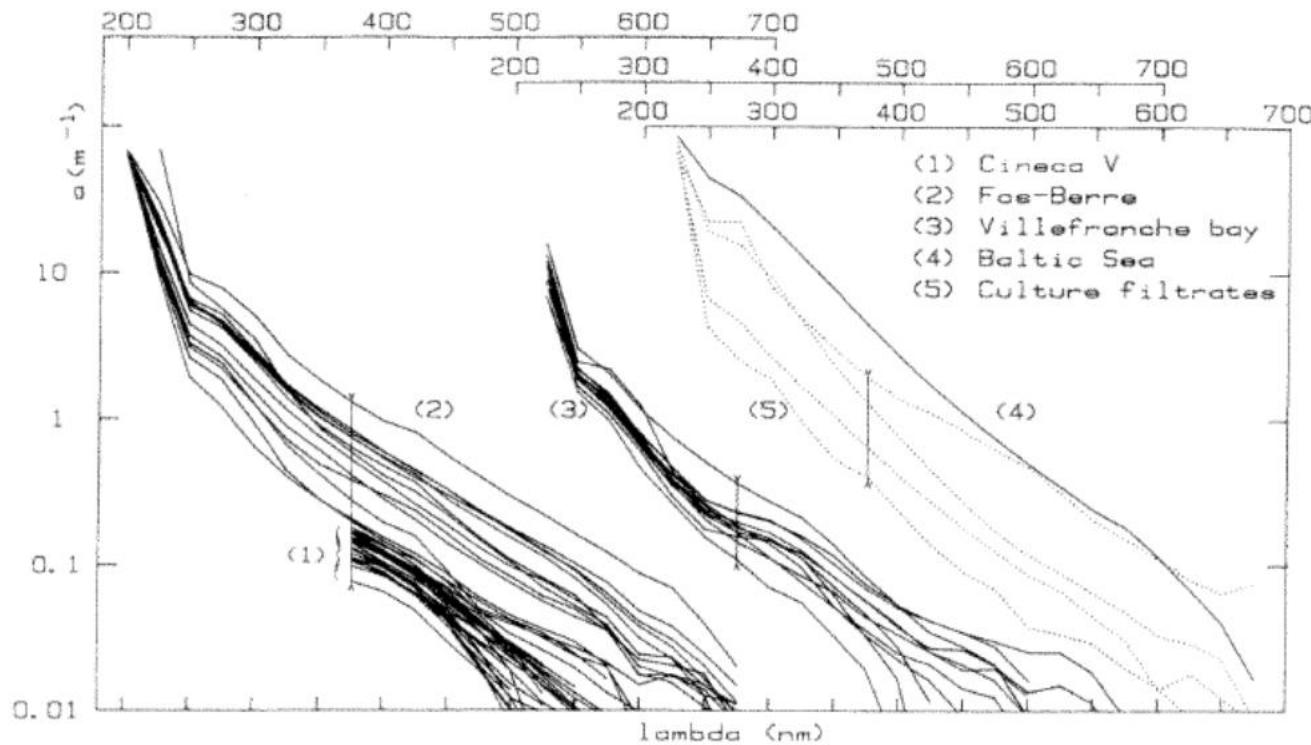
$$a \propto \rho S$$

$$W \propto \rho V$$

$$a_{ph}^* \propto \frac{a}{W} = \frac{S}{V} \propto \frac{1}{d}$$

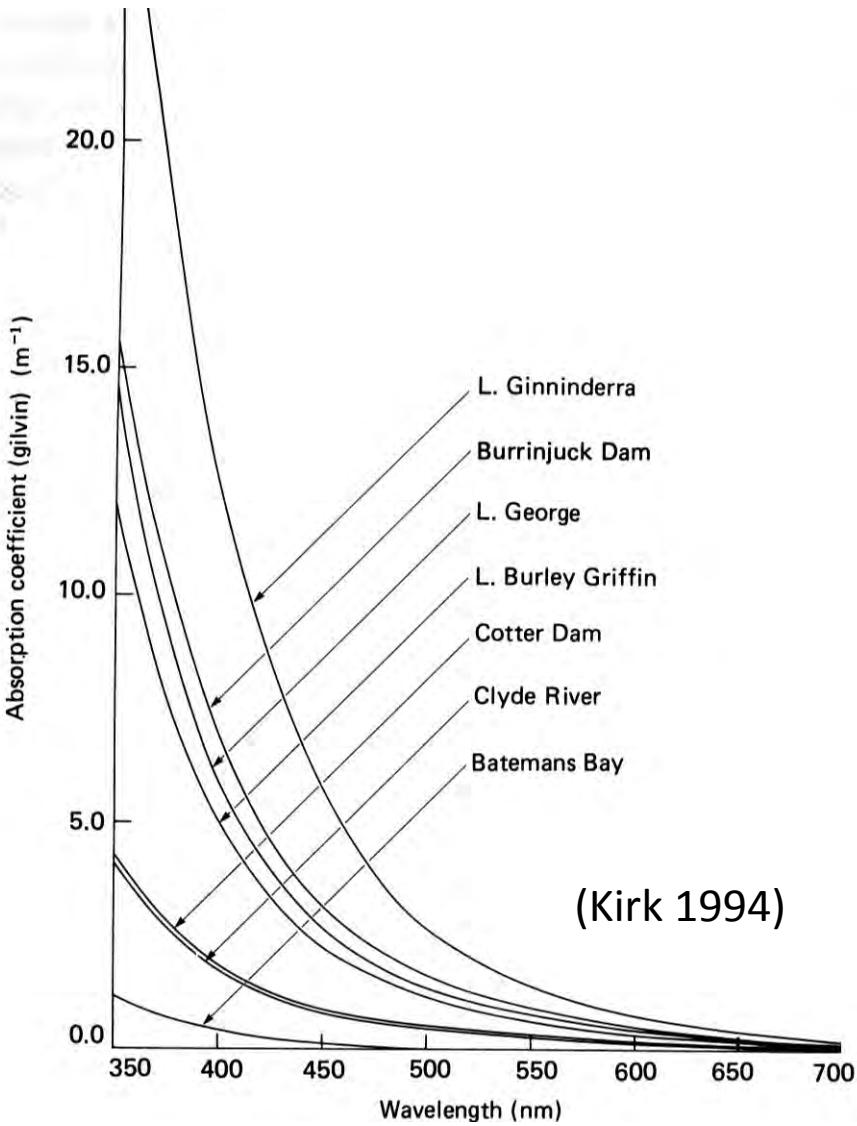
a_g spectrum

Absorption spectra of yellow substance (gelbstoff)



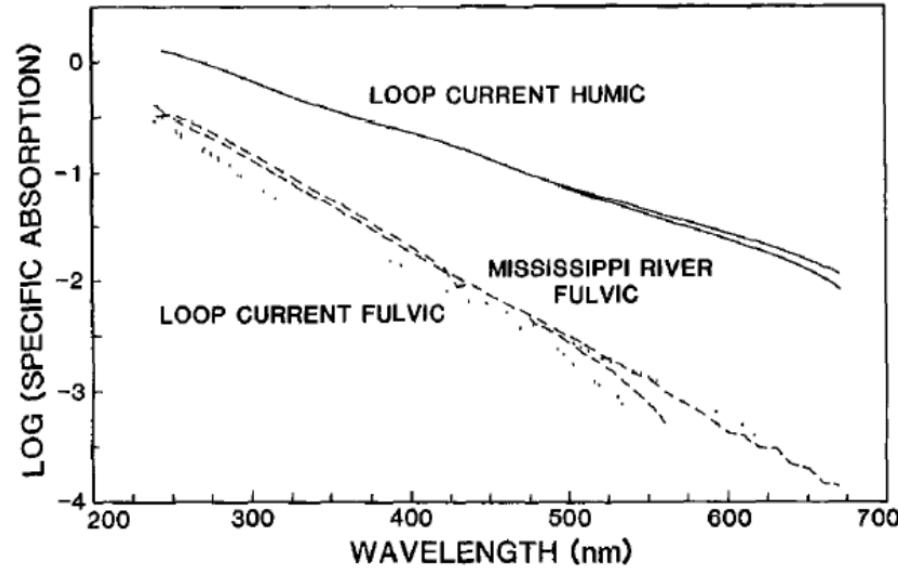
(Bricaud et al 1981)

a_g spectrum



$$a_g = a_g(\lambda_0) e^{-S_g(\lambda - \lambda_0)}$$

$$S_g: \sim 0.01 - 0.03 \text{ nm}^{-1}$$



(Carder et al 1989)

Table 1

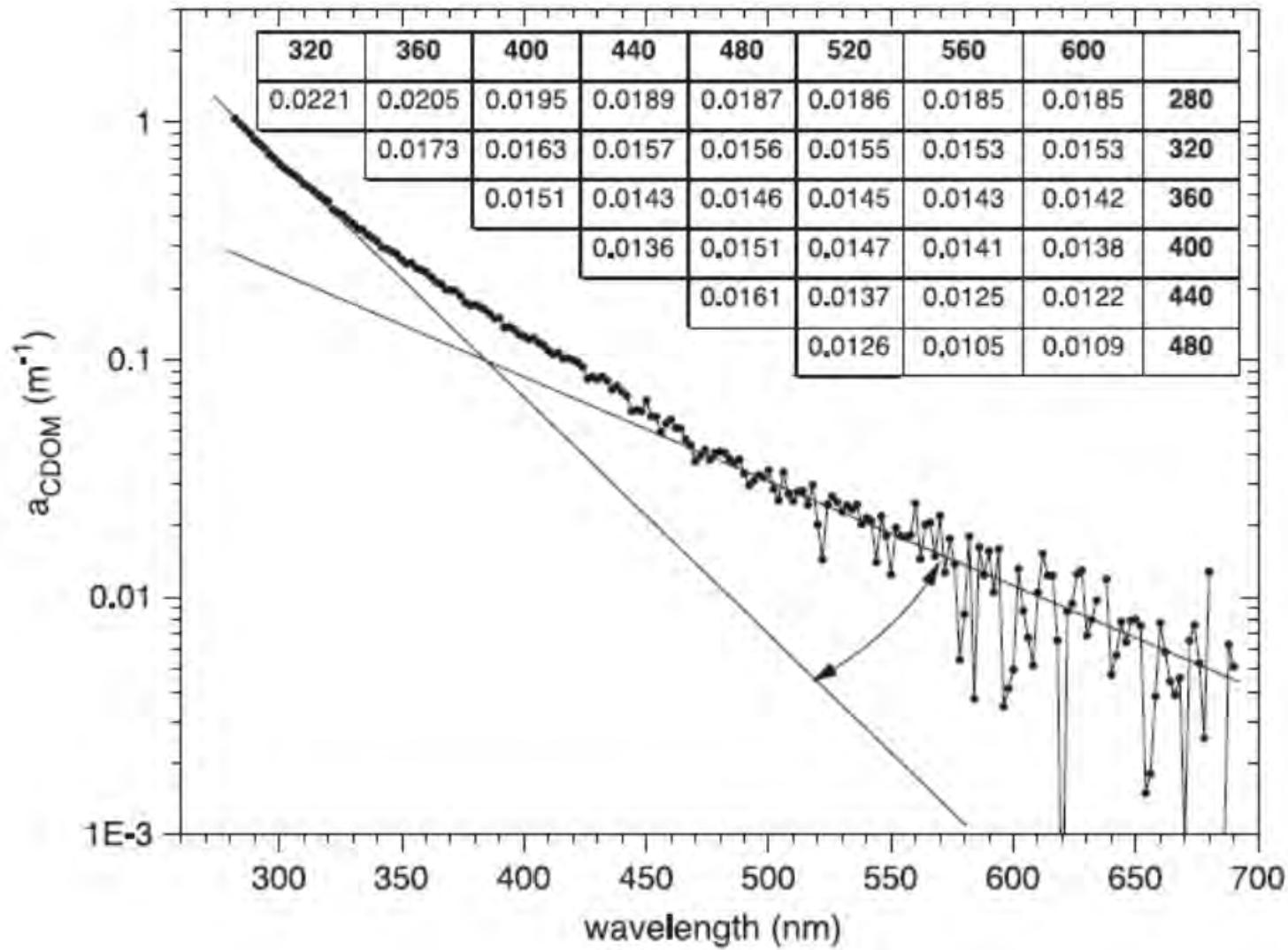
Spectral slope values for marine samples reported in the literature with spectral range, CDOM absorption at 412 nm, and reported precision (ordered according to starting wavelength range)

Reference	Location	n^a	Slope (nm^{-1}) ^b	Wavelength range	$a_g(412) (\text{m}^{-1})^c$	Prec (m^{-1})
Hojerslev and Aas (2001)	Kattegat–Skagerrak	1305	0.0234 ± 0.0036 , [0.0075–0.0420]	[250–450]	1.28 ± 0.70	0.002
Brown (1977)	North Sea	37	[0.0187–0.0306]	280,310	[0.022–0.327]	?
	Baltic proper	157	[0.0247–0.0305]	280,310	[0.136–0.284]	?
	Baltic riverine	1	0.0173	280,310	2.49	?
Nelson et al. (1998)	Bermuda	?	0.0235	280–350	~ 0.1–0.4	0.03
Blough et al. (1993)	Gulf of Paria (samples <30 ppb)	47	0.0140 ± 0.0003	[290–600+] ^d	[1.25–4.59]	0.092
Green and Blough (1994)	S. Florida/Gulf of Mexico	31	0.021 ± 0.005	[290–(330–675)] ^d	[0.01–6.32]	0.092
	Amazon R. estuary	12	0.019 ± 0.005	[290–(370–590)] ^d	[0.03–1.33]	0.092
			[0.014–0.033]			
Vodacek et al. (1997)	coastal Mid-Atlantic Bight: non-Nov.	~ 40	0.018 average	[290–(440–550)] ^d	[0.14–0.71]	0.092
	Nov.	~ 25	0.014 average	[290–(400–550)] ^d	[0.14–0.63]	0.092
	offshore Mid-Atlantic Bight	~ 150	[0.010–0.034]	[290–(340–440)] ^d	[0.009–0.14]	0.092
Del Castillo et al. (1999)	Gulf of Paria and surrounding waters	8	0.018 ± 0.002	[290–var] ^e	[0.09–1.34]	0.046
Zepp and Schlotzhauer (1981)	Gulf of Mexico, St. Marks, FL	1	0.0151 ± 0.0051	[300–500]	?	?
	'Marine aquatic humus'	3	0.0147	[300–500]	?	?
Davies-Colley (1992)	coastal N. Zealand	28	0.015 ± 0.002	[300–460]	[0.023–0.165]	0.017
	Doubtful Sound	6	0.014 ± 0.0004	[300–460]	[0.678–2.60]	0.017
Stedmon et al. (2000)	Danish fjords and nearby coastal waters	586	0.0194 ± 0.0032^f	[300–650]	[0.14–3.46]	?
Stedmon and Markager (2001)	Greenland Sea, Nov 98	20	0.02016 ± 0.00252	[300–650]	[0.04–0.08]	0.05
	Greenland Sea, Jun 99	107	0.01651 ± 0.00352	[300–650]	[0.04–0.70]	0.05
	Greenland Sea, Aug 99	67	0.01622 ± 0.00297	[300–650]	[0.04–0.31]	0.05
Bricsud et al. (1981)	Mauritanian upwelling	24	0.015 ± 0.0023	350:10:500 ^g	[0.03–0.12]	0.01
	Gulf of Guinea	35	0.014 ± 0.0041	350:25:500 ^g	[0.04–0.17]	0.01
	Villefrance Bay	11	0.014 ± 0.0024	350:25:500 ^g	[0.09–0.24]	0.01
	Var River	1	0.015	350:25:500 ^g	0.21	0.01
	Baltic Sea	1	0.018	350:25:500 ^g	2.18	0.01
	Gulf of Fos-sur-Mer	14	0.013 ± 0.0012	350:25:500 ^g	[0.12–0.82]	0.01
Kowalcuk et al. (in press)	Baltic, open sea	754	0.019 ± 0.004	[350–var]	[0.18–1.46]	0.023, 0.046
	Baltic, coastal	221	0.020 ± 0.003	[350–var]	[0.20–1.88]	0.023, 0.046
	Pomeranian Bight	312	0.020 ± 0.004	[350–var]	[0.21–1.71]	0.023, 0.046
	Bay of Gdańsk	1292	0.019 ± 0.004	[350–var]	[0.20–3.52]	0.023, 0.046
Schwarz et al. (2002)	Globally representative	877	0.01725 ± 0.0034	[350–var]	[~ 0.003–10.0]	0.046 ^h
Carder et al. (1989)	Gulf of Mexico	11	[0.0115–0.0172]	[370–440]	[0.002–0.074]	~ 0.01
Kopelevich and Burennkov (1977)	Deep Indian and Pacific	2	0.017	390:20:490 ^g	~ 0.06	?
Roessler et al. (1989)	San Juan Islands	21	0.017 ± 0.003	[400–750] ⁱ	0.32 average	?
Del Castillo et al. (1999)	Gulf of Paria and surrounding waters	8	0.015 ± 0.001	[400–500]	[0.09–1.34]	0.046
Maske et al. (1998)	Gulf of California	?	0.014	412,440,512	~ 0.095	0.002

(continued on next page)

(Twardowski et al 2004)

Slope changes with wavelength range



(Twardowski et al 2004)

Power-law model for a_g spectrum:

A generic, representative CDOM absorption model from this study which requires one absorption estimate at 412 nm as input is:

$$a_g(\lambda) = a_g(412) \left(\frac{\lambda}{412} \right)^{-6.92} \quad (4)$$

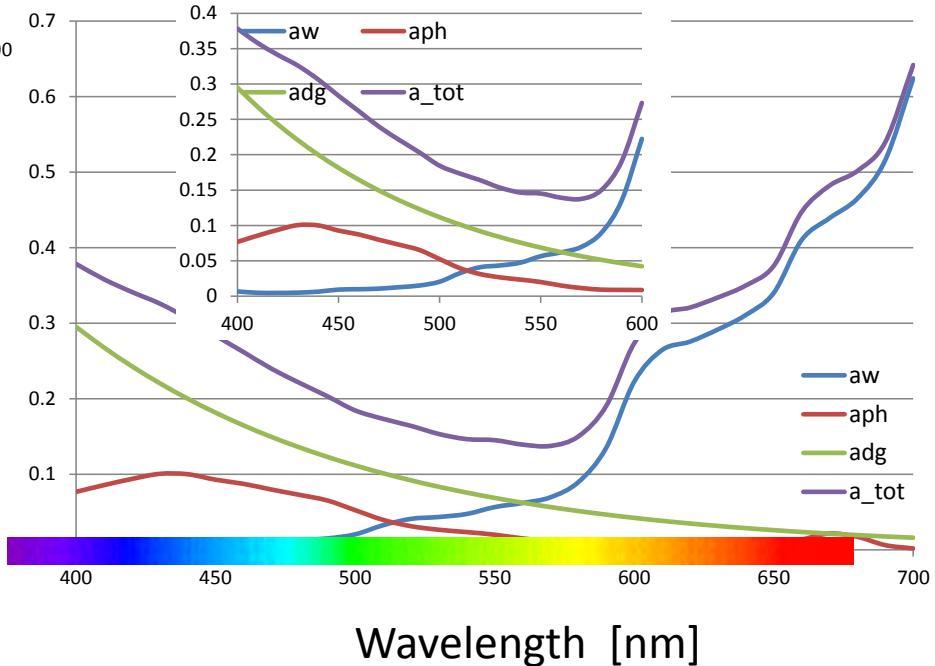
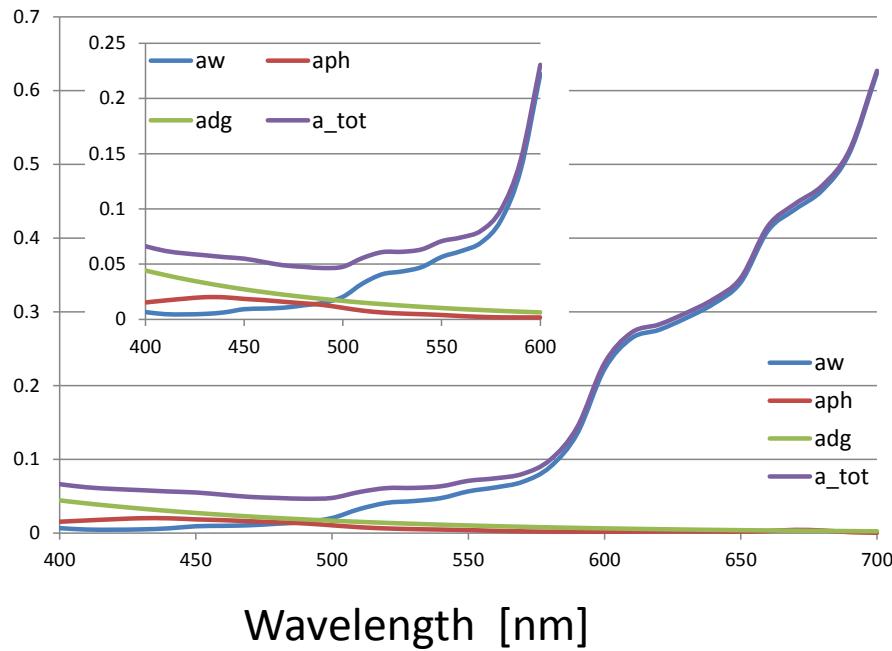
(Twardowski et al 2004)

Values of a_{ph} and a_g of natural waters

Water body	g_{440} (m ⁻¹)	p_{440} (m ⁻¹)	Reference
Adelaide L., Wisc., USA	1.85	—	
Otisco L., N.Y., USA	0.27	0.27	408
Irondequoit Bay, L. Ontario, USA	0.90	0.65	981
Bluff L., N.S., Canada	0.94	—	980
Punch Bowl, N.S., Canada	6.22	—	328
<i>South America</i>			
Guri Reservoir, Venezuela	4.84	—	558
Carrao R., Venezuela	12.44	—	558
<i>Australia</i>			
(a) <i>Southern tablelands</i>			
Cotter Dam	1.28–1.46	0.77	483, 495a
Corin Dam	1.19–1.61	0.11	483, 495a
L. Ginninderra (3-year range)	1.54±0.78 0.67–2.81	0.16–0.58	478, 479, 483, 495a
L. George (5-year range)	1.80±1.06 0.69–3.04	3.73–4.21	478, 479, 483, 495a
Burrinjuck Dam (5-year range)	2.21±1.13 0.81–3.87	0.63–1.44	478, 479, 483, 495a
L. Burley Griffin (5-year range)	2.95±1.70 0.99–7.00	2.91–2.96	478, 479, 483, 495a
Googong Dam	3.42	0.83	483
Queanbeyan R.	2.42	—	495a
Molonglo R.	0.44	—	495a
Molonglo R. below confluence with Queanbeyan R.	1.84	—	495a
Creek draining boggy ground	11.61	—	495a
(b) <i>Murray–Darling system</i>			
Murrumbidgee R., Gogeldrie Weir (10 months)	0.4–3.2	—	677
L. Wyangan	1.13	0.38	495a
Griffith Reservoir	1.34	3.73	495a
Barren Box Swamp	1.59	2.55	495a
Main canal, M.I.A.	1.11	5.35	495a
Main drain, M.I.A.	2.12	10.34	495a
Murray R., upstream of Darling confluence	0.81–0.85	—	677
Darling R., above confluence with Murray	0.7–2.5	—	677
(c) <i>Northern Territory (Magela Creek billabongs)</i>			
Mudginberri	1.11	1.13	498
Gulungul	2.28	1.68	498
Georgetown	1.99	18.00	498

(Kirk 1994)

Contrast of absorption spectra of optically active components



2. Scattering properties

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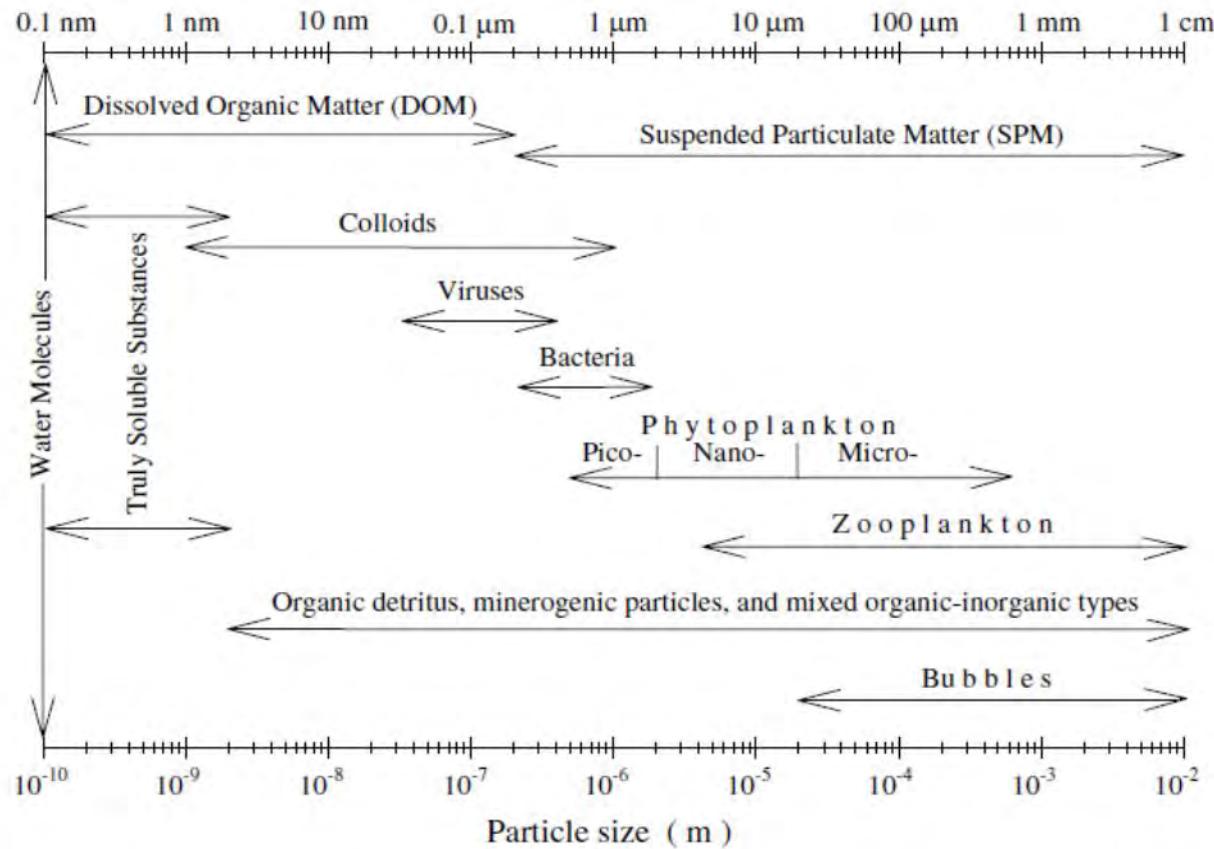
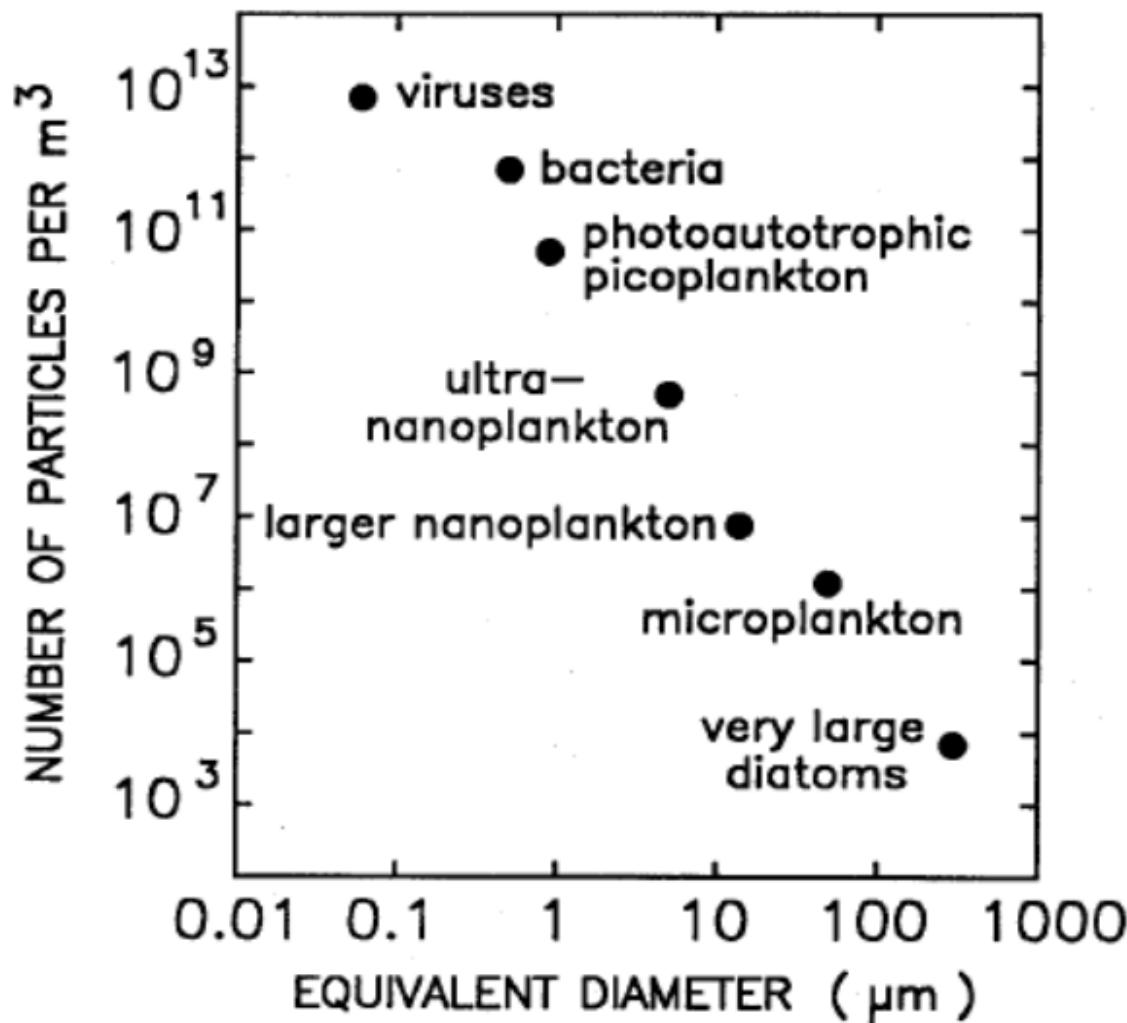


Fig. 1. Schematic diagram showing various seawater constituents in the broad size range from molecular size of the order of 10^{-10} m to large particles and bubbles of the order of 10^{-3} – 10^{-2} m in size. The arrow ends generally indicate approximate rather than sharp boundaries for different constituent categories.

Size distribution



(Stramski and Kiefer 1991)

$$b = b_w + \sum b_{xi} \quad b_b = b_w + \sum b_{bxi}$$

Very detailed:

$$\begin{aligned}
b(\lambda) &= b_w(\lambda) + \sum_{i=1}^{18} b_{\text{pla},i}(\lambda) + b_{\text{det}}(\lambda) + b_{\text{min}}(\lambda) + b_{\text{bub}}(\lambda) \\
&= b_w(\lambda) + \sum_{i=1}^{18} N_{\text{pla},i} \sigma_{b,\text{pla},i}(\lambda) + N_{\text{det}} \sigma_{b,\text{det}}(\lambda) \\
&\quad + N_{\text{min}} \sigma_{b,\text{min}}(\lambda) + N_{\text{bub}} \sigma_{b,\text{bub}}(\lambda),
\end{aligned} \tag{2}$$

(Stramski et al 2001)

Commonly used terms for scattering:

Molecules

Suspended ‘particles’

Bubbles

Turbulence

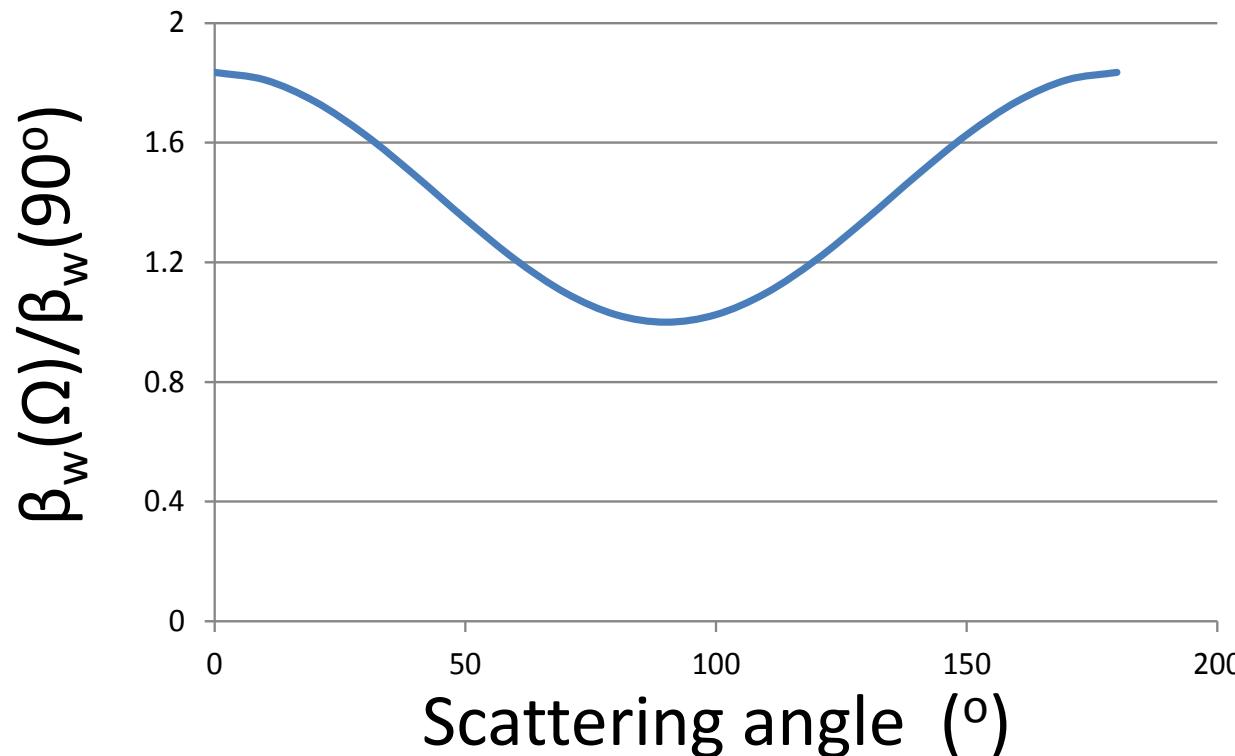
$$b = b_w + b_p$$

Or,

$$b = b_w + b_{PIM} + b_{POM}$$

Scattering of water molecules

VSF of pure water (β_w)



$$b_w = 2 b_{bw}$$

Spectral dependence

Morel 1974:

$$\beta_w = \beta_0 \left(\frac{450}{\lambda} \right)^{4.32}$$

Shifrin: 1988

$$\beta_w = \beta_0 \left(\frac{450}{\lambda} \right)^{4.17}$$

β_w is also found salinity dependent; its value could be ~30% higher for marine waters.

Value and spectrum of seawater b_{bw} :

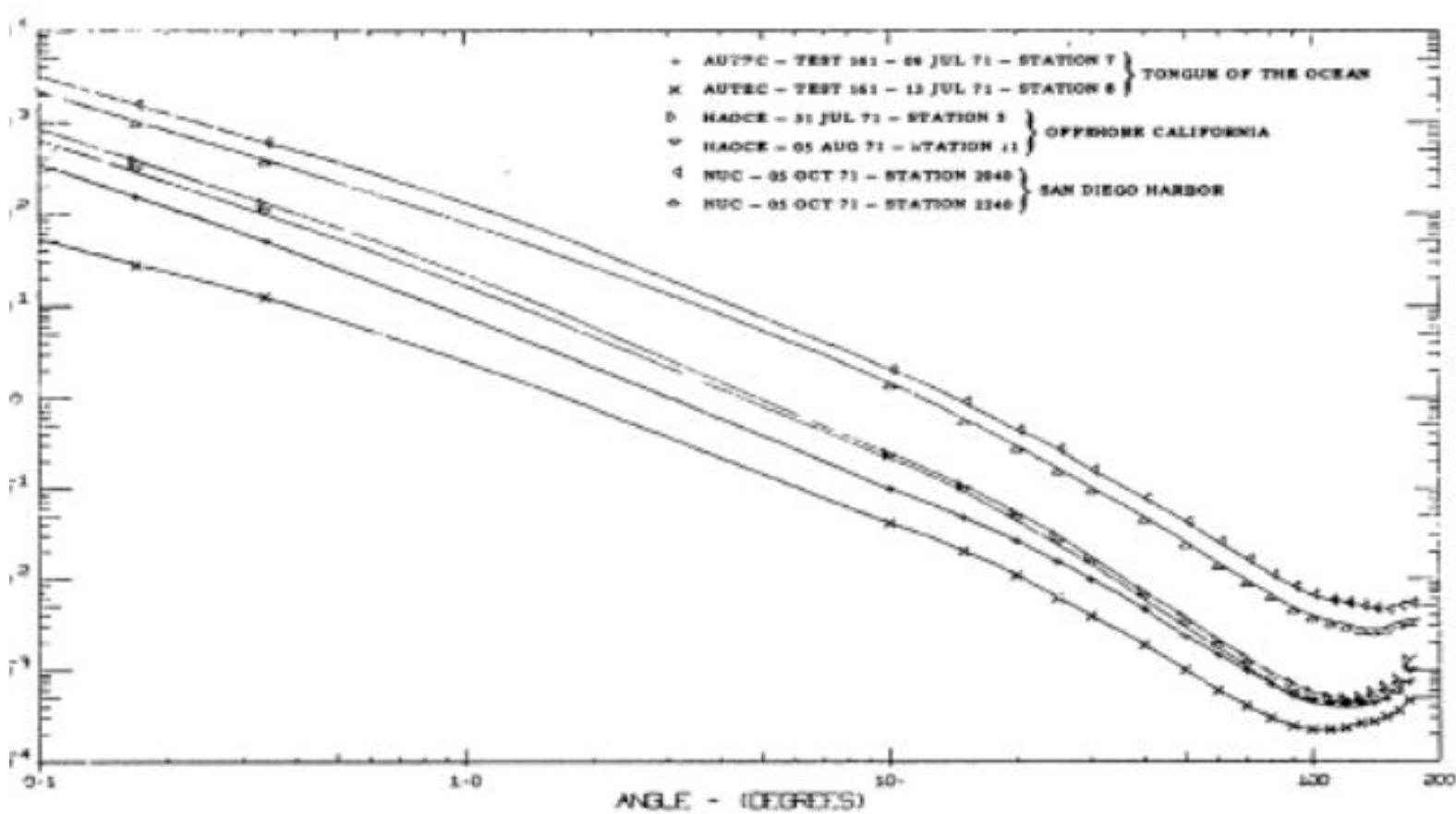
$$b_{bw}(\lambda) = 0.0023 \left(\frac{450}{\lambda} \right)^{4.32}$$

(Morel 1974)

$$b_{bw}(\lambda) = 0.0020 \left(\frac{450}{\lambda} \right)^{4.3}$$

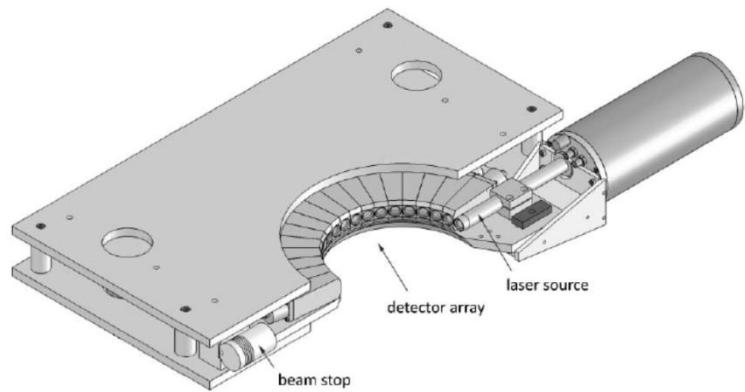
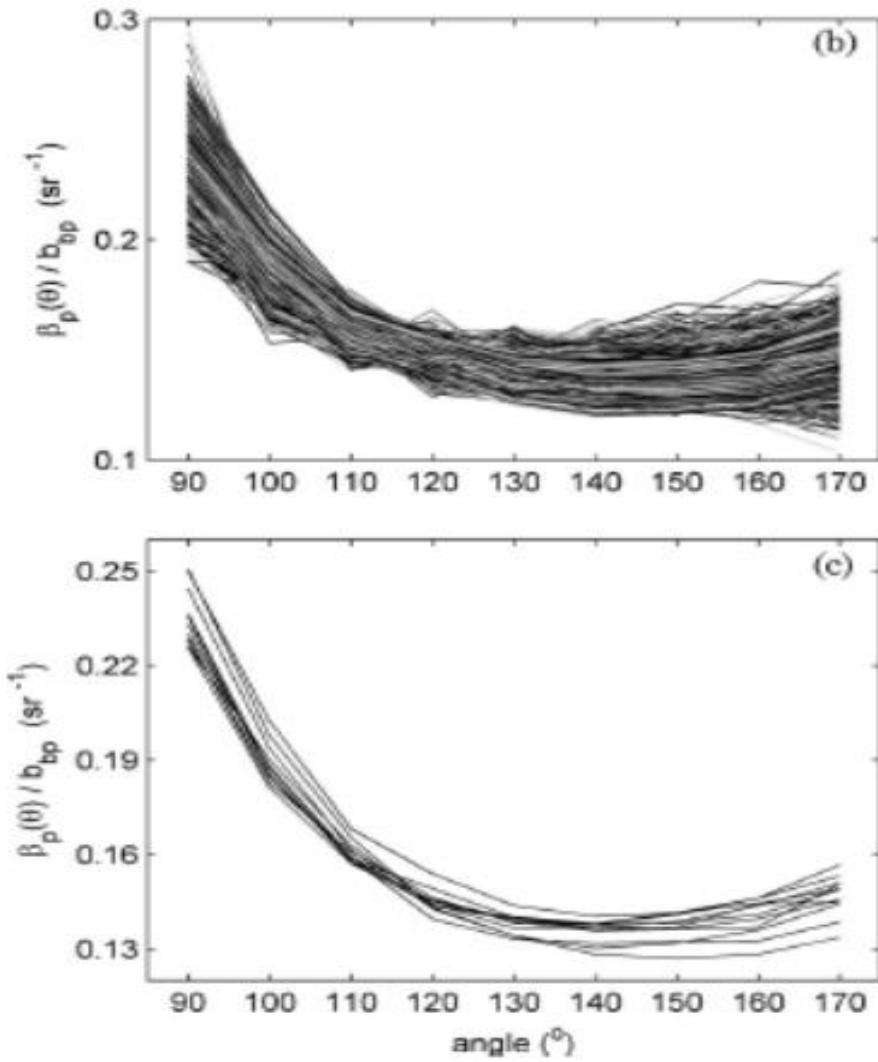
(Zhang et al 2009)

Volume Scattering Function with particles



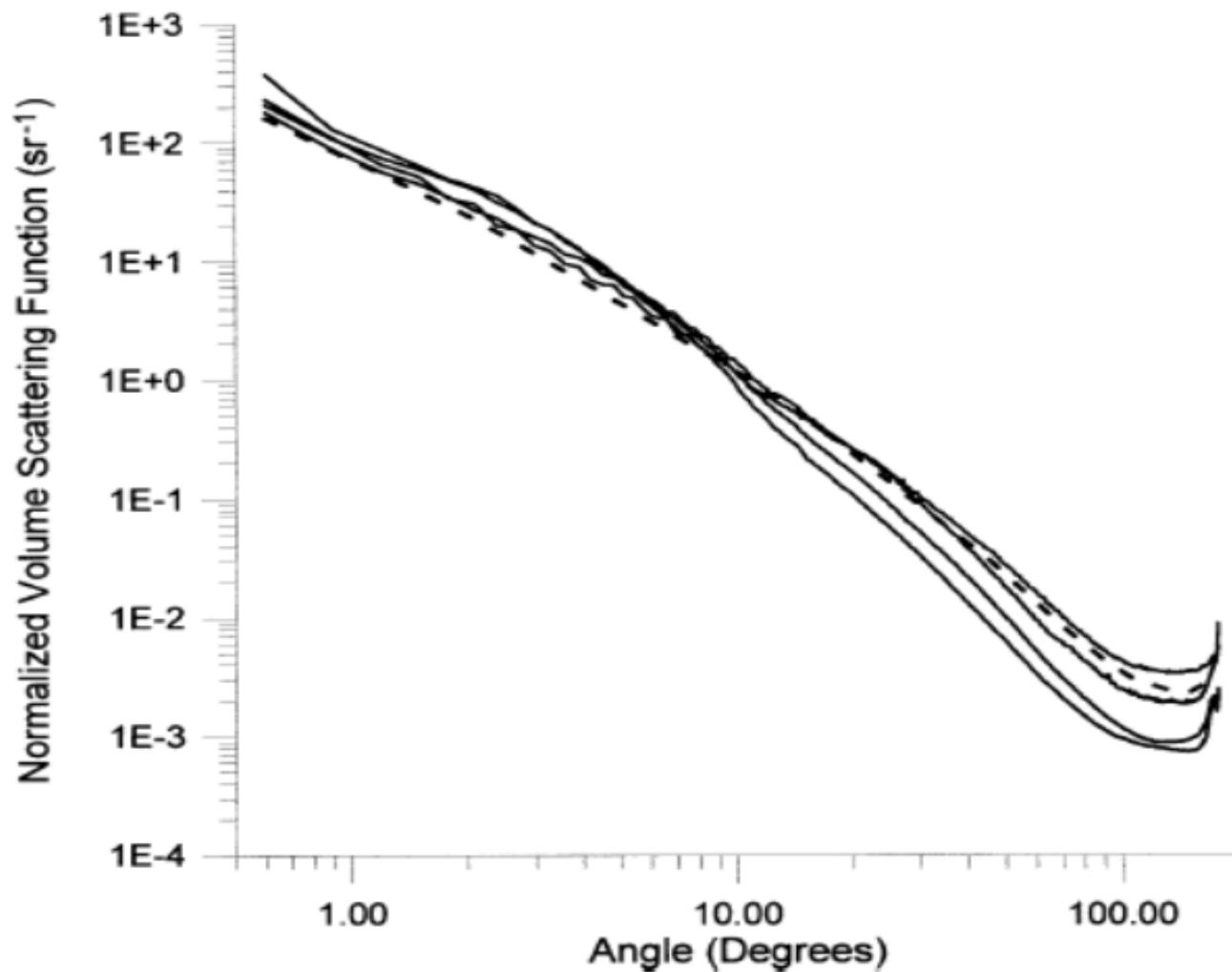
(Petzold 1972)

MASCOT measurements

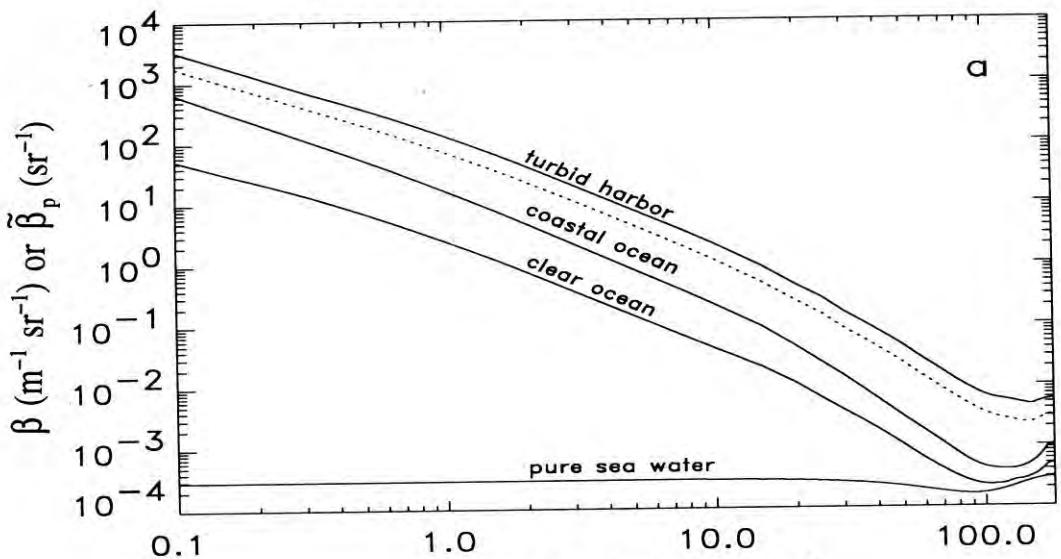


(Sullivan and Twardowski, 2009)

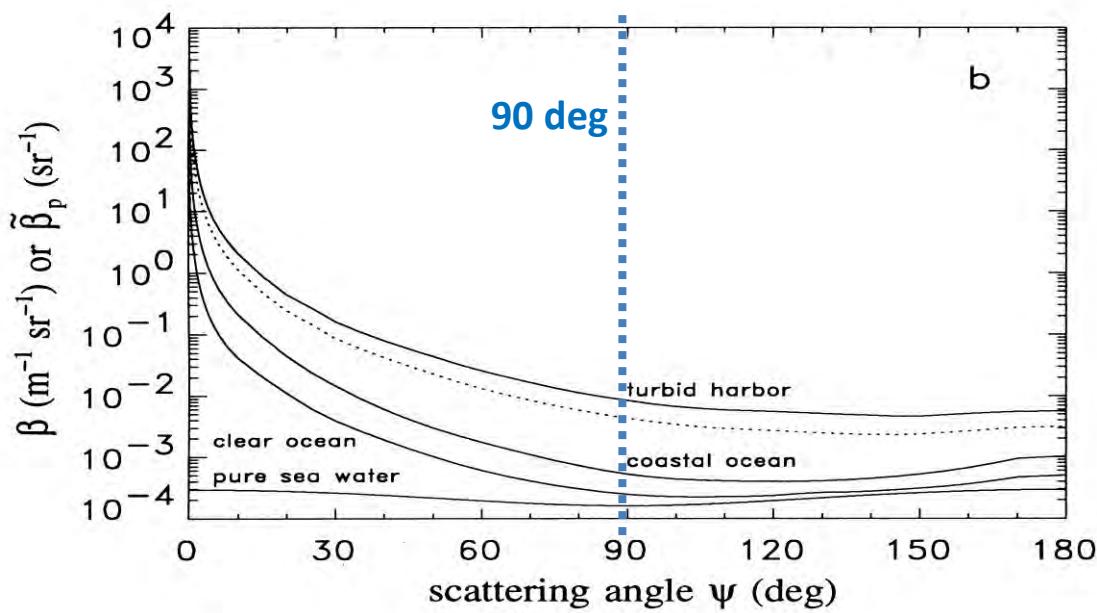
MVSM measurements



(Lee and Lewis, 2003)



(Mobley 1994)



β shape changes in a narrow range in the backward domain

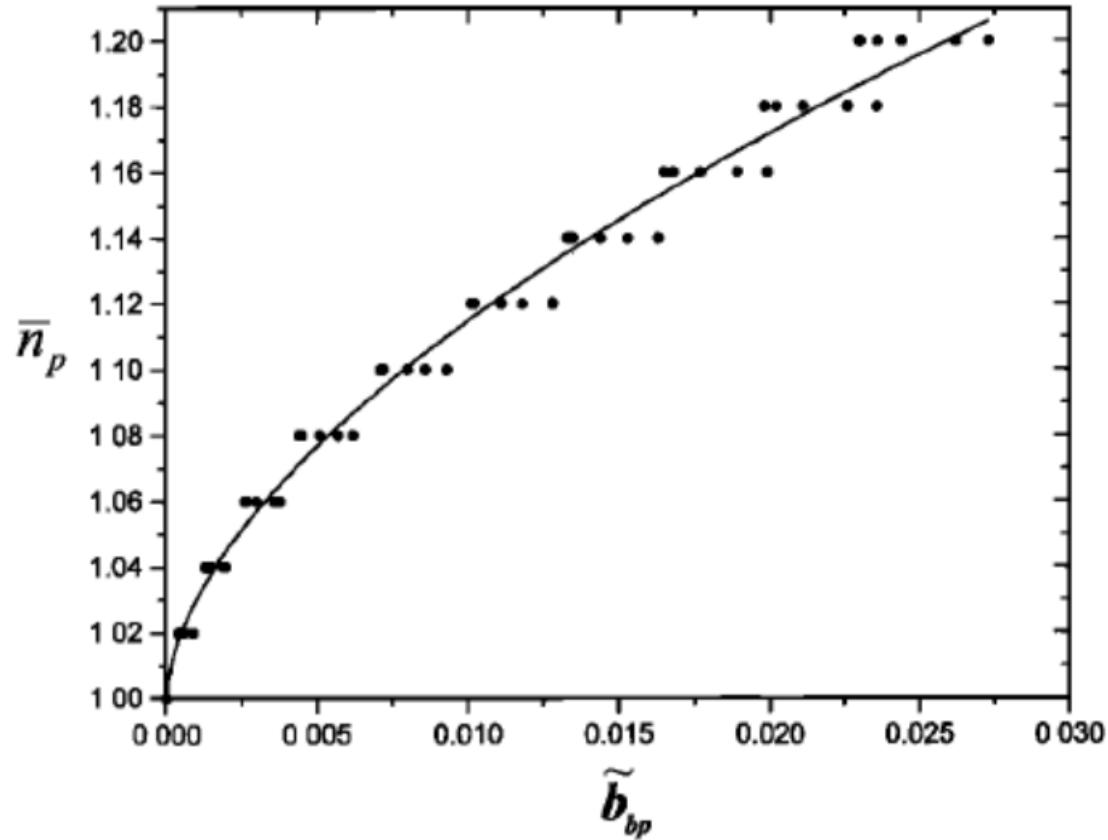
Particles are strongly forward scatters!

$$\tilde{b}_{bw} = 0.5;$$

$$\tilde{b}_{bp} \sim 0.005 - 0.05$$

Backscattering ratio: $\tilde{b}_b = \frac{b_b}{b}$

\tilde{b}_{bp} and refractive index



Twardowski et al (2001)

Examples of β model

Henyey-Greenstein (1941)

$$\beta = \frac{1}{4\pi} \frac{1 - g^2}{(1 + g^2 - 2g \cos \psi)^{1.5}}$$

Beardsley and Zaneveld (1969)

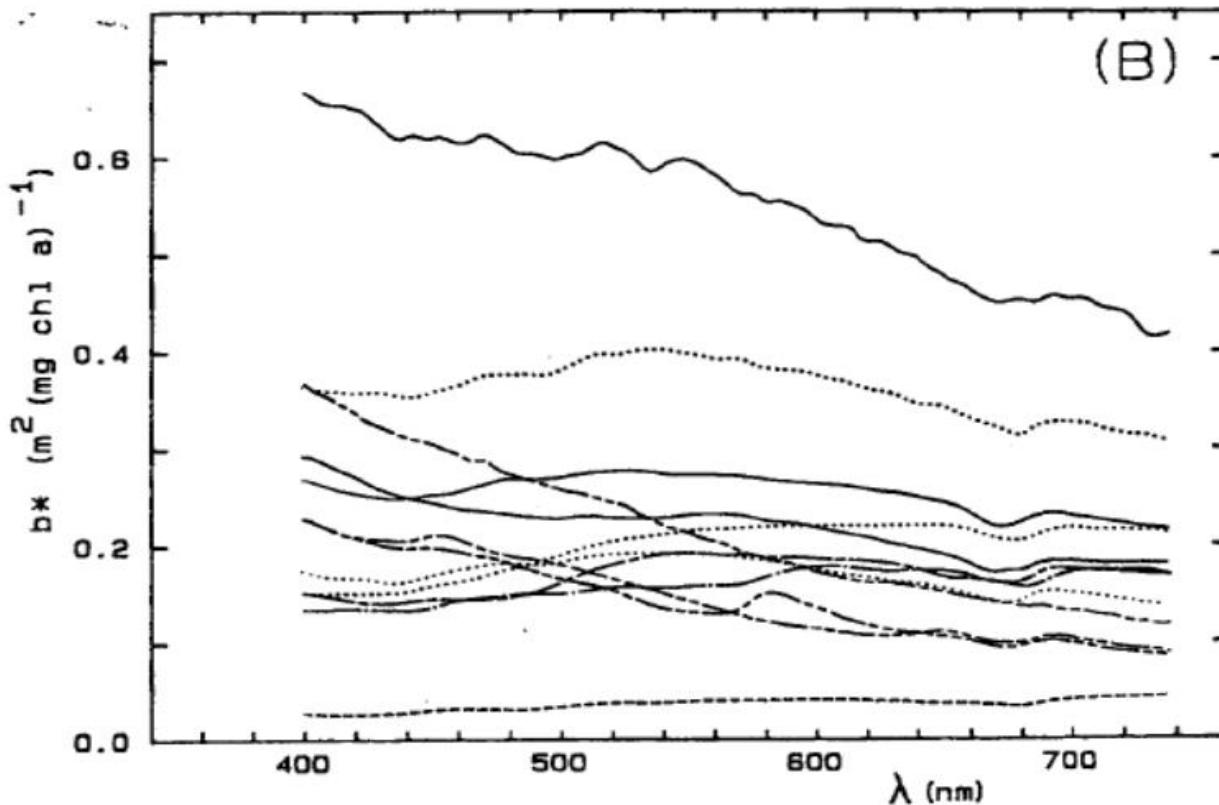
$$\beta \sim \frac{1}{(1 - \varepsilon_f \cos \psi)^4 (1 + \varepsilon_b \cos \psi)^4}$$

Fournier and Forand (1994)

$$\begin{aligned}\tilde{\beta}_{FF}(\psi) &= \frac{1}{4\pi(1-\delta)^2\delta^\nu} \left[\nu(1-\delta) - (1-\delta^\nu) + [\delta(1-\delta^\nu) - \nu(1-\delta)] \sin^{-2} \left(\frac{\psi}{2} \right) \right] \\ &\quad + \frac{1 - \delta_{180}^\nu}{16\pi(\delta_{180} - 1)\delta_{180}^\nu} (3 \cos^2 \psi - 1),\end{aligned}$$

$$\nu = \frac{3 - \mu}{2} \quad \text{and} \quad \delta = \frac{4}{3(n-1)^2} \sin^2 \left(\frac{\psi}{2} \right).$$

Spectrum of scattering coefficient



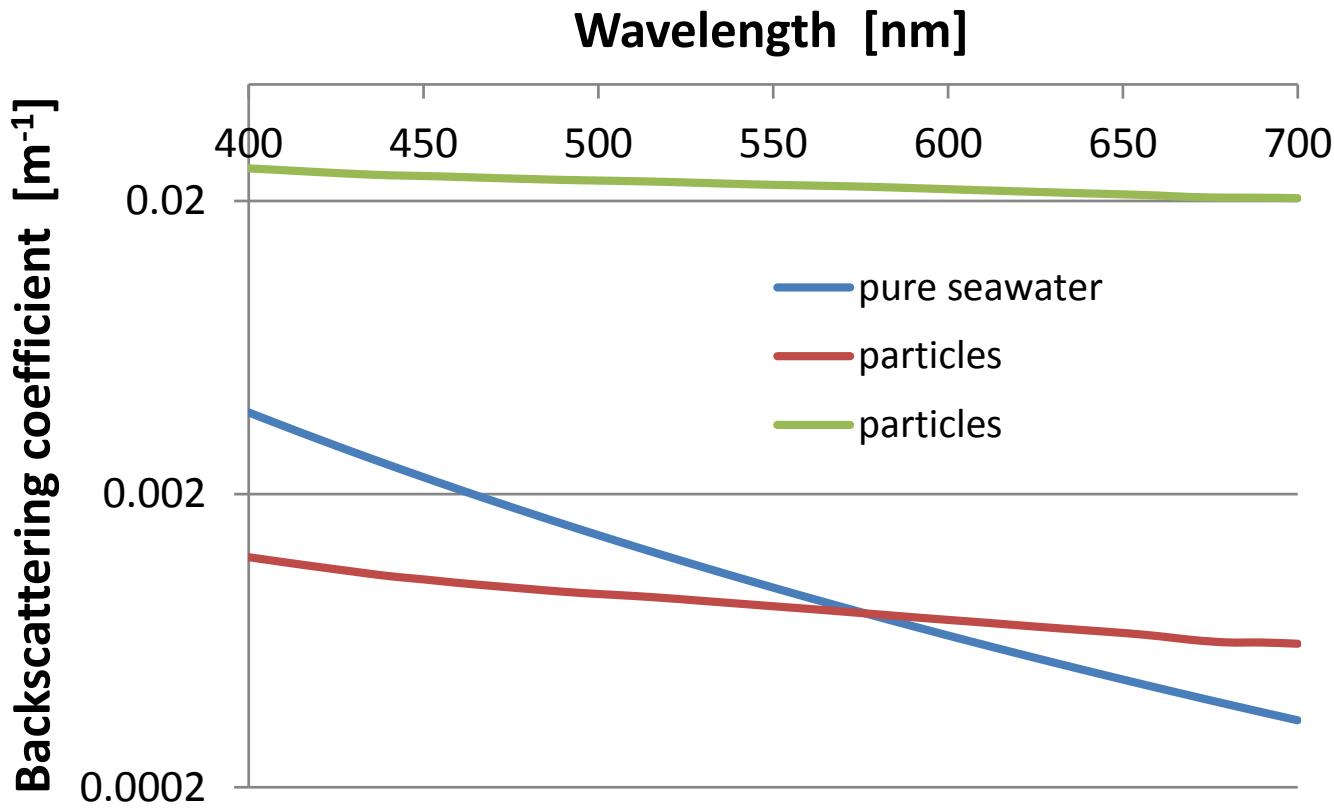
(Bricaud et al 1988)

vary weakly with wavelength

$$b(\lambda) = b(\lambda_r) \frac{-0.00113\lambda + 1.625}{-0.00113\lambda_r + 1.625}$$

(Gould et al 1999)

b_b spectrum contrast

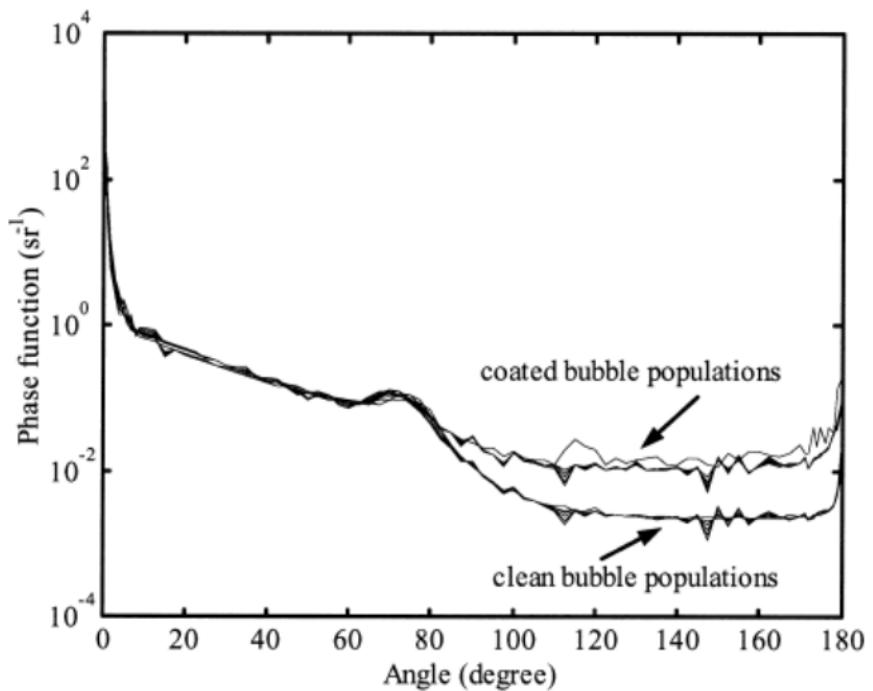
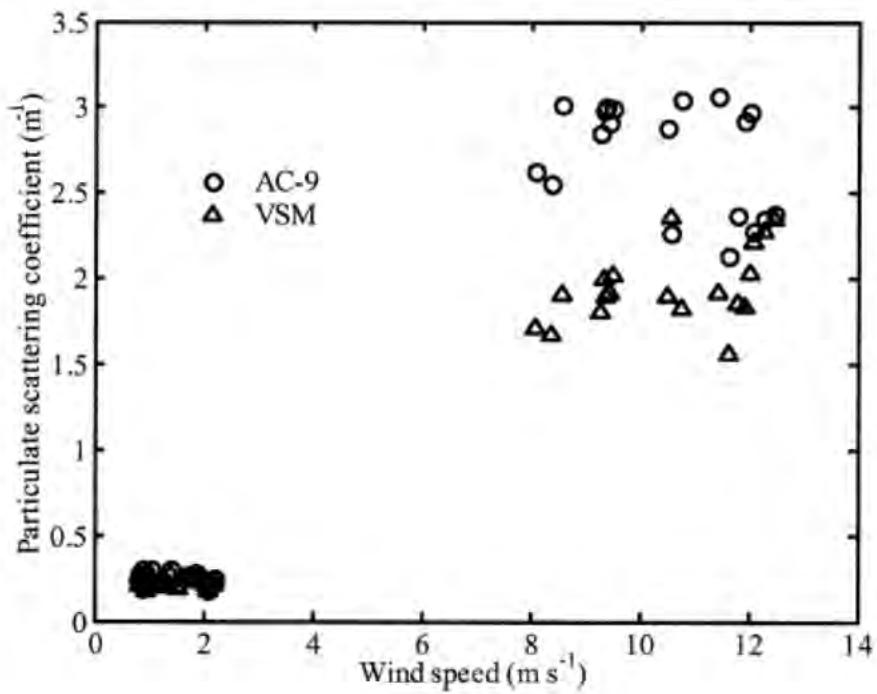


b_{bw} : ~0.0001 – 0.004 m⁻¹

$$b_{bp}(\lambda) = b_0 \left(\frac{\lambda_0}{\lambda} \right)^\eta$$

η : ~0-2.0

bubbles

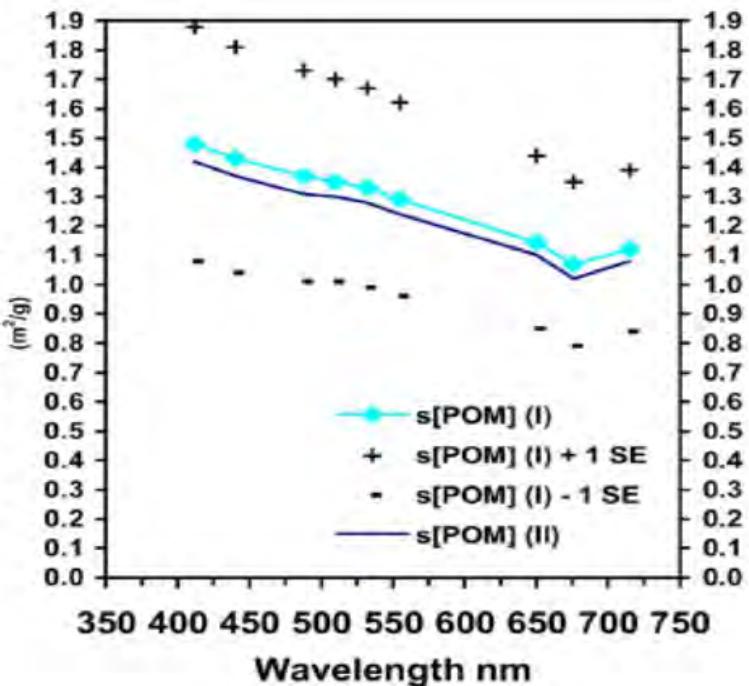


(Zhang et al 2002)

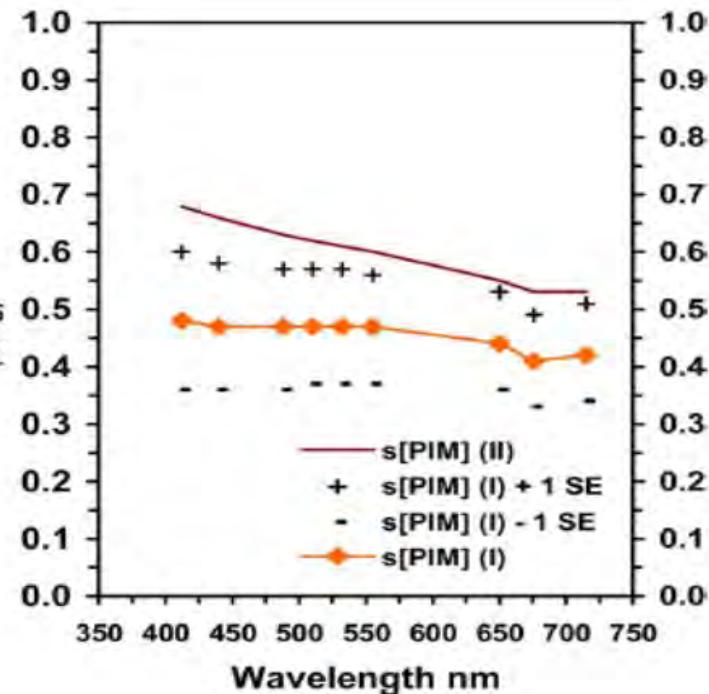
**Not known the spectral characteristics of bubble scattering,
considered spectrally flat**

Organic vs inorganic separation

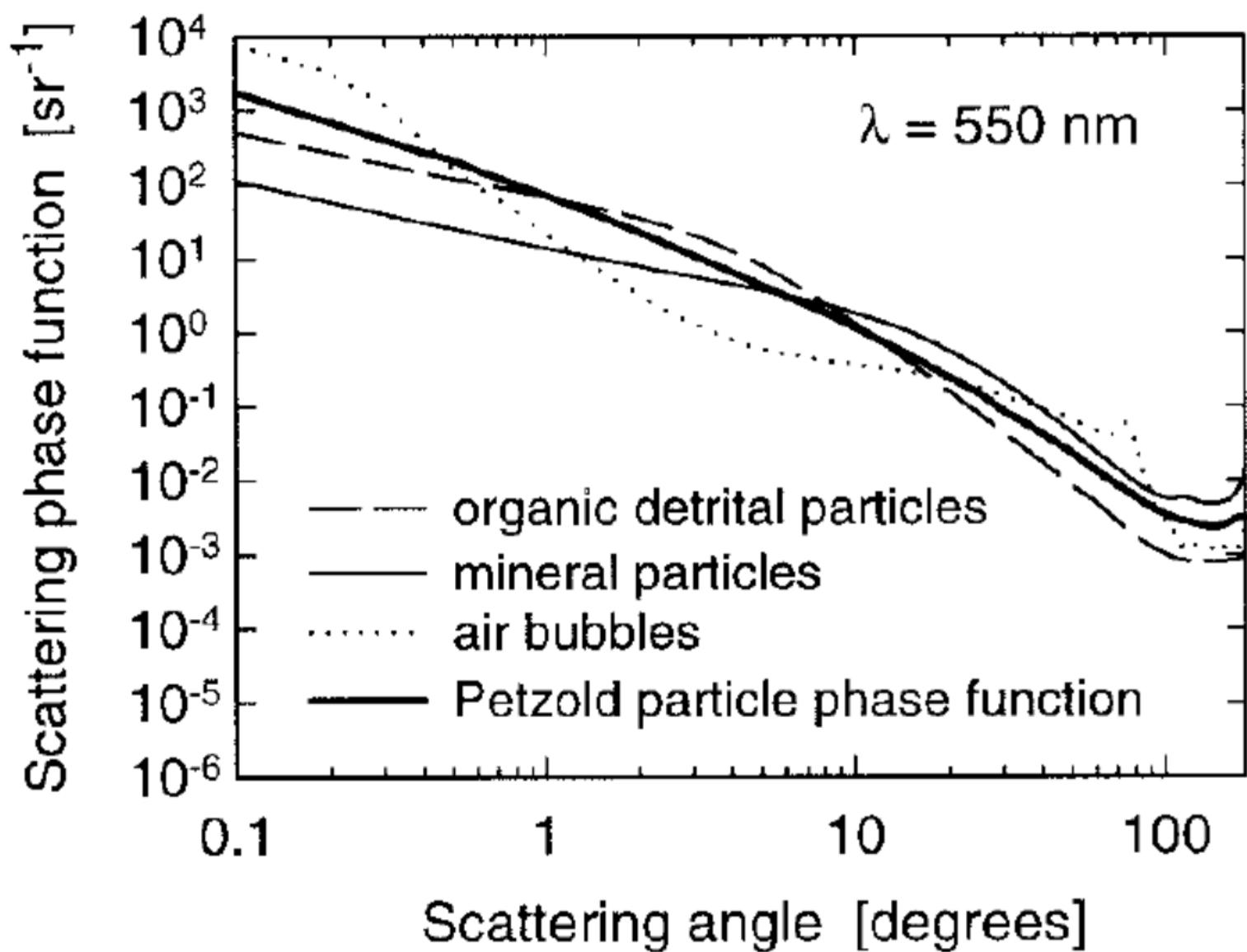
Organic Scattering Cross Sections:
Models I & II



Mineral Scattering Cross Sections:
Models I & II



(Stavn and Richter 2008)



(Stramski et al 2001)

Light scattering by microorganisms in the open ocean

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(Prog. Oceanog. 28, 343-383, 1991)



Progress in Oceanography 61 (2004) 27–56

Oceanography

www.elsevier.com/locate/pocean

Review

The role of seawater constituents in light backscattering in the ocean

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Available online 19 August 2004

Key points:

1. In addition to boundary conditions, IOPs play the key role in forming ocean/water color.
2. Primary IOPs include absorption and scattering coefficients; the latter is direction dependent.
3. Bulk IOPs are lump sum contributions of the many individual, dissolved and suspended, molecules and particles.
4. Absorption and scattering coefficients of pure (sea)water are considered constant (change with temperature/salinity), but uncertainties still exist, especially for absorption in the UV range.

5. In addition to water molecules, practically and generally, for absorption: there are three major optically active components: phytoplankton pigments, detritus and gelbstoff (CDOM); for scattering: there are organic and inorganic particulates, bubbles, and many times lumped into one term.

6. Spectrally,
water molecules are strong absorber in the longer wavelengths;
phytoplankton absorption generally has two distinct peaks with a stronger peak centered around 440 nm and weaker peak centered around 675 nm; have varying spectral shapes
detritus and gelbstoff are strong absorbers in the shorter wavelengths, and gelbstoff has steeper spectral slope;

Water molecules are strong scatter in the shorter wavelengths; ‘particle’ scattering is weakly wavelength dependent. It is strongly dependent on size, refractive index, and abundance.